Preliminary Studies on Antioxidative Potentials of Extracts of Defatted Locust Bean Condiment

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ABSTRACT: Preliminary investigation on antioxidative potentials of defatted fermented locust bean extract in comparison to defatted non-fermented locust bean was conducted. Qualitative antioxidative assessment on the extract was accomplished by screening the phytochemical endowment, and similar bioactive components. Also some antioxidative indices were quantitatively evaluated. Results showed that the defatted fermented samples were characterized with high amount of phytochemical and bioactive compounds notably phenolic compounds, saponins, peptides, amino acid and reductones. However, alkaloids and flavonoids were not detected. UV-Spectral characteristics of the extracts corroborated same. Similarly, quantitative antioxidative markers evaluated showed that the antioxidative capacity in terms of relative reducing power, total phenolic content and DPPH of fermented locust bean of defatted fermented locust bean extract were higher than the antioxidative capacity of the corresponding defatted non-fermented locust bean extract. Also, the antioxidant activity rate content, and EC₅₀ of the extract were evaluated. This study demonstrated the antioxidant endowment of fermented locust bean condiment in comparison to non-fermented locust bean condiment.

Keywords: Antioxidative Potentials, Defatted Extract, Locust Bean Condiment, Rate Constants.

Introduction

Nutritional enhancement is one of the values added to a food by a desirable food fermentation process. Fermented foods can be more nutritious than unfermented pair as a result of catabolic or breakdown of complex components or synthesis complex vitamins such as Vit B₁₂ and other growth factor. Also lock nutrients in indigestible cells could be released. Another means is by the enzymatic splitting of carbohydrate simpler structural into such as sugar (Potter and substances Hotkiss, 1996). More importantly, facilitates fermentation process detoxification of inherent food inhibitors. This is by conversion of inedible, raw beans to digestible edible commodities due to degradation non-digestible of oligosaccharides such as stachyose and

Condiments are one of the principal food fermentation products and are desirable for culinary application with sole purpose of enhancing the taste and aroma of foods. Fermented locust bean is the most important condiment of natural origin in West Africa and central savannah region (Odunfa, 1986; Daramola *et al.*, 2009). Previous studies (Omafuvbe *et al.*, 2004) showed that fermentation enhanced the nutritional status of locust beans condiment. This claimed was expressed by increase in nutrients factors such as reducing sugars and free amino acids

raffinose which result to decreased flatulence potentials. addition. In fermentation of some in- edible foods lend reduction, or elimination of phytic and oxalic acids, proteins are hydrolysed to peptides, and amino acids, that could function directly in taste or serve as precursors for aroma active molecules (Beaumont, 2002)

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and peptides. Increase in the enzymic activities due to the fermentation processes were also reported. The presence of amino acids, sulphur containing amino acid, enzymes and tocopherols are reported in locust bean condiment by fermentation that suggest the condiment could possess antioxidant activity. This is partly the reasons for this study. In addition, recently Daramola and Osanyinlusi (2013) reported the possibility of enhancing the shelf-life of locust bean condiment by dehydration and removal of fat.

In furtherance of our studies, on locust bean condiment, this paper to my knowledge for the first time reports on the antioxidative potentials of defatted locust bean condiment extracts in comparison to the extracts of the defatted non fermented locust bean.

Materials and Methods

- Materials

Seeds of locust bean (*Parkia of biglobosa*) used in this study were obtained from commercial centre at Ado-Ekiti, Nigeria. The seeds were decorticated and subsequently divided into four portions prior to the fermentation process.

- Preparation of locust bean condiment

One of the decorated locust beans was processed into condiment following the methods elicited by Omafuvbe *et al.* (2004) and the samples obtained were tagged unfermented locust bean, while the second part (half) was fermented and the sample obtained was tagged fermented locust bean. Both samples were milled afterward.

- Solvent extraction of locust:

Solvent extraction of both the fermented and non-fermented locust bean was accomplished in two phases. Phase one was to defat the samples and phase two was to extract the bioactive components. This was accomplished in accordance with the method

described by Adegoke and Gopalakrishna (1998). Briefly, an amount of 10g of finely ground locust bean (fermented and nonfermented) was extracted with 100ml of nhexane in soxhlet extractor for 2h. The residue obtained from the two samples (fermented and non-fermented) were then subjected to a repeat of extraction protocol but the solvent was replaced with ethanol and diethylether-chloroform mixture. The crude extracts were evaporated to remove the solvent and extract recovered.

