

A Rapid Combinatorial Library Synthesis of Benzazolo[2,1-*b*]quinazolinones and Triazolo[2,1-*b*]quinazolinones

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ABSTRACT: Benzazolo [2,1-*b*]quinazolinones and triazolo[2,1-*b*]quinazolinones were synthesized in high yields by the condensation reaction of an aldehyde and a cyclic β -diketone with 2 - aminobenzimidazole, 2 - aminobenzothiazole or 1, 2, 4 - triazole derivatives in 1 - butyl - 3 - methylimidazolium bromide as an ionic liquid under classic heating conditions within 5-60 minutes.

KEY WORDS: 2-Aminobenzimidazole, 2-Aminobenzothiazole, 3-Amino-1,2,4-triazole, 3,5-Diamino-1, 2,4-triazole, 3-Amino-5-thiomethoxy-1,2,4-triazole, Ionic liquid.

INTRODUCTION

Quinazolines are reported to have a broad spectra of biological activities such as propyl hydroxylase inhibitor [1], antidiabetics [2], anti-inflammatory [3], antiviral [4], antimicrobial [5], antineoplastic [6], and potent immunosuppressive agents [7]. They also emerged as an integral backbones of calcium channel blockers[8]. So, the analogs of these heterocycles show potent antitumor activity[9,10]. Quinazoline heterocycles are the subunit of several natural alkaloids and active pharmaceuticals [1,11,12] (Scheme 1).

The useful methods, have been reported for synthesis of tetrahydrobenzimidazo [2,1-*b*] quinazolin-1(2*H*)-ones and tetrahydro-1,2,4-triazolo[5,1-*b*]quinazolin-8(4*H*)-ones ring system skeletons, are the condensation of aminoazoles with benzilidene compounds[13-16] or three-component condensation of 2-aminobenzothiazole or 2-aminobenzimidazole and an aldehyde with cyclic β -diketone [17-18]. However, these synthetic protocols

suffer from one or more disadvantages such as harsh reaction conditions, prolonged reaction times, or use of hazardous and often expensive acid catalysts. Moreover, the synthesis of these heterocycles is usually carried out in polar solvents such as methanol, acetic acid, THF and DMF leading to complex isolation and recovery procedures. These processes also generate waste containing both catalyst and solvent, which have to be recovered, treated, and disposed off.

Recently, Room-Temperature Ionic Liquids (RTILs) have attracted much attention as promising alternative 'green' solvents to hazardous traditional organic solvents, due to their properties such as non-flammability, negligible vapor pressure, high thermal stability, solvating ability and easy recyclability. They have the potential to be highly polar yet non-coordinating [19-23]. In addition to the above-mentioned salient features of Ionic Liquids (ILs) as reaction media, we have also

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Table 1: Solvent effects on the reaction of dimedone and *p*-methylbenzaldehyde with 2-aminobezimidazole on the synthesis of 3,3-dimethyl-12-(4-methyl-phenyl)-3,4,5,12-tetrahydrobenzimidazo[2,1-*b*]quinazolin-1(2*H*)-one 5b^a.

Entry	Solvent	Time (h)	Temperature (°C)	Yield (%)
1	CH ₂ Cl ₂	2	Reflux	0
2	CHCl ₃	2	Reflux	0
3	CH ₃ COOEt	2	Reflux	0
4	CH ₃ C ₆ H ₅	2	Reflux	0
5	CH ₃ CN	2	Reflux	25
6	CH ₃ OH	2	Reflux	43
7	CH ₃ CH ₂ OH	2	Reflux	66
8	Water	6	80	trace
9	[Bmim]Br	2	80	81
10	[Bmim]Br	1	80	79
11	[Bmim]Br	0.5	80	79
12	-	2	80	trace

a) 2-Aminobenzimidazole (1 mmol), dimedone (1 mmol) and *p*-methylbenzaldehyde (3 mmol).

were obtained on solutions in DMSO-d⁶. The elemental analysis was performed with an Elementar Analysensysteme GmbH VarioEL CHNS mode. All the products are known compounds, and were characterized by melting points, IR, ¹H NMR, ¹³C NMR and mass spectra. In some cases we used CHNS for compounds which because of low solubility we could not obtain ¹³C NMR.

