### Journal of Physical and Theoretical Chemistry

of Islamic Azad University of Iran, 7 (1) 1-6: Spring 2010 (J. Phys. Theor. Chem. IAU Iran)
ISSN: 1735-2126

# Thermodynamics Study of Polychlorinated Biphenyls (PCBs) Passing through SWNT and Their Removal from Environment

# L Mahdavian<sup>1,\*</sup>

<sup>1</sup> Department of Chemistry, Doroud Branch, Islamic Azad University, P.O. Box: 133, Doroud, Iran Received February 2010; Accepted April 2010

### **ABSTRACT**

Polychlorinated Biphenyls (PCBs) have been widely used to flame-retard products common in homes and the workplace, and subsequently. Polychlorinated biphenyls (PCBs) typify a class of highly toxic, stable aromatic pollutants. PCBs are recently being scrutinized for potential environment damage in groundwater and in the atmosphere. The elimination of chemical pollutants from a contaminated environment is one of the most important steps towards achieving the goal of environmental remediation.

Recent studies have used carbon nanotubes (CNTs) for adsorption of pollution in environment, which are significantly higher dioxin elimination efficiencies. In this study, mechanism and interaction of single-walled carbon naotube (SWNT) for removal of PCBs are calculated by Gaussian program package. Inter SWNT; there are four situations for PCBs near SWNT (6, 6) with length of 1.54 nm that we calculated simulation of passing PCB through SWNT. The thermodynamics properties are calculated for passing PCBs across from SWNT that their results are showed this method can use for removal PCBs in environmental and their interaction are low potential in SWNT middle. So there is a place for adsorption of PCBs into SWNT, PCBs trap in it. This passing is exothermic, spontaneous, and favorable.

**Keywords**: Polychlorinated biphenyls (PCBs); SWNT (single-walled carbon nanotube); Environment pollution; MNDO method; DFT

# INTRODUCTION

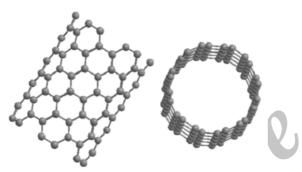
PCBs were used for many different industrial purposes, including electrical industry and hydraulic fluids, casting wax, carbonless carbon paper, compressors, heat transfer systems, plasticizers, pigments, adhesives, liquid cooled electric motors, fluorescent light ballasts and etc. Bioremediation might be an effective, cost competitive and environment-friendly solution for remediation environment matrices contaminated by PCBs but it is still unsatisfactory, mostly for the limited biodegradation potential of bacteria involved in the processes [1]. Polychlorinated biphenyls (PCBs) are highly toxic priority pollutants widespread and they accumulate in the food chain and end up in human bodies when people consume animal and fish products

[2-5]. Also, PCBs with large numbers of chlorines stable and thus resistant to more biodegradation. The most highly favored PCBs tended to be the ones with large numbers from chlorine. These congeners are also proving to be the ones that present the greatest environment and health risks [6]. Activated carbons are widely used as the traditional adsorbents in Japan and Europe, for the elimination of dioxins from the gaseous emissions of waste incinerators [7]. The CNTsbased adsorbent column designed which was obtained by simply packing a stainless-steel tube with CNTs [8]. However, its potential for adsorptive applications seems to be restricted to the purification of polluted gas streams.

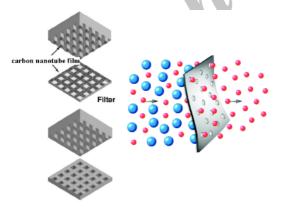
<sup>\*</sup>Corresponding author: Mahdavian\_leila@yahoo.com

Single-walled carbon nanotubes (SWNTs) have attracted great interest due to their unique electronic properties and nanometer size. Because of these unique properties, they have great potential candidates in many important applications such as nano-scale electronic devices, chemical sensors and field emitters [9-11].

Theoretical studies have confirmed the remarkable change in electronic properties of SWNT due to the detection and removed of gas molecules [12, 13]. SWNTs (single-walled carbon nanotubes) can be chiral or non chiral. The tubes are uniquely characterized by a pair of integers (n, n) that are components of the vector defining the direction in which the graphite sheet was rolled up in Fig. 1.



