

Optimal Concentration of Zinc Sulfate in Foliar Spray to Alleviate Salinity Stress in *Glycine soja*

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ABSTRACT

Zinc has previously been reported to alleviate salinity stress in plants. In this study, we monitored various biomass and chlorophyll fluorescence parameters to determine the optimum zinc sulfate concentration that can be used as foliar spray to alleviate salinity stress in *Glycine soja*. The plants subjected to a series of salinity levels (NaCl concentration of 0, 100, 200, and 300 mmol L⁻¹), applied via the nutrient solution, were sprayed with different concentrations of zinc sulfate (0, 5, 10, 15, 20, 25 μmol L⁻¹). The results showed that the biomass and chlorophyll fluorescence parameters of seedlings were significantly affected by salt stress ($P < 0.05$). However, zinc sulfate sprays helped the plants to cope with the stress condition. The zinc sulfate concentrations that helped *G. soja* to cope with the salinity stress of 100, 200, and 300 mmol L⁻¹ were 15 to 20, 15 to 20, and 10 to 20 μmol L⁻¹, respectively. Lower zinc concentration was ineffective in alleviating stress and higher zinc concentration inhibited plant growth because of toxicological damage to plants. The zinc sulfate spray of 15 μmol L⁻¹ was found to be the most appropriate at all salinity stress levels. The growth measurements such as true leaves part and dry weight of total seedlings were in agreement with the chlorophyll fluorescence parameters, indicating a visible enhancement of leaf photosynthetic activity at 10-20 μmol L⁻¹ zinc concentrations.

Keywords: Biomass, Chlorophyll fluorescence parameters, Nutrients, Photosynthetic activity, Salt tolerance, Soybean.

INTRODUCTION

Soybean [*Glycine max* (L) Merr.] is the most important grain legume crop that is used as an oil and protein source throughout the world (Keyser and Li, 1992). Due to domestication and selective plant breeding, cultivated soybean has lost much of its genetic variability (Concibido *et al.*, 2003). *Glycine soja* (wild soybean) is an annual self-pollinating plant and is believed to be ancestor of cultivated soybean (Liu *et al.*, 2010). It is widely distributed throughout the world including the far east of Russia, Korean peninsula, Japan,

and China (Wang and Takahata, 2007). As an herbaceous plant with outcrossing rates of 2.4–19.0%, it has been utilized as an important genetic resource for improving soybean (Kiang *et al.*, 1992, Fujita *et al.*, 1997, Wang *et al.*, 2008). In fact, QTLs for resistance to important pest and diseases as well as for enhanced yield have been successfully transferred from wild soybean to the cultivated species (Concibido *et al.*, 2003).

Soil salinity is one of the major environmental factors that limit plant growth and yield in many areas of the world (Munns, 2002; Geilfus *et al.*, 2010; Mantri *et al.*, 2012). Plants respond to salinity stress with

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biochemical and physiological changes in order to maintain their photosynthetic activity and retain water despite the high external osmotic pressure (Ashraf and Harris, 2004, Du et al., 2010). Tavakkoli et al. (2010) have recently shown that the reduction in growth of faba bean under salinity stress is chiefly caused by Cl^- toxicity. It has been reported that Na^+ and Cl^- toxicity can be alleviated by supplementing the soil with zinc (Alpaslan et al., 1999; Tavallali et al., 2010; Aregheore, 2012). Zinc (Zn) is an essential nutrient for optimum growth of crops, and plays critical role in protecting vital cellular components such as chlorophyll by preventing their oxidation (Cherif et al., 2010; Cakmak, 2000; Zago and Oteiza, 2001). However, excessive zinc in plants could disrupt the metabolic processes such as transpiration and photosynthesis, which could result in reduced growth because of restricted root development and leaf chlorosis (Bonnet et al., 2000; Rout and Das, 2003; Lingua et al., 2008; Andrade et al., 2009; Wang et al., 2009). Leaf chlorosis is mainly due to excessive zinc that inhibits chlorophyll biosynthesis, and disrupts the electron transport of photosystem II (Chaloub et al., 2005, Dhir et al., 2008).

Different methods of zinc application have been used to investigate the importance of zinc in alleviating salinity stress condition. These include soil application of zinc sulfate (Verma and Neue, 1984), treatment of seedlings with zinc sulfate before transplanting (Alpaslan et al., 1999; Tzortzakis, 2010), and foliar spray of zinc sulfate (Yildirim et al., 2008). Whilst all studies have found zinc to relieve symptoms of salt stress, continuous soil application of large amounts of zinc sulfate is not recommended as it may result in zinc accumulation in the soil, which is undesirable. Further, treatment of seedlings with zinc sulfate can only help plants tolerate salt stress for a short period. In comparison, foliar application of zinc sulfate is more feasible as small amounts of zinc can be sprayed on the plants when required. Currently, there is no data on the optimal zinc concentration that can be used as a foliar spray in soybean to alleviate salinity stress. In one recent study, using

hydroponic setting, $20 \mu\text{mol L}^{-1}$ zinc was shown to enhance the leaf chlorophyll and carotenoid content of *G. max* and *G. soja* seedlings under salt stress (Qu et al., 2009). Therefore, the objective of this study was to determine the optimum concentration of zinc that can be used in a foliar spray to alleviate salinity stress in *G. soja*. The assessment was supported by biometric observations and measurement of chlorophyll fluorescence.