- Analytical methods

- Phytochemicals evaluation of the defatted fermented and non-fermented locust beans extracts

Phytochemical investigations on extracts were performed as described by Trease and Evans (2002) and Harborne (1984) as follows:

- Test for reducing sugars

To 0.5ml of the extracts solution, 2ml of a mixture (1:1) of Fehling's solution 1 (A) and Fehling's solution II (B) were added and the mixture were boiled in a water bath for five minutes. A brick-red precipitate indicated the presence of free reducing sugars.

- Test for flavonoids

To 0.5ml of the extracts solution, a few drops of 10% ferric chloride solution were added. A green or blue colour indicated the presence of phenolic nucleus.

- Test for free amino acids

0.5ml extracts solution were treated with few drops of ninhydrin reagent, heated in water bath, a purple colour indicated the presence of amino acids.

- Test for alkaloids

0.5ml of extract solution was added with 0.2ml of 36.5% hydrochloric acid and 0.2ml Dragendroff's reagent. Production of orange precipitate denoted the presence of alkaloids.

- Test for glycosides

0.5ml extract solution was added to 2ml of 50% hydrochloric acid. The mixtures were hydrolized for 2hrs on a water bath. After that 1ml pyridine, few drops of 1% sodium nitroprusside solution, and 5% sodium hydroxide solution were added. Pink to red colour designated the presence of glycosides.

- Test for reducing compounds

To 1ml of ethanolic extracts of samples, a few drops of 10% ferric chloride and potassium hexaferrate (III) solution were added. A green or blue colour indicated reducing activity.

- Test for saponins

2ml of double distilled water was added with 1ml of each extract solution, few drops of olive oil were added and agitated. Formation of soluble emulsion indicated the presence of saponin.

- UV – visible absorption spectra assessment

UV – visible absorption spectra of defatted fermented and non-fermented locust bean extract at 0.006% in ethanol were recorded on an uv-visible double recording spectrophotometer (Unicam Helios & uv/Vis/Spectrometer V.2.05 Serial No UVA 072519 year 2000 complant).

- Quantitative assessment of antioxidative potentials
- Evaluation of total phenolic content.

Total phenolic content was evaluated according to the method described by Taga et al. (1984). A 100µl of Folin-Ciocalteau reagent (2N wrt acid Fluka Chemic AG-Ch-9470 BUCHS) was added to each sample (20µl) and well mixed after the addition of 1.58ml of water. After 30 seconds, 300µl of 2% sodium carbonate solution was added and the sample tubes were left at room temperature for 2h. The absorbance (A) of

the developed blue colour was measured at 750nm using Unicam Helios & UV/VIS/Spectrophotmeter. A plot of A750nm against conresponding concentration was used to calculate phenolic content (g/g ascorbic acid equivalent).

- Determination of relative reducing power Reducing power of each sample was determined in accordance with the method of Oyaizu (1986). Simply, each sample (1mg/ml) in ethanol (2.5ml) was mixed with sodium phosphate buffer (pH 6.6). the buffered sample was mixed conditioning reagents (1%K₃-Fe-CN₆, 10% TCA, 0.1% FeCl₃), centrifuged, diluted using distilled water and absorbance was measured at 700nm. Higher absorbance indicates a higher reducing power. A plot of A_{750nm} against corresponding concentration was used to calculate the phenolic content

- Total flavonoids contents determination

(g/g ascorbic acid equivalent).

The total flavonoids contents were determined by colorimetric method and expressed as mg ascorbic acid equivalent per g of dry weight, using the method described by Chang et al. (2002) with little modification. 0.5ml of each extracts (prepared from 1mg of crude extract dissolved in 1ml of methanol) were mixed with 1.5ml of methanol, 0.1ml of 10% aluminium chloride (AlCl₃) solution, 0.1ml of (IM) sodium hydroxide solution (NaOH) and 2.8ml of ddH₂O. The resulting mixtures were well mixed and incubated for 30 minutes at obscurity. The absorbance of the reaction mixture was measured at 430nm with a UV/visible spectrophotometer. A plot of against corresponding A_{430nm} concentration was used to calculate the phenolic content (g/g ascorbic acid activity equivalent).