**Fig. 1**. Optimized configuration top-view and Sideview SWNT (6, 6).

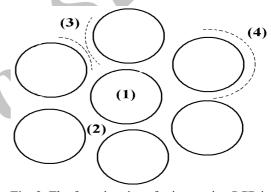


**Fig. 2**. Homogeneous two-phase separation by SWNT membrane.

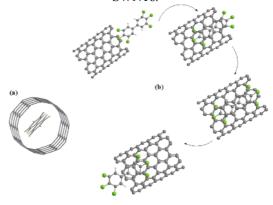
In previous reported vapor sensors based on SWNT Field Effect Transistors (FETs), the structure of the FET sensor and the experimental geometric response are schematic [14]. Nanotubes have high capillarity properties and can pass or adsorb gases and liquids. The carbon nanotube membranes can be made by covering a

silicone paper or metal catalyst particles; they are vertical row, which can be used as a compact package (Fig. 2). Then, the spaces between SWNT are filled with ceramic materials that durability of membrane increases.

In this work, interactions between gas molecular species and SWNTs are simulated and investigated. Exposing to gas molecules such as PCBs, the electrical resistance of a semiconducting SWNT is found to dramatically increase or decrease. The PCB is neared to SWNT and is passing through it or is been adsorption on surface reduced to other produces, by this method is decreasing PCB's environmental damage. There are four situations for PCB beside SWNTs that are showed in Fig.3.



**Fig. 3.** The four situations for interaction PCB in SWNTs.



**Fig. 4.** Configuration (a): Top-view and (b): Sideview passing PCB across from SWNT in environmental.

The PCB molecules may pass through SWNT, adsorb on SWNT or convert to other product. For investigation and calculation the passing of PCB through SWNT, is neglected from other interaction in environmental and the passing of PCB through SWNT is simulated in

Fig.4. The nanotube sensors exhibit a fast response and a substantially higher sensitivity than that of existing solid-state sensors in 298K. Sensor reversibility is achieved by slow recovery under environmental conditions or by heating to high temperatures.

Interaction between PCBs molecules and SWNT is investigated by MNDO method in semi-empirical methods. We study the structural and total energy, thermodynamics properties of passing PCBs and SWNT in 298K. All the geometry optimization structures were carried out using Gaussian program package. Density Functional Theory (DFT) optimized intermediates and their transient states. The results show a sensitivity enhancement in resistance and capacitance when PCBs is passing through SWNT.

### **THEORETICAL**

# Synthesis of PCB isomers and their properties

PCBs consist of a biphenyl (two benzene rings with a carbon to carbon bond between carbon 1 on one ring and carbon 1' on the second ring) with a varying number of chlorines. Symmetrical isomers are synthesized in Fig.5. Chloraniline was treated with sodium nitrite to give the corresponding diazonium derivative, which was replaced by iodine, followed by The condensation [15]. 3,4,5,3`,4`,5`hexachlorobiphenyl isomers have the melting points about 201°C, 300 retention times among and spectroscopic observations on K bands and  $\epsilon$ values is 266 nm and 23.20M [16].

**Fig. 5.** Synthesis PCB, which it is the most toxic isomer for PCB structure.

The toxicity of a PCB is dependent not only upon the number of chlorines present on the biphenyl structures, but also the positions of the chlorines. For instance congeners with chlorines in both para positions (4 and 4') and at least 2 chlorines at the meta positions (3, 5, 3', 5') are considered to be "dioxin like" and are particularly toxic [17]. When there is just 1 or no substitution in the ortho position, the atoms of the congener are able to line up in a single plane (sometimes referred to as coplanar). The planar or flat configuration is particularly toxic. In this study, PCB is been simulation in Ball-and-stick models, then geometry optimizations were performed using a B3LYP/6-31G\*\* by DFT methods.

### **COMPUTATIONAL METHODS**

The geometry optimizations were performed using an all-electron linear combination of atomic orbital and the first-principles methodology. We used density functional theory (DFT) [18] with MNDO and IR-DFT methods by the Gaussian program package. In this work, the B3LYP/6-31G level of theory was used for the optimizations of solids  $(3, 4, 5, 3^{\circ}, 4^{\circ}, 5^{\circ})$ -hexachlorobiphenyl and  $C_{120}$  (6, 6) [19].