General procedure for synthesis of 3,4,5,12-tetrahydrobenzimidazo [2,1-*b*]quinazolin-1(2*H*)-one (5a-*r*)

A mixture of cyclic β-diketones (1 mmol), aldehydes (3 mmol), and 2-aminobenzazoles (1 mmol) was successively added to a screw-capped vial containing [bmim]Br (0.3 g) and a magnetic stirring bar. The mixture was heated at 100 °C in a preheated oil bath for 5-60 min. Then, the reaction mixture was washed with cold water and filtered. The solid residue was washed with acetone (3×5 mL) to produce **5** as a white powder in high yield.

General procedure for synthesis of tetrahydro-1,2,4-triazolo[5,1-*b*]quinazolin-8(4*H*)-ones(6a-*r*)

A mixture of cyclic β-diketones (1 mmol), aldehydes (1 mmol), and amino-1,2,4-triazoles (1 mmol) was added to a screw-capped vial containing [bmim]Br (0.3 g) and a magnetic stirring bar. The reaction mixture was heated

at 80 °C in a preheated oil bath for 15-30 min. Then, the reaction mixture was washed with cold water and filtered. The solid residue was washed with acetone (3×5 mL) to produce **6a-r** as a white powder in high yields.

RESULTS AND DISCUSSION

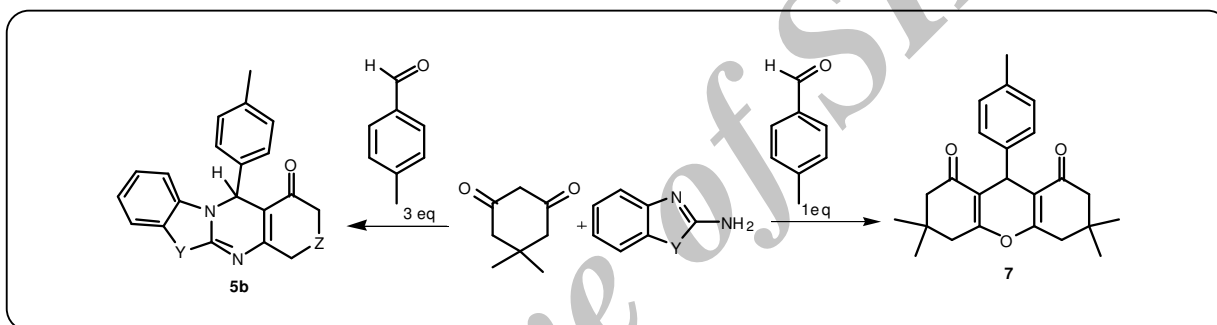
The one-pot method involves the classical heating of an aldehyde **1** and a cyclic β-diketone **2** with 2-aminobenzothiazole or 2-aminobezimidazole **3** or 3-amino-1,2,4-triazole or 3,5-diamino-1,2,4-triazole or 3-amino-5-thiomethoxy-1,2,4-triazole **4** without using any catalyst in [bmim]Br to give a family of tetraheterocyclic tetrahydrobenzimidazo[2,1-*b*]quinazolin-1(2*H*)-ones **5** and triheterocyclic tetrahydro-1,2,4-triazolo[5,1-*b*]quinazolin-8(4*H*)-ones **6** in high yields.

