



## Solvent-free Promoted One-pot Synthesis of H-quinolizine, pyrido[*a*]isoquinoline and pyrido[*a*]quinoline Derivatives

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### Abstract

This work describes a fast, mild, convenient and simple method for preparing of nitrogen heterocyclic derivatives by MCR reaction under solvent-free condition.

**Keywords:** Solvent free reaction, Multi-component reactions, Acetylenic esters, H-quinolizine, H-pyrido[*a*]isoquinoline, aH-pyrido[*a*]quinoline.

### Introduction

Quinolizines are of considerable interest due to their widespread occurrence in natural products, particularly in the field of alkaloids [1]. The importance of these nitrogen heterocyclic derivatives to the pharmaceutical industry has spurred a great amount of research, and numerous methods have been devised for their construction [2]. Although many routes to the basic ring systems are known, new general synthetic approaches are still highly desirable [3].

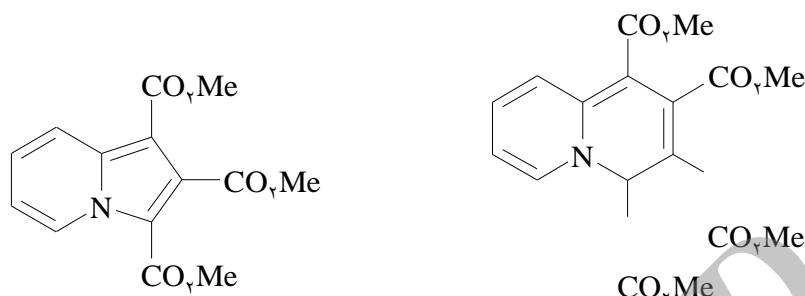
The possibility of performing chemical

reactions in the absence of solvent has been receiving more attention now-a-days [4-8]. The examples reported [9-14], demonstrate that solvent-free reactions are generally faster giving higher selectivities and excellent yields. A large variety of nitrogen heterocycles are known to form zwitterionic species on addition of activated olefins or acetylenes. Pyridine deserves special mention owing to the variety of transformations that it mediates. The earliest work in the area was reported by Diels and Alder, and their study [15] and subsequently the structure elucidation of Acheson [16-20]

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showed that pyridine reacts smoothly with *H*-quinolizine in methanol as a solvent dimethyl acetylenedicarboxylate (DMAD) (Scheme 1) [21].

to form indolizine-1,2,3-tricarboxylate and



**Scheme**

However, the above method suffers from drawbacks such as longer reaction time, the need for unfriendly solvent, and moderate yield. Recently H. Valizadeh and *et.all* have reported an addition reaction of Nitrogen-containing heterocyclic compounds with DMAD under neat condition [22]. Following, as part of our ongoing research program on the development of new protocols in heterocyclic synthesis [23-26], herein, we applied this methodology to describe the synthesis of *H*-pyrido[2,1-*a*]isoquinoline, 4*H*-quinolizine, and *aH*-pyrido[1,2-*a*]quinoline derived from the reaction between diethyl acetylenedicarboxylate, di-*tert*-butyl acetylenedicarboxylate and isoquinoline, pyridine, and quinoline under the same reaction conditions.

## Experimental

### General

Chemicals were purchased from Fluka and

used without further purification. Melting points were measured on an Electrothermal 9100 apparatus. Elemental analyses for C, H, and N were performed using a Heraeus CHN-O-Rapid analyzer, and the results agreed favorably with the calculated values. Mass spectra were recorded on a Finnigan MAT 4430 spectrometer operating at an ionization potential of 70 eV. IR spectra were measured on a Shimadzu IR-470 spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were measured on a Bruker Avance DRX-300 spectrometer using CDCl<sub>3</sub> as applied solvent and TMS as internal standard at 300 and 75 MHz, respectively.

### General procedure for the preparation of compound

In a typical reaction, a mixture of isoquinoline (0.26 g, 1 mmol) and dimethyl acetylenedicarboxylate (0.32 ml, 1 mmol) under solvent free condition was stirred for 1 hour. The progress of reaction was

monitored by TLC. The resulting precipitate was separated by filtration and recrystallized from diethyl ether (Et<sub>2</sub>O) to afford the pure compounds.