The accuracy of semi-empirical quantum mechanics method depends on the database used to parameterize the method. Configuration interaction (or electron correlation) improves energy calculations using CNDO, INDO, MINDO/3, MNDO, AM1, PM3, ZINDO/1, and ZINDO/S for these electron configurations. We can use the information obtained from semiempirical calculations to investigate many thermodynamics and kinetic aspects of chemical processes. Energies and geometries of molecules have clear relationships to chemical phenomena. The heat of formation is calculated for these methods by subtracting atomic heats of formation from the binding energy. MNDO has been used widely to calculate heats of formation, geometries, molecular dipole ionization energies, electron affinities, and other properties [20, 21]. The interaction parameters between the PCB molecules and SWNT structures were taken from the study of nanosurface that the interactions between them were refined for this work in DFT calculations. Balland-stick models of the SWNT and PCB are showed in Fig.4. The electronic structure and the conductance properties are calculated for PCB passing through SWNT by DFT.

### RESULTS

Polychlorinated biphenyls (PCBs) are persistent pollutants that are ubiquitous in the food chain, and detectable amounts are in the blood of almost every person in most populations that have been examined [22]. The suggestion is that water with PCBs can then passes through the precisely drilled sapphire nozzle and separated

into droplets of equal size on exiting the nozzle. These droplets passed through the electrostatic field between the nozzle and the ring electrode and acquired an electrostatic charge on their surfaces. Electrostatic repulsion forces dispersed the droplets as they fell into the SWNT hardening solution. In this manuscript, we simulated this interaction by computer programs of chemistry, so the thermodynamics properties of PCBs passing through SWNT are calculated for them (table 1), which PCBs passed in-side to out-side in length of tube by six stages.

Table 1. Thermodynamic properties of interaction PCB-SWNT in 298K

Distance (nm)	G <sub>ele</sub> (MJ/mol)	E <sub>ele</sub> (V)	Dipole moment (D)	$\frac{E_{bin}}{(MJ/mol)}$	H <sub>f</sub> (MJ/mol)	RMS (kcal/molÅ)
0	-27886.35	-68.82	11580	7904.74	8001.09	4574
0.246	-29117.10	-71.85	10470	8146.11	8242.45	5011
0.493	-38783.19	-95.71	1.024	-82.45	13.90	83.30
0.739	-31706.70	-78.24	8570	8156.80	8253.15	4664
0.985	-32093.54	-79.20	8141	8486.03	8582.38	4946
1.584	-31353.04	-77.37	7230	7496.59	7592.93	4214
2.135	-27979.52	-69.05	11520	7929.76	8026.11	4691

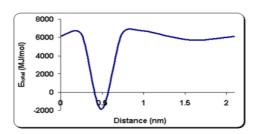
With the objective to understand how the pattern and degree of chlorination influence on the properties of the title molecules, a computational study on chlorinated biphenyls has been undertaken. The study includes conformational searches (and further refinement by DFT) and the ab initio calculation of MNDO methods in semi empirical and the dipole moments for all the steps into nanotube. The most significant property is the MNDO, finding a good correlation between the MNDO and the substitution pattern on chlorinated biphenyls. The most toxic congeners possess highly positive values of electrostatic potential on the aromatic rings and highly negative values of electrostatic potential on the chlorine atoms.

The PCB is passing through SWNT by six distances from in-side to out-side of length nanotube, which it is investigated by MNDO and its results are showed in Table 1. A current versus voltage curve recorded with a SWNT sample after time exposure to PCB showed upfold conductance depletion. Exposure to PCB molecule increased the conductance of the SWNT sample, such as that in 0.493 nm

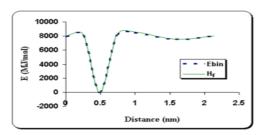
distance, the  $E_{ele}$  is -95.71V, so it is increased until out-side nanotube. RMS gradient (kcal/mol·Å) is different for passing of PCB in length of SWNT in 298K.