To achieve suitable conditions for the above transformations, series of experiments were carried out (Tables 1 and 2). First we investigated the reaction of dimedone and *p*-methylbenzaldehyde with 2-aminobezimidazole under solvent-free conditions, various solvents, water and ionic liquids under classic heating conditions. In the absence of solvent, the reaction was very slow and the yield of the product was trace at 80 °C after 2 h (Table 1, Entry 12). The reaction did not proceed in aprotic organic solvents even under reflux conditions after 2 h (Table 1, Entries 1-5). The yield of product was improved in protic organic solvents (Table 1, Entries 6-7), but we

Table 2: One-pot synthesis of tetrahydrobenzimidazo[2,1-b]quinazolin-1(2H)-one 5b at different ionic liquid at 30 min^a.

Entry	Ionic liquid	Temperature (°C)	Yield (%)
1	BPB	80	48
2	TMGT	80	43
3	TMGA	80	57
4	TBAB	80	81
5	TBAC	80	79
6	MIT	80	64
7	[bmim]Br	80	89
8	[bmim]Br	100	96
9	[bmim]Br	120	96
10	[bmim]PF ₆	80	82
11	-	80	trace

a) 2-Aminobenzimidazole (1 mmol), dimedon (1 mmol) and p-methylbenzaldehyde (3 mmol).

**Scheme 3: Reaction optimization for the synthesis of 5b**

found the reaction yield was trace in water after 6 h at 80 °C (Table 1, Entry 8). The best solvent was found to be TetraButyl Ammonium Bromide (TBAB) and the reaction yield was 79% after heating 30 min at 80°C (Table 1, Entry 11).

In order to examine the best ionic liquid, different ionic liquids such as Butyl Pyridinium Bromide (BPB), Tetra Methyl Guanidinium Trifluoroacetate (TMGT), Tetra Methyl Guanidinium Acetate (TMGA), TetraButyl Ammonium Bromide (TBAB), TetraButyl Ammonium Chloride (TBAC), Methylimidazolium Trifluoroacetate (MIT), 1-butyl-3-methylimidazolium bromide ([bmim]Br), 1-butyl-3-methylimidazolium hexafluorophosphate ([bmim]PF₆) in this condensation reaction were applied (Table 2). In the course of this study it was found that 1-butyl-3-methylimidazolium bromide was the best ionic liquid for this reaction in terms of yield, price and easy at work-up.

To illustrate the need for [bmim]Br in these reactions, the reaction of aldehyde **1**, cyclic β -diketone **2** and 2-aminobenzimidazole **3** was studied in the absence of [bmim]Br. The yield of product was trace at 80 °C after 2 h (Table 2, Entry 11).

In order to improve the yields, we performed reaction using different quantities of reagents. The best result was obtained with 3:1:1 molar ratios of aldehyde, cyclic β -diketone and 2-aminobenzimidazole or 2-aminobenzothiazole. However, the 1:1:1 molar ratios of these reagents gave the compound **7** (Scheme 3). In the case of 3-amino-1,2,4-triazole or 3,5-diamino-1,2,4-triazole or 3-amino-5-thiomethoxy-1,2,4-triazole, the optimum molar ratio was 1:1:1.

One of the advantages of Ionic Liquids (ILs) is the ability to be recyclable as reaction medium. We were able to separate [bmim]Br from reaction medium easily by washing with water and evaporating the solvent

Table 3: One-pot synthesis of tetraheterocyclic tetrahydrobenzimidazo[2,1-b]quinazolin-1(2H)-ones by the condensation reaction of an aldehyde, a cyclic β -diketone and 2-aminobenzothiazole or 2-aminobenzimidazole in [bmim]Br at 100 °C after 5-60 min.

