

# Antibiofilm Efficacy of Positively Charged Imidazolium-Based Silver Nanoparticles in *Enterococcus faecalis* Using Quantitative Real-Time PCR

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## Abstract

**Background:** Nanoparticles are made by different methodologies, which can affect the particle's features. Recently, imidazolium-coated silver nanoparticles with a positive surface charge (PC Im-based AgNPs) have revealed favorable results as a root canal disinfectant. However, the antibacterial potency of these particles against biofilm form of *Enterococcus faecalis*, as the most resistant organism to eliminate in endodontic treatment, has not been investigated. It can be noted that removing this microorganism is associated with extremely effective disinfection.

**Objectives:** This study investigated the antibacterial efficacy of PC Im-based AgNPs at  $5.7 \times 10^{-8}$  mol L<sup>-1</sup> in comparison with 2.5% sodium hypochlorite (NaOCl) and 2% chlorhexidine as the two broadly used endodontic irrigation solutions against biofilm *E. faecalis* using quantitative real-time polymerase chain reaction.

**Methods:** In total, 48 premolar teeth with a single root were infected with *E. faecalis* and then prepared with ProTaper rotary instruments. The samples were randomly allocated into 4 groups of 12 samples. Sterile saline, PC Im-based AgNPs, NaOCl, and chlorhexidine were used as irrigants. Sampling the root canals was implemented with paper points and Gates-Glidden drills. The reduction in *E. faecalis* counts was calculated and statistically analyzed by means of the Kruskal-Wallis and Mann-Whitney U tests.

**Results:** Irrigation with PC Im-based AgNPs or NaOCl was significantly more effective in bacterial count reduction compared to irrigation with chlorhexidine or sterile saline ( $P < 0.05$ ). There was no significant difference between PC Im-based AgNPs and NaOCl irrigants when either Gates-Glidden drills ( $P = 0.751$ ) or paper points ( $P = 0.488$ ) were employed. Chlorhexidine was significantly less efficient than PC Im-based AgNPs and NaOCl solutions ( $P < 0.001$ ); however, it was significantly better than sterile saline in both sampling approaches ( $P > 0.001$ ).

**Conclusions:** The PC Im-based AgNP solution revealed promising results as a root canal irrigant. This solution at  $5.7 \times 10^{-8}$  mol L<sup>-1</sup> was effectively able to eliminate biofilm *E. faecalis* and this was not significantly different from that of 2.5% NaOCl.

**Keywords:** Ionic Liquids, Real Time Polymerase Chain Reaction, Root Canal Therapy, Nanoparticles, *Enterococcus faecalis*

## 1. Background

Root canal infections are usually known as biofilm-related infections induced by intracanal microorganisms (1). The success of endodontic treatment generally is determined by the competence of root canal disinfection. However, eliminating bacteria and their metabolites may not be thoroughly achieved because of the anatomical complexity of the root canal system, the buffering action of dentin, and the persistence of microorganisms (2, 3). *En-*

*terococcus faecalis* is one of the most resistant organisms that survive after root canal treatments and play an important role in persistent periradicular lesions (4, 5). This bacterium is capable to persist in strict nutritional states within the root canal system, maintain pH homeostasis, attack the dentinal tubules, and attach to dentin and collagens in the biofilm form (6, 7). Hence, irrigants, which could remove this bacterium from the root canal space, are essential for successful disinfection.

Sodium hypochlorite (NaOCl) and chlorhexidine are

the most broadly used endodontic irrigation solutions with exemplary antibacterial effectiveness to root canal pathogens, including *E. faecalis*. Despite favourable qualities, NaOCl has disadvantages including the caustic effect, if inadvertently extruded to the periapical region, and the corrosive effect on metal objects (8, 9). The lack of ability to remove tissue remnants and little capability to kill gram-negative bacteria are the two major shortcomings of chlorhexidine, which makes it a complement rather than a major endodontic irrigant (10, 11). Therefore, studies are continually attempting to investigate and find efficient, non-toxic, and cost-effective disinfectants for root canal treatment.

In recent years, trends in application of silver nanoparticles (AgNPs) as antimicrobial mediators have developed in many fields, including endodontics, due to their high level of biocompatibility, particularly in low concentrations (12) and their broad spectrum of antibacterial efficacy (13-15). It has been speculated that AgNPs can attach to the cell wall of organisms and cause dysfunction in cellular permeability. Furthermore, they may enter the bacteria and affect their DNA and enzymes, eventually causing cell extermination (16, 17). The interaction between AgNPs and the target cells happens in an extremely short fraction of time because AgNPs have large surface area to volume ratio (18, 19). Therefore, AgNPs appear to prevent the development of resistance by the bacteria (17).