The  $G_{ele}$  (MJ/mol) for this passing has two minimum amounts in 0.493 and 0.985 nm of SWNT length. The dipole moment (D) in both ends of SWNT is at most amounts about 11550D and in 0.493 nm distance is the least quantity. These places of inter nanotube are snared PCB and can eliminate it in environment.

The total energy for this passing is showed in Fig. 6. The  $E_{total}$  is decreased in middle of length of nanotube, which this is a potential for entered PCB to SWNT. For interaction between them, the heat of formation (enthalpy) is calculated in MNDO methods by subtracting atomic heats of formation from the binding energy. The heat of formation and binding energy are fitted together for this interaction (Fig. 7). The least among of them is in 0.493nm length of SWNT (6,6) for the  $E_{bin}$  and  $H_f$  is ~-82.45 and 13.90 MJ/mol respectively, which is more than total energy.



**Fig. 6**. The total energy (MJ/mol) of passing PCB across from SWNT.



**Fig. 7**. The heat of formation and binding energy (MJ/mol) of passing PCB across from SWNT.

In the interaction, we correlate the sensor signals with the relative changes of the electrical resistance ( $\Omega$ ) so we have to convert calculation data (in Table 1) to the electrical resistance ( $\Omega$ ) that showed in Fig. 8. The SWNT is an up hole-doped semiconductor, as can be gleaned from the current versus gate voltage curve shown in Fig. 8 (middle curve), where the resistance of the SWNT-PCB is observed to decrease.

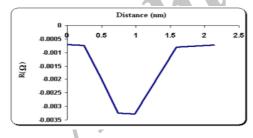
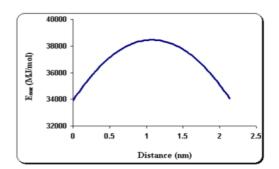


Fig. 8. The electronic resistance ( $\Omega$ ) calculated for passing PCB across from SWNT.



**Fig. 9**. The nuclear energy (MJ/mol) calculated for passing PCB across from SWNT.

Band bending induced by charged molecules causes the increase or decrease in surface conductivity responsible for the gas response signal. The nuclear energy for PCB passing is observed in Fig. 9. It has maximum amount in the middle of the tube, because all atoms of SWNT applied on PCB in the middle of tube. The enthalpy difference for them is negative, which has exothermic interaction and spontaneous and PCB is separated from airwater in environment.

Thermodynamic parameters ( $\Delta G_{ele}$ ,  $\Delta H_{ele}$ , and  $\Delta S_{ele}$ ) are calculated for the passing from top to middle and middle to end of tube length, and the results distinguished that the nature of passing to inter SWNT is exothermic, spontaneous and favorable in initial of tube, so it is endothermic in end. According to the values of electronic energy and enthalpy, the passing of PCB to inter SWNT is spontaneous and strong, that is showed in Table 2, so this method is suggested for removal for PCB in environment.

<b>Table 2.</b> Thermodynamic properties of interaction PCB on the both end SWN1
--

Distance (nm)	$\Delta G_{ m ele} \ (MJ/mol)$	ΔH <sub>ele</sub> (MJ/mol)	ΔS <sub>ele</sub> (MJ/mol)	lnK
0-0.493 (top to middle tube)	-10896.84	-7987.19	9.76	$4.4 \times 10^{6}$
0.985-2.135 (Middle to end tube)	4114.01	-556.27	-15.67	-1.7×10 <sup>6</sup>

### CONCULUSION

PCBs compounds contain 209 congeners, each of which is chlorinated to various degrees. These

compounds are still detectable, even at the edge of the arctic and in the depths of the oceans.

After absorption, PCBs undergo little catalysis. They are highly lipophilic, accumulate in the liver and adipose tissue, and easily transfer to the embryo through the placenta and via breast milk. Thus, PCB contamination is inheritable. At low doses, PCB might affect embryonic and neonatal development. Brain development seems to be particularly affected by PCB.

The high thermal and chemical resistance of PCBs means that they do not readily break down when exposed to heat or chemical treatment. The toxic congeners possess conformations with low dipole moments, a fact that may be linked to the ready accumulation on the adipose tissue. The results on the geometry and electrostatic properties of chlorinated biphenyls can be useful to rationalize their selective toxicities.