*Tetraethyl H-pyridof[2,3-b]isoquinoline-6,7,8-tricarboxylate (a)*

Yellow powder; yield: 0.85 g (90%), mp 102-

104°C. IR (KBr) ( $\nu_{\max}/\text{cm}^{-1}$ ): 1734, 1710, and

1666 (C=O), 1626-1478 (C=C). <sup>1</sup>H-

NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.14 (3H, t,  $J$  = 7.1

Hz, CH<sub>3</sub>), 1.21 (3H, t,  $J$  = 7.1 Hz, CH<sub>3</sub>), 1.30

(3H, t,  $J$  = 7.1 Hz, CH<sub>3</sub>), 1.38 (3H, t,  $J$  = 7.1

Hz, CH<sub>3</sub>), 4.08 (3H, q,  $J$  = 7.1 Hz, OCH<sub>2</sub>),

4.11 (3H, q,  $J$  = 7.1 Hz, OCH<sub>2</sub>), 4.31 (3H, q,  $J$  = 7.1 Hz, OCH<sub>2</sub>), 4.38 (3H, q,  $J$  = 7.1

Hz, OCH<sub>2</sub>), 6.77 (1H, s, CH), 7.44 (1H, d,  $J$  = 7.2

Hz, CH), 7.52 (1H, d,  $J$  = 7.2 Hz, CH), 7.66

(1H, m, CH), 7.70 (1H, m, CH), 7.78 (3H, m,

3 CH). <sup>13</sup>C-NMR (70 MHz, CDCl<sub>3</sub>):  $\delta$  = 13.7,

13.8, 14.0, and 14.2 (3 CH<sub>3</sub>), 60.8, 61.0,

61.5, and 62.7 (3 OCH<sub>2</sub>), 97.8 (CH), 111.3 (C), 119.0 (CH), 123.8 (CH), 123.9 (CH),

126.0 (CH),

127.4 (CH), 128.0 (CH), 129.8 (C), 130.0,

(C), 139.0 (C), 147.7 (C), 151.8 (C),

1608 (C=O), 1618-1470 (C=C). <sup>1</sup>H-

NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.44, 1.48,

1.57 and

1.58 (3H, s, 12 CH<sub>3</sub>), 6.74 (1H, s, CH),

7.41 (1H, d,  $J$  = 7.2 Hz, CH), 7.50 (1H, d,  $J$

= 7.2 Hz, CH), 7.04 (1H, m, CH), 7.11

(1H, m, CH), 7.30 (3H, m, 3 CH). <sup>13</sup>C-

NMR (70

MHz, CDCl<sub>3</sub>):  $\delta$  = 28.0, 28.1, 28.3 and 28.4 (3

CMe<sub>3</sub>), 79.9, 80.1, 83.8 and 84.2 (3 O-CMe<sub>3</sub>),

97.4 (CH), 112.8 (C), 120.3 (CH), 120.0 (CH),

120.2 (CH), 127.2 (CH), 128.7 (CH), 129.7

(CH), 130.1 (C), 131.0 (C), 141.2 (C), 150.8

(C), 153.4 (C), 162.7, 163.3, 164.1, and

166.3

(3 C=O). Anal. Calcd for C<sub>27</sub>H<sub>27</sub>N<sub>3</sub>O<sub>6</sub> (401.53):

C, 67.00; H, 6.76; N, 6.24; Found: C,

67.73; N, 6.00%.

*Tetraethyl H-quinolizine-6,7,8-tricarboxylate (a)*

Yellow powder; yield: 0.88 g (90%), mp 190-

196°C. IR (KBr) ( $\nu_{\max}/\text{cm}^{-1}$ ): 1741, 1710, and

1666 (C=O), 1619-1481 (C=C). <sup>1</sup>H-NMR

(300 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.18 (3H, t,  $J$  = 7.1