In a recent study evaluating the potential role of AgNPs in future endodontic practice, the positively charged imidazolium-based silver nanoparticles (PC Im-based AgNPs) have revealed promising results as a root canal disinfectant (20). Later on, it has been found that these NPs could affect the physicochemical properties of dentin and raised its surface roughness (21). Previous findings proved the outstanding antibacterial efficacy of these NPs against planktonic *E. faecalis* with or without dentin and their great cytocompatibility level. However, the antibacterial potency of these particles against biofilm *E. faecalis* has not been investigated.

## 2. Objectives

The aim of this study was to assess the antibacterial efficacy of these particles in root canals infected with *E. faecalis* relative to 2.5% NaOCl and 2% chlorhexidine using quantitative real-time polymerase reaction (qPCR).

## 3. Methods

### 3.1. Ethics Statement

This study was approved by the ethics committee of Shiraz University of Medical Sciences (code: 5763).

### 3.2. Irrigants

The irrigants in this study were PC Im-based Ag NP solution (average size of 9 nm) at  $5.7 \times 10^{-8}$  mol L<sup>-1</sup>, which was synthesized based on the method defined by Abbaszadegan et al. (20), 2.5% NaOCl (Sigma Aldrich Co., St. Louis, MO, USA), 2% chlorhexidine (FGM, Joinville, PR, Brazil), and 0.9 % sterile saline.

### 3.3. Sample Collection and Preparation

Forty-eight human premolar teeth with a single root and closed apices without any signs of fracture or caries were selected to be studied. These teeth were gathered from a group of teeth that had been extracted for orthodontic purposes. Each tooth was radiographed. Teeth with more than single canal or with complex anatomy or root length more or less than 12 mm were excluded. Any residual periodontal tissue or calculus was cleared from the root surfaces with an ultrasonic scaler. After disinfection of the teeth with 5.25% NaOCl for 1 h, they were kept in saline solution before use.

The access cavity was prepared. The working length was determined. To standardize the apical constriction diameter, each root canal was reamed and enlarged by means of a #20 K-file (Mani Inc, Japan). The removal of smear layer was done using 5 mL 17% EDTA (Dia-Prep Plus, Diadent Group International Inc., Chongju, Korea) for 3 minutes and 5 mL 2.5% NaOCl irrigation. Afterwards, to neutralize the remaining NaOCl, 5% sodium thiosulfate (Merck, Germany) was used for 1 minute.

The teeth were dipped in sterile brain heart infusion (BHI) (Himedia, India) broth and ultrasonicated for 1 minute. This was done in order to release any trapped air and to help penetration of the culture media into the irregularities of root canals. Afterwards, they were autoclaved for 20 minutes at 121°C. Every 12 teeth were placed in a flask and then, each flask was immersed in BHI broth. Then, they were kept warm at 37°C for 24 hours and tested again to assess the efficiency of the sterilization procedures.

### 3.4. Bacterial Biofilm Formation

The pure strain of *E. faecalis* (PTCC1394) was used in this study. The samples were prepared for 24 hours at 37°C in BHI using a shaker incubator (G-25 Shaker Incubator, New Brunswick Scientific Co. Inc., New Brunswick, NJ, USA) at 150 rpm. Microbial growth was assessed based on turbidity changes over 24 hours. To prepare bacterial suspension 1 mL of a pure culture of *E. faecalis* was added to 5 mL of fresh BHI. Further, cell suspension was set spectrophotometrically at  $1.5 \times 10^8$  CFU mL<sup>-1</sup>, which is equivalent to 0.5 McFarland standards. To inoculate each flask, 1 mL of this suspension was used. *Enterococcus faecalis* was grown up

for 21 days where it was under a gentle shaking at the temperature of 37°C.