In waste and water pollution, there are some other contaminants, which they make trouble for

### REFERENCES

- [1] V. Tigini, V. Prigione, S. Di Toro, F. Fava, and G. C Varese. *Microb Cell Fact*. 8 (2009) 5.
- [2] J. Kaiser. Science. 288 (2000) 1941.
- [3] J. Kaiser. Science. 288 (2000) 1313.
- [4] J. Borja, DM. Taleon, J. Auresenia, S. Gallardo. *Process Biochem.* 40 (2005)1999.
- [5] GK. Vasilyeva, ER. Strijakova. *Microbiology*. 76 (2007) 639.
- [6] W. Miyazaki, T. Iwasaki, A. Takeshita, Y. Kuroda, and N. Koibuchi. *J. Bio. Chem*, 279, 18 (2004) 18195.
- [7] R. T. Yang, R. Q. Long, J. Padin, A. Takahashi and T. Takahashi. *Ind. Eng. Chem. Res.* 38 (1999) 2726.
- [8] R. Q. Long and R. T. Yang, *J. Am. Chem. Soc* 123 (2001) 2058.
- [9] J. Zhao, A. Buldum, J. Han, J.P. Lu. *Nanotechnology*. 13 (2002)195.
- [10] Y. Cai, G. Jiang, J. Liu and Q. Zhou, *Anal. Chem.* 75 (2003) 2517.
- [11] Y. Cai, G. Jiang, J. Liu and Q. Zhou, *Anal. Chim. Acta.* 494 (2003) 149.
- [12] Y. Doo Lee, W. S. Cho, S.I. Moon, Y.H. Lee, J. K. Kim, S. Nahm, B.K. Ju. *Chemical Physics Letters*. 433 (2006)105.
- [13] R. G.Olmos and M. Iglesias. *J. Chem. Eng. Data*, 53 (11) (2008) 2556.

separated PCBs by SWNT, but in this study, we neglected them and just calculated interaction between them.

Then for removal of PCBs in environment, we suggest using SWNT for it; SWNT is a good nano-filter for pollution in environment. For this work, we choose armchair carbon nanotube (6,6) and is investigated PCB passing through it. The interaction between them is calculated by ab initio. A change in the potential of all atoms of inter surface of SWNT is observed passing result of PCB. These surface phases have different properties resulting in altered molecule–surface interactions. In SWNT (6,6), there are two high potential places at 0.493 and 0.985 nm, respectively. These places of inter nanotube are snared PCB and can eliminate it in environment.

- [14] S. J. Kim., S. H. Lee., and C. J. Lee. *J. phys. D.* 34, **2001**, 3505.
- [15] S. Kawanishi, T. Mizutani and S. Sano. *Biochim. Biophys. Acta.* 540 (1978) 83.
- [16] S. Sano, S. Kawanishi and Y. Seki. *Environmental health perspectives*. 59 (1985) 137.
- [17] M. Okumura, T. Akita, M. Haruta, X. Wang, O. Kajikawa and O. Okada. *Appl. Catal. B*, 41 (2003) 43-52.
- [18] P. Hohenberg, W. Kohn. *Phys. Rev.* 136 (1964) B864.
- [19] C. Lee, W. Yang, G.R. Parr. *Phys. Rev. B*. 37 (1988) 785.
- [20] M. J. S. Dewar and M. L. McKee. *J. Am. Chem. Soc.* 99 (1977) 5231.
- [21] M. J. S. Dewar and W. Thiel. *J. Am. Chem. Soc.* 99 (1977) 4899.
- [22] M. P. Longnecker, M. S. Wolff, B. C. Gladen, J. W. Brock, Ph. Grandjean, J. L. Jacobson, S. A. Korrick, W. J. Rogan, N. Weisglas-Kuperus, I. Hertz-Picciotto, P. Ayotte, P. Stewart, G. Winneke, M. J. Charles, S. W. Jacobson, É. Dewailly, E. R. Boersma, L. M. Altshul, B. Heinzow, J. J. Pagano, and A. A. Jensen. Environmental Health Perspectives.111 (1) (2003) 65.