Hz, CH<sub>3</sub>), 1.28 (3H, t,  $J$  = 7.1 Hz, CH<sub>3</sub>), 1.36

(3H,

162.4,

t,  $J = 7.1$  Hz, CH),  $1.39$  (3H, t,  $J = 7.1$  Hz, CH<sub>3</sub>),  $4.10$  (2H, q,  $J = 7.1$  Hz, OCH<sub>2</sub>),  $4.23$   
 $1.32$  H. Djahaniani et al., J. Appl. Chem. Res., , ,  
 $1.63, 4.166, 9$ , and  $1.67, 7$  ( $\delta$  C=O). (2H, q,  $J = 7.1$  Hz, OCH<sub>2</sub>),  $4.33$  (2H, q,  $J = 7.1$   
 Anal. Calcd

for C<sub>10</sub>H<sub>12</sub>NO<sub>2</sub>N (469.49): C, 63.96; H, 5.80; Hz, OCH<sub>2</sub>),  $4.42$  (2H, q,  $J = 7.1$  Hz, OCH<sub>2</sub>),  
 $2.98$ ; Found: C, 63.90; H, 5.83; N, 2.96%.  $5.96$  (1H, s, CH),  $6.79$  (1H, dt,  $J = 6.7$  Hz,  $J$   
 $= 1.3$  Hz, CH),  $7.46$  (2H, m, 2 CH),  $8.63$

*Tetra-tert-butyl H-pyrido[2,3-b]isoquinoline* (1H, dd,  $J = 9.7$  Hz,  $\delta J = 1.3$  Hz, CH). C-  
 -*-tetracarboxylate* (b) NMR (40

Yellow powder; yield: 0.92 g (80%), mp MHz, CDCl<sub>3</sub>):  $\delta = 13.4, 13.8, 14.1$ , and  
 $14.1$

177°C. IR (KBr) ( $\nu_{\max}$ /cm<sup>-1</sup>): 1730, 1711, and ( $\delta$  CH<sub>3</sub>),  $6.0, 6.1, 6.2, 6.3$ , and  $6.4$  ( $\delta$  OCH<sub>2</sub>  
 $90.8$  (C),  $110.0$  (CH),  $123.8$  (CH),  $136.3$   
 (C),

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137,9 (C), 142,7 (CH), 146,2 (C), 149,3 (CH), 157,0 (C), 163,8, 164,8, 168,1, and 169,8 (C=O). Anal. Calcd for  $C_{11}H_{10}NO_4$  ( $\epsilon$ 19,  $\epsilon$ ): C, 70,1%; H, 5,1%; N, 2,3%; Found: C, 70,1%; H, 5,9%; N, 2,30%.

*Tetra-tert-butyl H-quinolizine-tetracarboxylate (b)*

Yellow powder; yield: 1,1 g (98%), mp 190-197°C. IR (KBr) ( $\nu_{max}/cm^{-1}$ ): 1743, 1712, and 1686 (C=O), 1620-1436 (C=C).  $^1H$ -NMR (300 MHz,  $CDCl_3$ ):  $\delta$  = 1,40, 1,46, 1,03 and 1,00 (36H,  $\epsilon$  s, 12  $CH_3$ ), 0,08 (1H, s, CH), 7,79-7,82 (1H, dt,  $J$  = 7,7 Hz,  $J$  = 1,3 Hz, CH), 7,47-7,50 (2H, m, 2 CH), 8,72-8,74 (1H, dd,  $J$  = 9,6 Hz,  $J$  = 1,4 Hz, CH).  $^{13}C$ -NMR (75 MHz,  $CDCl_3$ ):  $\delta$  = 27,7, 27,9, 28,0 and 28,3 ( $\epsilon$  CMe), 79,0, 80,0, 83,7 and 84,3 ( $\epsilon$  O-CMe), 94,7 (CH), 114,9 (CH), 122,8 (CH), 136,4 (C), 138,0 (C), 142,6 (CH), 140,0 (C), 148,4 (CH), 157,4 (C), 163,7, 160,2, 166,8, and 167,3 (C=O). Anal. Calcd for  $C_{24}H_{24}NO_8$  ( $\epsilon$ 31, 74): C, 70,0%; H, 7,7%; N, 2,3%; Found: C, 70,4%; H, 7,7%; N, 2,6%.