- Measurement of radical-scavenging activity

Radical scavenging activity of the samples on DPPH was estimated according to the method of Yamaguchi *et al.* (1998). An aliquot of samples (200uL, 0.31-2.5mg.mL), ascorbic acid (0.04-1.25mg/mL) was mixed with the 100mM Tris-HCl buffer (800uL, pH 7.4) and then added to 1mL of 500um DPPH in ethanol (final concentration of 250um). The mixture was shaken vigorously and left to stand for 20min at room temperature in the dark. The absorbance of the resulting solution was measured spectrophotometrically at 517nm. The capability to scavenge the DPPH radical was calculated using the following equation.

Scavenging effect (%) = [1- (absorbance of sample at 517nm/absorbance of control at 517nm)] x100

- Determination of free amino acids

Free amino acids in the partially defatted locust bean condiment and non-defatted samples were extracted with 80 % ethanol (v/v) in accordance with the method of Odibo *et al.* (1990). The free amino acids in the ethanolic extract were estimated using the ninhydrin colorimetric method (Rosen, 1957) using glycine as standard.

Results and Discussion

- Phytochemical characteristics of defatted fermented and unfermented locust bean extract

results of the The phytochemical assessment of the extracts of the defatted locust bean fermented condiment in comparison to unfermented are shown in Table 1. The samples showed reducing activity. However, the degree of reducing property was much exhibited by the extract of the defatted fermented sample in comparison to the extract of defatted unfermented sample. Consequently, the fermentation process enhanced the reducing activity of locust beans. It is important to note that the reducing power of a substance

defines the primary antioxidant activity (Giese, 1996).

The screen test also showed that the reducing sugars were present in all the samples with higher intensity mark in fermented extract sample. Increase in the presence of reducing sugar is most probably due to the breaking down of both structural and non-structural polysaccharide in agreement with elicitation of Potter and Hotkiss (1996).

More importantly, the presence of free amino acid was screened. The intensity of the presence of free amino acids in the fermented sample was much higher in comparison to the unfermented sample. This expected because, fermentation processing of locust bean for formation of condiment is based on hydrolysis of protein into peptides, amino acids, nucleotides, that are responsible for the desired sensory property of the condiment. The assessment also showed little presence of saponins, with no test positive for the presence of flavonoids and alkaloids. This does not suggest that there are no secondary of metabolites in locust beans but fermentation contributes to detoxification of inherent phenolic anti-nutritious factors present in locust beans.it is speculated that some of the phenolic components could be alkaloids or saponins. As expected glycosides are present in all the samples

- UV- Spectral characteristics of defatted fermented and unfermented locust bean extracts

The uv-spectral characteristics of the unfermented and fermented extracts are shown in Table 2. UV- reading ranges from 180nm to 420nm (Silverstein *et al.*, 1981). A comparison of the spectral characteristics showed that there was poor primary absorption of the samples of the unfermented in comparison to the uv-absorption of the fermented extract.

Table 1. Phytochemical Screen of fermented and unfermented defatted locust bean extracts

Samp	Reducing	Reducing	Free amino	Saponins	Flavonoids	Alkaloids	Glycosides
le I.D	compounds	sugars	acids				
1.	+++	-	++	+	-	-	++
2.	+++++	++	+	+	-	-	++
3.	++	+++	+++++	+	-	-	+++
4.	+++++	++++	+++++	+	-	-	+++

⁺ strength of presence, Means Not detected; Sample ID = Sample Identification Number; 1 = Defatted unfermented locust bean cake ethanol extract; 2 = Defatted unfermented locust bean cake diethylether-chloroform extract; 3 = Defatted fermented locust bean cake ethanol extract

Table 2. UV – spectral characteristics of fermented and unfermented defatted locust bean extracts

Sample	A	2	Α.	2		2
I.D No	Α	λ	A	λ	A	λ
1	0.48	396	0.21	400	-	-
2.	0.54	396	0.28	400	-	-
3.	8.65	396	8.83	400	-	-
4.	8.80	396	1.37	450	0.27	500

A = absorbance, λ = wavelength, Sample Identification Number = sample ID No,

Table 3. Some Antioxidative and Biochemical y parameters

Sample	Total Phenolic Content	Free Amino Acid	DPPH %	Relative Reducing Power	
ID	(TPC)	(FAA)		(RPP)	
1	8.80	0.3	57	13.30	
2	17.00	0.215	53	10.00	
3	10.00	3.3	89	29.50	
4	8.48	1.66	82	23.40	