Every other day, half of the culture media contaminated with the microorganism was replaced with fresh BHI under a laminar-flow hood to avoid medium saturation. Following the contamination period, four teeth were randomly observed using a scanning electron microscope (JSM-5800LV; JEOL, Tokyo, Japan) to confirm the colonization and biofilm formation of *E. faecalis*. The experimental teeth (n = 12 in each group) were smoothly washed with phosphate-buffered saline to eliminate any remaining culture medium or floating bacteria. The apex of all teeth became impermeable with epoxy resin. Later, to expedite the handling and identification, the teeth were fixed vertically up to the cervical area using blocks made from silicon impression materials. Next, the outer surfaces of the crowns and the assemblies were decontaminated using 2.5% NaOCl. In total, in order to inactivate NaOCl, 5% sodium thiosulfate was used for 1 minute. Later, initial (S1) sampling was conducted by three paper cones (Orca, China), which were placed in the root canals for 1 minute.

### 3.5. Irrigation Procedure

The teeth were coded and randomly allocated into four groups based on the irrigants used as follows:

- Group 1 (n = 12): Sterile saline
- Group 2 (n = 12): PC Im-based AgNPs
- Group 3 (n = 12): 2.5% NaOCl
- Group 4 (n = 12): 2% CHX

All the root canals were cleaned and shaped in a similar way by ProTaper rotary instruments (Dentsply, Switzerland) following the instruction conducted by the manufacturer up to file size F3. Irrigation of the root canals was done in each group by employing 2 mL of the designated solutions after changing each file (total of 10 mL). This was done with the application of disposable syringes via 27-gauge needles (NaviTip, Ultradent, South Jordan, UT) that was taken along to the canal long axis up to 1 mm short of the working length. The total time for the preparation of each sample was taken about 20 minutes. Before taking post-instrumentation (S2) samples, each canal was washed using the following solutions for 1 minute to inactivate the related irrigants: 0.07% lecithin and 0.5% Tween 80 for CHX, 0.4% sodium thioglycolate for AgNPs, and 5% sodium thiosulfate for NaOCl. Then, S2 samples were obtained sequentially using three #30-paper cones placed to the working length for 1 minute.

Furthermore, dentin chips were collected by the sequential use of Gates-Glidden drills sizes no. 3, 4, and 5 (Mani, Tochigi, Japan). Averagely, each Gates-Glidden drill was employed for 10 seconds for three times up to 10 mm of

the canal length. Then, the collected shavings were maintained at -70°C in an incubator for 4 days until a fixed weight was obtained. Thereafter, they were confirmed to have similar weights using an electronic balance (AX200; Shimadzu Corp, Kyoto, Japan) and transferred to tryptic soy broth. The aliquots were analyzed by quantitative real-time PCR (Bio-Rad, Hercules, CA, USA) to obtain the threshold cycle (CT) value of the samples. Assessment of the samples was conducted with real-time PCR.

### 3.6. DNA Extraction and Primer Design

A bacterial genomic DNA isolation Kit (Promega, Madison, WI) was used to isolate the genomic DNA, based on the recommended protocol by the manufacturer, and it was scaled down to accommodate a 100 sample. UV-VIS spectrophotometry was employed to determine the purity and concentration of DNA. Pure *E. faecalis*-obtained DNA was diluted in sterile Tris-EDTA buffer at pH 8 and stored at -20°C before the commencement of the experiment (22). The primers were designed and the specificity of the primer sequences was identified similar to the method suggested by Malorny et al. (23).

### 3.7. Efficiency of the Reaction and Determination of the Linearity

Negative controls were set in duplicate along with the experimental total DNA samples in order to normalize any probable background signals following amplification. To determine *E. faecalis* counts, total DNA standards from an *E. faecalis* laboratory strain were prepared in 10-fold dilutions from 10 ng mL<sup>-1</sup> to 100 fg mL<sup>-1</sup>. Real time PCR was implemented. Based on data analysis, the standard curves were obtained from a series of dilution in concentration range of 2 × 10<sup>6</sup> to 2 CFU/mL organisms. Then, standard curves were created for the extracted, quantified, and amplified DNA. The efficiency of the amplification and the correlation coefficient (R<sup>2</sup>) were determined (22-24).

### 3.8. Statistical Analysis

The normality of data was evaluated by Kolmogorov-Smirnov test. The Wilcoxon signed rank test was employed to evaluate the reduction in *E. faecalis* counts from S1 to S2 samples. Intergroup analysis was performed using Kruskal-Wallis and Mann-Whitney U tests. The correlation between the two sampling methods (using paper point vs. Gate-Glidden drills) was determined by Spearman's correlation coefficient test. SPSS 15.0 software package (SPSS, Chicago, IL, USA) was employed for all statistical analyses. A P value below 0.05 was considered significant.