CH), 3,93 (q, 2H,  $J$  = 7,1 Hz, OCH), 4,09 (q, 2H,  $J$  = 7,1 Hz, OCH), 4,18 (q, 2H,  $J$  = 7,1 Hz, OCH), 4,32 (q, 2H,  $J$  = 7,1 Hz, OCH), 5,17 (1H, dd,  $J$  = 2,8 Hz,  $J$  = 2,8 Hz, CH), 6,03 (1H, dd,  $J$  = 9,4 Hz,  $J$  = 2,8 Hz, CH), 6,06 (1H, dd,  $J$  = 9,4 Hz,  $J$  = 2,8 Hz, CH), 7,14-7,17 (3H, m, CH), 7,23-7,28 (1H, m, CH).  $^{13}C$ -NMR (75 MHz,  $CDCl_3$ ):  $\delta$  = 13,2, 13,4, 13,8, and 14,0 ( $\epsilon$  CH $^3$ ), 60,7, 61,0, 61,7, and 62,1 ( $\epsilon$  OCH), 97,8 (CH), 111,1 (CH), 121,7 (CH), 120,4 (C), 127,0 (C), 127,0 (C), 128,2 (CH), 129,7 (CH), 130,8 (CH), 131,6 (C), 136,0 (CH), 138,6 (C), 151,6 (C), 162,7, 163,6, 163,8, and 167,6 ( $\epsilon$  C=O). Anal. Calcd for  $C_{20}H_{16}NO_4$  ( $\epsilon$ 79, 49): C, 73,9%; H, 5,8%; N, 2,9%; Found: C, 73,9%; H, 5,8%; N, 2,9%.

*Tetra-tert-butyl aH-pyrido[2,3-b]quinolone-*

*-tetracarboxylate (b)*

Yellow powder; yield: 1,4 g (90%), mp 108-110°C. IR (KBr) ( $\nu_{max}/cm^{-1}$ ): 1741, 1720, and 1690 (C=O), 1623-1440 (C=C),  $^1H$ -NMR (300 MHz,  $CDCl_3$ ):  $\delta$  = 1,40, 1,46, 1,03 and 1,00 (36H,  $\epsilon$  s, 12  $CH_3$ ), 0,08 (1H, s, CH), 7,79-7,82 (1H, dt,  $J$  = 7,7 Hz,  $J$  = 1,3 Hz, CH), 7,47-7,50 (2H, m, 2 CH), 8,72-8,74 (1H, dd,  $J$  = 9,6 Hz,  $J$  = 1,4 Hz, CH).  $^{13}C$ -NMR (75 MHz,  $CDCl_3$ ):  $\delta$  = 27,7, 27,9, 28,0 and 28,3 ( $\epsilon$  CMe), 79,0, 80,0, 83,7 and 84,3 ( $\epsilon$  O-CMe), 94,7 (CH), 114,9 (CH), 122,8 (CH), 136,4 (C), 138,0 (C), 142,6 (CH), 140,0 (C), 148,4 (CH), 157,4 (C), 163,7, 160,2, 166,8, and 167,3 (C=O). Anal. Calcd for  $C_{24}H_{24}NO_8$  ( $\epsilon$ 31, 74): C, 70,0%; H, 7,7%; N, 2,3%; Found: C, 70,4%; H, 7,7%; N, 2,6%.

*Tetraethyl aH-pyrido[2,3-b]quinolone-*

*-tetracarboxylate (a)*

Yellow powder; yield: 1,0 g (90%), mp 143-



۱۶۳.۷, ۱۶۴.۲, and ۱۶۷.۹ ( $\delta$  C=O). Anal. Calcd for

$C_{17}H_{13}NO_4$  (۳۰۱.۲۳): C, ۶۷.۰۰; H, ۴.۷۶; N, ۲.۰۴;

Found: C, ۶۷.۰۲; H, ۴.۷۹; N, ۲.۰۷ %.

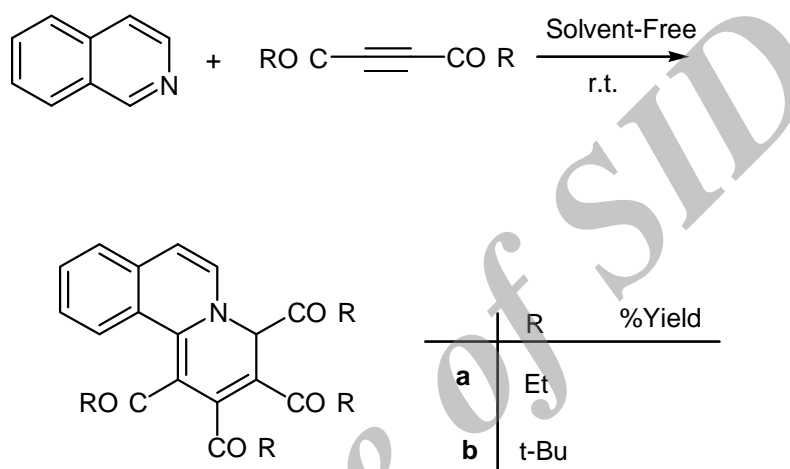
acetylenedicarboxylate  $\xi$  in the absence of

solvent at ambient temperature produces

$\xi$ H-pyrido[۲,۱-a]isoquinoline  $\rho$  in an excellent yields (Scheme ۲).