^yTPC = mg ascorbic acid activity equivalent per mg sample; FAA = mg glycine activity equivalent per mg sample; RPP = mg ascorbic acid reducing activity equivalent per mg sample

Absorption (primary and secondary) in the ultraviolet region by samples are diagnostic feature of unsaturation or unbound election in the absorbing molecules (Shriner *et al.*, 1979). Free election is a prerequisite for antioxidative activity (Giese, 1996). However, there was no secondary absorption at the uv region both for the fermented and unfermented samples. This result signalled the absence of complex

^{4 =} Defatted fermented locust bean cake diethylether-chloroform extract

^{1 =} Defatted. Unfermented locust bean cake ethanol extract, 2 = Defatted unfermented locust bean cake ethanol-diethylether extract, 3 = Defatted fermented locust bean cake ethanol extract 4 = Defatted fermented locust bean cake ethanol-diethylether extract

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phenolics and other aliphatic hydroxyl compounds. This result corroborates the detection of no or low principal phenolic compounds such as flavonoids, alkaloids, and anthraquinone.

- Selected biochemical and antioxidative indices

Four biochemical and antioxidative indices were assessed with view to gain insight into the quantitative antioxidative endowment of defatted unfermented and fermented extracts of locust bean. The result is presented in Table 3. The Total Phenolic ascorbic Contents (mg acid equivalent/mg sample as determined in this study) of the unfermented extract were high, ranging from 8.80 to 17.00 in comparison to a low values (8.48-10.00) obtained for the fermented samples. This result suggests that the enzymatic activity during fermentation has resulted in hydrolysis of complex phenolic compounds like flavonoids. alkaloids. This was attested in the result of the phytochemical screen test (Table 1).

Free amino acids: The free amino acids, FAA (mg glycine equivalent per mg sample) of the defatted unfermented extract of locust bean ranged from 0.215 to 0.300. This was extremely low as compared to the high FAA of the extract of the defatted fermented locust bean with values that ranged from 1.66 - 3.30. This result is similar to the earlier work by Odunfa (1986), as reported by Beaumont (2002). The protein is hydrolysed into amino acids, peptides during fermentation and is the foremost characteristics of bio-transformation of constituents of locust beans (Odunfa, 1986; Beaumont, 2002; Omafuvbe et al., 2004; Daramola and Osanyinlusi, 2013).

Free radical scavenging activity of the extracts: The result of the free radical scavenging activity (FRSA) of fermented

and unfermented locust bean extracts evaluated using DPPH is shown in Table 3. The FRSA (%) of defatted extract of unfermented locust bean is low (53%) in comparison to the high (82-89%) FRSA of the defatted extract of fermented locust beans. There has not been a report on FRSA of locust bean condiment or its extract and therefore one is unable to compare the findings.

Relative reducing power: The results of the relative reducing power of the defatted unfermented locust bean extract is low (10.00 - 13.00) when compared to high (23.40-29.50) relative reducing power of the defatted fermented locust bean extract. Considering the result of the total phenolic content, free amino acids in relation to DPPH and relative reducing power it is important to infer that DPPH and RRP is not largely due to phenolic compounds but due to the activity of reducing substances usually free amino acids, peptides and most probably oxidized lipid-amino acid reaction products that are known for antioxidative activity (Hidalgo et al., 1998) and other hydrolysate products in the fermented extract of the locust beans.

Antioxidant activity constant, EC₅₀, and ARP: Table (4) shows the summary of antioxidant activity constant, EC₅₀ and ARP of the unfermented and fermented defatted extracts of locust bean. It can be seen in the table that the antioxidant activity rate constant of the extract of the fermented locust bean is 2 to 8 times more potent than the antioxidant activity rate constant of the extract of the unfermented locust bean. Similarly, the result of the EC₅₀ showed that lesser amount of the extract of fermented locust bean is required to accomplish EC₅₀ activity in comparison with extract of the unfermented locust bean. The obtained ARP is also showed in the Table (4).