#### 4. Results

Scanning electron microscopic analysis of the four control teeth revealed successful colonization of *E. faecalis* in the root canal walls. This was also reconfirmed by the results of qPCR in S1 samples of all experimental teeth. Intragroup analyses demonstrated a highly significant bacterial reduction from S1 to S2 for both sampling methods in all the experimental groups ( $P < 0.001$ ). Table 1 demonstrates the mean and median of the number of CFUs obtained for all groups. Intergroup analysis of S1 samples revealed no significant differences ( $P = 0.147$ ) between the groups, indicating an identical reference for bacterial load induced by the employed method of contamination.

Irrigation with AgNPs or NaOCl was significantly more effective in bacterial count reduction compared to irrigation with chlorhexidine or sterile saline ( $P < 0.05$ ). No significant differences were observed between AgNP and NaOCl irrigants when either Gates-Glidden drills ( $P = 0.751$ ) or paper points ( $P = 0.488$ ) were employed through different sampling approaches. Chlorhexidine was significantly less efficient than AgNPs and NaOCl solutions ( $P < 0.001$ ); however, it was significantly better than sterile saline in both sampling approaches ( $P > 0.001$ ).

#### 5. Discussion

Mature biofilm formation requires a multiparty process. It starts with a linkage of an organism to a surface and continues with cellular interaction and colony formation (25). Likewise, bacterial biofilms in root canal are a population of cells, enclosed in a hydrated matrix of exopolymeric substances, nucleic acids, proteins, polysaccharides that are attached irreversibly to dentin surfaces (26). In comparison with the planktonic form, bacteria in a biofilm structure behave in a different mode because the physiological characteristics of biofilms provide a safety shelter for them against antibacterial irrigation solutions (27, 28).

Enterococci, known as opportunistic microorganisms, are normal residents of the oral cavity. *Enterococcus faecalis* as the most well-known enterococci species is in charge of almost all persistent periradicular lesions (4, 5). Recent studies have reported the excellent antibacterial activity of different synthesized AgNPs against planktonic *E. faecalis* (29-31). However, these studies were limited by the lack of investigation on antimicrobial properties of these nanoparticles against the biofilm *E. faecalis*. The current in vitro study was directed to shed light on the antimicrobial effectiveness of PC Im-based AgNPs when employed as a clinical irrigant against *E. faecalis* and compare it with the current clinical standards 2.5% NaOCl and 2% CHX.

We found that the nanosilver solution at  $5.7 \times 10^{-8}$  mol  $L^{-1}$  was equally potent as 2.5% NaOCl against the biofilm form of *E. faecalis*. Electrostatic interaction plays a major role in antibacterial activity of AgNP, bearing in mind the negative charge of the bacteria due to their surface proteins (32). These proteins are necessary for membrane transport, cell wall synthesis, and other vital cellular roles (33). The synthesized AgNPs in this study were 9 nm in size. The particles with small sizes can provide a greater surface area and are able easily to reach the bacterial nuclear content (34). Moreover, previous studies have reported that small nanoparticles with the size ranging from 1 to 10 nm had the highest antibacterial effectiveness due to their ability to bind to the cell membrane and weakening its respiration and permeability (14, 35).

The results of the present investigation indicated that all experimental irrigants, including sterile saline, could promote a significant reduction in intracanal bacterial counts. This was in line with several previous studies showing the efficacy of the flushing action (36-38). Studies comparing the antibacterial effectiveness of NaOCl and chlorhexidine against *E. faecalis* revealed inconsistent results. While some studies found that NaOCl has lower effectiveness against *E. faecalis* (39-41), others indicated that chlorhexidine is less effective (42, 43) and many studies show little or no difference between their antibacterial efficacy (38, 44, 45). The results of our study indicated that NaOCl was significantly more effective than chlorhexidine in reducing *E. faecalis* biofilm. The discrepancy between the results of studies might be attributed to the different methodology used.