## Results and Discussion

The reaction of isoquinoline  $\psi$  and dialkyl



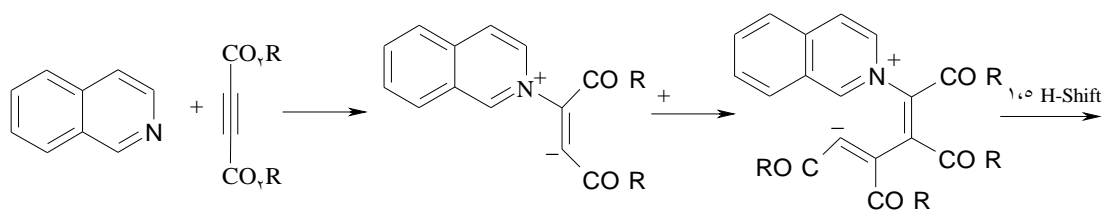
Scheme 2.

Isoquinoline undergoes a smooth reaction with dialkyl acetylenedicarboxylates  $\xi$  in the absence of solvent at ambient temperature to produce functionalized  $\xi$ H-pyrido[2,1-a]isoquinoline (**a-b**) in an excellent yields.

The reaction was completed within an hour (monitored by TLC). The  $^1H$  and  $^{13}C$  NMR spectra of the crude products clearly indicated the formation of  $\rho$ . The structures of compounds **a-b** were deduced from their elemental analyses and their IR,  $^1H$  and  $^{13}C$  NMR spectra. The mass spectra of these

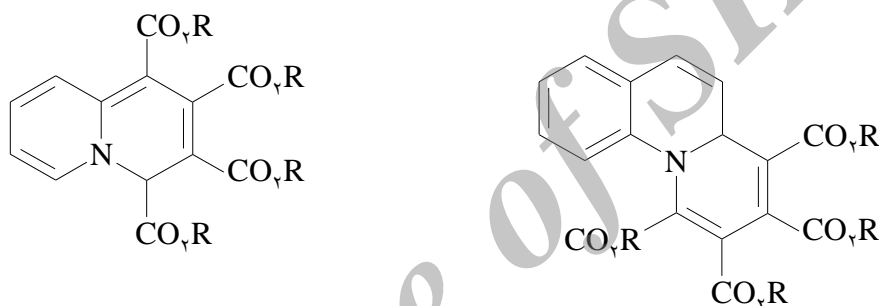
compounds displayed molecular ion peaks at the appropriate  $m/z$  values.

Although the mechanistic details of the reaction are not clearly known, a plausible rationalization may be advanced to explain the product formation. Presumably, the zwitterions [۲۷-۲۸] formed from isoquinoline and dialkyl acetylenedicarboxylate, adds to second acetylenic compound to furnish intermediate  $\psi$ . This intermediate undergoes cyclization and then  $^{\delta}H$ -shift to furnish the fused structure  $\rho$  (Scheme ۲).



Scheme 4.

Reaction of pyridin and quinoline with dialkyl  $\alpha$ H-pyrido[1,2-a]quinoline  $\beta$  respectively acetylenedicarboxylates  $\gamma$  under above (Scheme 4) conditions produce  $\alpha$ H-quinolizine  $\delta$  and



Scheme 5.

### Conclusion

The presented reaction provides a simple entry to the one-pot synthesis of  $\alpha$ H-pyrido[2,1-b]quinoline,  $\alpha$ H-quinolizine, and  $\alpha$ H-pyrido[1,2-a]quinoline derivatives of potential synthetic interest. The present procedure carries the advantage that, not only is the reaction performed under neutral conditions, but also the substances can be mixed without any activation or modification.

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