Table 4. Antioxidative activities of the defatted fermented and non-fermented locust bean extracts as expressed by antioxidant activity constant (K), half inhibition concentration (EC_{50}) and antiradical power (ARP)

Antioxidative Reaction	Sample ID	K (mL/mg)	EC ₅₀ (mg/mL	ARP
DPPH Radical	1	1.3807	0.5020	1.992
	2	1.452	0.4774	2.095
	3	3.667	0.18902	5.291
	4	2.4366	0.2845	3.515
[@] RRP	1	2.6775	0.2588	NA
	2	2.500	0.2773	NA
	3	6.120	0.1133	NA
	4	18.12	0.03825	NA

[@] relative to be absorbance, NA = Not applicable; Note: ARP = $(EC_{50})^{-1}$

Conclusion

Antioxidative potential of defatted fermented locust bean extract is superior to the antioxidative potential of unfermented defatted locust bean extract as shown by both qualitative assessment using phytochemical screen test and quantitative assessment using TPC, RRP and DPPH.

References

Adegoke, G. O. & Gopalakrishna, A. G. (1998). Extraction and identification of antioxidants from the spice *Aframomum danielli*. Journal of American Oil and Chemical Society, 75, 1047-1052.

Beaumont, M. (2002). Flavouring composition prepared by fermentation with *Bacillus spp*. International Journal of Microbiology, 75, 189-196.

Chang, C., Yang, M., Wen, H. & Chen, J. (2002). Estimation of total flavonoid content in propolis by two complementary colorimetric methods. J. Food Drug Analysis, 10, 178-182.

Daramola, B., Fasominu, O.A., Oje, O. J. & Makanju, O. O. (2009). Influence of dietary supplementation on biotransformation of locust beans *Parkia biglobosa* to condiment. African Journal of Biotechnology, 8, 6, 1116-1120.

Daramola, B. & Osanyinlusi, S. A. (2013). Effect of processing on some biochemical characteristics and storability of locust bean condiment. Journal of cereals and oilseeds, 4, 3, 36-41.

Giese, J. (1996). Antioxidants: Tools for preventing lipid oxidation. Food Technol., 50, 1, 73-81.

Harborne, J. B. (1984). Phytochemical Methods. 2nd ed Chapman and Hall, London, 85-196.

Hidalgo, F. J., Ahmad, I., Alaiz, M. & Zamora, R. (1998). Effect of oxidized lipid/amino acid reaction products on the antioxidative activity of common antioxidants. J Agric Food Chem 46, 3768 – 3771.

Odibo, F. C., Nwabunnia, F. & Osuigwe, D. I. (1990). Biochemical changes during fermentation of *Telfaria* seeds for ogiri production. World Journal Microbiology Biotechnol., 6, 425-427.

Odunfa, S. A. (1986). Dawadawa In: Reddy NR, Pierson MD, Salunkhe DK (Eds). Legume-Based fermented Foods (RC Press Boca Raton pp 173-189.

Omafuvbe, B. O., Falade, O. S. Osuntogun, B. A. & Adewusi, S. R. A. (2004). Chemical and biochemical changes in Africa locust bean (*Parkia biglobosa*) and melon (*Citrullus vulgaris*) seeds during fermentation to condiments. Pak. J. Nutr., 3, 3, 140-145.

Oyaizu, M. (1986). Studies on extracts of browning reaction: Antioxidative activities of extracts of browning reactions prepared from glucosamine. Japan Journal of Nutrition, 44, 307-315.

Potter, N. N. & Hotchkiss, J. H. (1996). Food Science CBS publishers and Distributor 1st Indian edition, Daryaganj, NewDelhi, 313-314.

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Rosen, H. (1957). A modified ninhydrin Colorimetric Analysis for amino acids. Arch. Biochem. Biophys, 67, 10-15.

Shriner, R. L., Fuson, R. C., Curtin, D. Y. & Morill, T. C. (1979). The Systematic Identification of Organic Compounds: A Laboratory Manual. 6th edn. John Wiley, New York, pp 416-430.

Silverstein, R. M., Bassler, G. C. & Morill, T. C. (1981). Spectrometric Indentification of Organic Compounds. 4th edn. John Wiley and Sons, New York, USA, pp. 166-170.

Taga, M. S., Miller, E. E. & Pratt, D. E. (1984). Chia seeds as a source of natural lipid antioxidants. Journal of American Oil and Chemical Society, 61, 928-932.

Trease, G. E. & Evans, W. C. (2002). Pharmacognosy. 15th edn. Harcourt Publishers, Edinburg, UK.

Yamaguchi, T., Takamura, H., Matoba, T. & Terao, J. (1998). HPLC method for evaluation of the free radical scavenging activity of foods by using 1, 1 dipheny l-2-picrylhydrazyl. Biosci Biotech Biochem., 62, 1201-1204.