Product	Y	Z	X	Yield (%)	Time(min)
5a	NH	C(CH ₃) ₂	H	94	5
5b	NH	C(CH ₃) ₂	4-CH ₃	96(94,93,92,93) ^a	10
5c	NH	C(CH ₃) ₂	4-OCH ₃	89	15
5d	NH	C(CH ₃) ₂	4-Cl	91	10
5e	NH	C(CH ₃) ₂	4-Br	93	10
5f	NH	C(CH ₃) ₂	3-OCH ₃	85	15
5g	NH	C(CH ₃) ₂	3-Cl	89	5
5h	NH	C(CH ₃) ₂	3-Br	91	15
5i	NH	C(CH ₃) ₂	3-NO ₂	88	15
5j	NH	C(CH ₃) ₂	2-CH ₃	94	5
5k	NH	C(CH ₃) ₂	2-Cl	91	10
5l	NH	CH ₂	H	86	20
5m	NH	CH ₂	4-CH ₃	91	15
5n	NH	CH ₂	4-OCH ₃	88	20
5o	NH	CH ₂	3-NO ₂	90	15
5p	S	C(CH ₃) ₂	H	73	60
5q	S	C(CH ₃) ₂	2-OCH ₃	69	60
5r	S	C(CH ₃) ₂	3-OCH ₃	72	60

a) The same ionic liquid runs for four times.

under vacuum, and reuse it for subsequent reactions (Table 3, Product **5b**).

To explore the scope and limitations of this reaction, we have extended it to various *ortho*-, *meta*- and *para*-substituted benzaldehydes in the presence of 2-aminobenzimidazole, 2-aminobenzothiazole, 3-amino-

1,2,4-triazole and 3,5-diamino-1,2,4-triazole or 3-amino-5-thiomethoxy-1,2,4-triazole. As indicated in Tables 3 and 4, the reaction proceeded efficiently with benzaldehyde and electron-withdrawing and electron-donating *ortho*-, *meta*- and *para*-substituted benzaldehydes.

Table 4: One-pot synthesis of tetrahydro-1,2,4-triazolo[5,1-b]quinazolin-8(4H)-ones by the condensation reaction of an aldehyde, a cyclic β -diketone and 3-amino-1,2,4-triazole, 3,5-diamino-1,2,4-triazole, 3-amino-5-thiomethoxy-1,2,4-triazole in [bmim]Br at 80 °C after 15-30 min.

Product	Y	Z	X	Yield (%)	Time(min)
6a	H	C(CH ₃) ₂	H	93	20
6b	H	C(CH ₃) ₂	4-CH ₃	96	15
6c	H	C(CH ₃) ₂	4-OCH ₃	86	20
6d	H	C(CH ₃) ₂	4-Cl	87	20
6e	H	C(CH ₃) ₂	3-OCH ₃	89	20
6f	H	C(CH ₃) ₂	3-NO ₂	91	30
6g	H	C(CH ₃) ₂	2-CH ₃	93	20
6h	H	CH ₂	H	90	30
6i	H	CH ₂	4-CH ₃	94	20
6j	H	CH ₂	4-OCH ₃	85	20
6k	H	CH ₂	4-Cl	87	30
6l	H	CH ₂	3-NO ₂	89	30
6m	NH ₂	C(CH ₃) ₂	H	93	20
6n	NH ₂	C(CH ₃) ₂	4-CH ₃	87	20
6o	NH ₂	C(CH ₃) ₂	4-OCH ₃	84	20
6p	NH ₂	C(CH ₃) ₂	4-Cl	90	30
6q	SCH ₃	C(CH ₃) ₂	H	85	20
6r	SCH ₃	C(CH ₃) ₂	4-OCH ₃	84	20

CONCLUSIONS

Three - component condensation of 2-aminobenzimidazole, 2-aminobenzothiazole or various 3-amino-1,2,4-triazoles, aldehydes, and 1,3-diones in ionic liquid afforded the novel library of quinazoline molecules. It has been demonstrated that the ionic liquids are very effective in speeding up both the soluble supported multistep synthesis, as well as accelerating the rate of subsequent multicomponent reaction. Great substrate scope, high yields, short reaction time, pure products and easy work up are advantages of this procedure

in comparison to other methods.

This approach provides a high speed path for the rapid synthesis of molecular libraries with high degree of structural diversity.

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