We also revealed that AgNPs and NaOCl were significantly more effective than chlorhexidine or normal saline; however, no significant difference was detected between AgNPs and NaOCl. On the contrary, Wu et al. (46) showed that AgNPs were unable to eliminate the residual biofilm of *E. faecalis* formed on dentinal blocks when used as an irrigant rather than a medicament. In their study, a short-time irrigation of 2 min was employed while in the present study a 20-min irrigation protocol was used in a simulated clinical condition. Importantly, when the resistance of biofilm polymeric matrix is considered, longer contact times may provide variation in the observed results, leading to different conclusions. Inconsistent with the findings of Wu et al. study (46), Mozayeni and co-workers (47) concluded that AgNP gel was ineffective against *E. faecalis* when compared to calcium hydroxide, 2% chlorhexidine gel, and triple antibiotic paste as intracanal medicaments. This discrepancy may be due to different applied concentrations of AgNPs, different synthesis procedures, their relevant physical properties, and the possible effect of carriers on inhibiting the release of AgNPs from different for-

**Table 1.** CFU Counts in Pre- and Post-Instrumentation Samples<sup>a, b</sup>

Group	S1	S2 (Paper Point)	S2 (Gates-Glidden Drills)
Sterile saline	$3.60 \times 10^9$ ( $4.24 \times 10^9 \pm 2.86 \times 10^9$ )Aa	$1.08 \times 10^9$ ( $1.76 \times 10^9 \pm 1.94 \times 10^9$ )Ab	$0.16 \times 10^9$ ( $0.41 \times 10^9 \pm 0.62 \times 10^9$ )Ab
PC Im-based AgNPs	$2.42 \times 10^9$ ( $4.98 \times 10^9 \pm 8.01 \times 10^9$ )Aa	$1.41 \times 10^2$ ( $2.26 \times 10^2 \pm 1.92 \times 10^2$ )Bb	$8.15 \times 10^1$ ( $9.65 \times 10^1 \pm 7.66 \times 10^1$ )Bb
NaOCl	$2.22 \times 10^9$ ( $6.94 \times 10^9 \pm 10.37 \times 10^9$ )Aa	$1.50 \times 10^2$ ( $2.11 \times 10^2 \pm 1.86 \times 10^2$ )Bb	$7.15 \times 10^1$ ( $8.83 \times 10^1 \pm 8.17 \times 10^1$ )Bb
CHX	$1.35 \times 10^9$ ( $3.03 \times 10^9 \pm 4.42 \times 10^9$ )Aa	$4.28 \times 10^2$ ( $9.20 \times 10^2 \pm 1.24 \times 10^3$ )Cb	$9.32 \times 10^2$ ( $1.57 \times 10^3 \pm 1.87 \times 10^3$ )Cb

<sup>a</sup>Read vertically, uppercase letters denote between-group comparisons regarding the number of CFUs in S1 or S2 samples. Read horizontally, lowercase letters denote pre-instrumentation and post-instrumentation CFU comparisons in each group. Equal letters denotes lack of statistically significant difference.

<sup>b</sup>Data here are as Median (mean  $\pm$  SD).

mulation of medicaments.

Furthermore, in this study, extracted human teeth, which might better simulate the clinical setting as compared to dentinal blocks or culture medium, were used. Accordingly, it was possible to accompany the instrumentation procedures with irrigation protocols that mimic clinical situations. In this study, the apical diameter of each root canal was first standardized to K-file #20. After that, root canals were sterilized, inoculated, and prepared and the efficacy of the test irrigants was examined. This methodology was similar to the method of a previous study by Tran et al. (48) and it was adopted because it was closer to clinical condition. Microbiological sampling by Gates-Glidden drills up to size # 5 was selected to harvest dentinal debris up to 400-micron depth because of the known propensity of *E. faecalis* to proliferate in the deeper layers of dentinal tubes. In addition, sterile paper points were employed to engage the organisms floating in the root canal space.

In this study, the bacterial colony counts of microorganisms inside the dentinal tubules were evaluated by qPCR. PCR is a sensitive method for quick detection of microbial species that are difficult or impossible to culture (49). Real time PCR enables us to detect individual target species along with the quantification of bacteria in clinical samples. The other advantage of this technique is its limited contamination due to the avoidance of post amplification manipulation. One limitation of this study was that teeth with a single root and simple anatomic structures were used, which may differ from the clinical conditions. In addition, bacterial species were cultured under optimal in vitro conditions, which included nutrients that are rarely found in clinical settings. Therefore, data extrapolation to clinical situations should be made with caution because of inherent limitation of in vitro studies. Further investigations are suggested to determine the tooth staining, allergic potential, and long-term safety of this new irrigant.

### 5.1. Conclusions

The PC Im-based AgNP solution revealed promising results proposed for future endodontic treatments. This irrigant at  $5.7 \times 10^{-8}$  mol L<sup>-1</sup> was effectively able to eliminate biofilm *E. faecalis* and this was not significantly different from the effectiveness of 2.5% NaOCl.

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