MATERIALS AND METHODS

Plants Material and Growth Conditions

Seeds of *Glycine soja* were collected from the North Mountain area of Jinhua city ($29^{\circ}7' \text{N}$, $119^{\circ}35' \text{E}$, 1,128 m in altitude, Zhejiang, China), air dried and stored in paper seed packets at room temperature ($15\text{--}20^{\circ}\text{C}$). Mature seeds were sterilized with 10% hydrogen peroxide for 25 minutes and treated with 98% concentrated sulfuric acid to soften the seed coat. The seeds were subsequently washed with sterile distilled water and were sown in sand culture to germinate. The seedlings were transplanted into 15 cm diameter pots (five plants per pot) filled with sands. For the experiment, the pots were divided into four groups of six pots per group. The seedlings were watered with distilled water for a week. They were grown in a growth chamber at $25/20^{\circ}\text{C}$ day/night temperature and 70/90% day/night relative humidity. Eight days after transplantation, sodium chloride (NaCl) was added to the irrigation water in different concentrations of 0, 100, 200, 300 mmol L^{-1} (four concentrations of NaCl ; one for each group). At the same time, the six pots in each group were sprayed with different concentrations (0, 5, 10, 15, 20, 25 $\mu\text{mol L}^{-1}$) of zinc sulfate. All treatments were conducted in five biological replications.

Seedlings Biomass Measurement

The biomass was measured in separate segments as showed in Figure 1. Seedlings

were cut into different parts and fresh weight (FW) and dry weight (DW) of roots, cotyledon, and true leaves part (TLP), and total seedlings (TS) were recorded. The length and diameter of hypocotyls and epicotyls were also measured.

Chlorophyll Fluorescence Measurements

A MINIPAM (pulse-amplitude modulation) fluorometer (WALZ, Effeltrich, Germany) was used in this study as reported by Zheng *et al.* (2010) and Jiang *et al.* (2013). The samples were kept in the dark for 30 minutes before measuring F_v/F_m . Yield, qP , NPQ , and ETR of chlorophyll fluorescence were obtained over a range of PAR values from 0 to $1,800 \mu\text{mol m}^{-2} \text{s}^{-1}$ and light-adapted for nearly 10 minutes prior to measurements. The values were calculated with the Fluorescence Monitoring System (FMS) according to Genty *et al.* (1989). Measurements were replicated three times, using three plants each time.

Data Analysis

The experiments were conducted in a completely randomized block design with three replicates. The data was analyzed using SPSS version 17.0 (SPSS Inc., Chicago, USA). One-way ANOVA was performed on the mean \pm SD values at 95% significance. Duncan's multiple-range test was subsequently used and the differences between variables were considered significant set at $P < 0.05$.

RESULTS

Leaf Morphology

Leaf morphology characteristics under different NaCl stress were significantly affected by different concentrations of zinc (Figure 2). Under all salt treatments, the true leaves grown with zinc spray of $15 \mu\text{mol L}^{-1}$ were found to be healthier than leaves

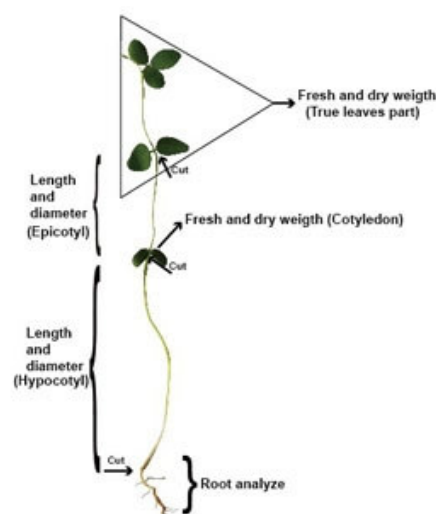


Figure 1. Schematic diagram of biomass measurement.

sprayed with other zinc concentrations. However, there was no significant difference in appearance of cotyledons. Under 300mmol L^{-1} NaCl treatment, the leaves withered and dropped at zinc concentrations of 20 and $25 \mu\text{mol L}^{-1}$, indicating toxic effects of high NaCl combined with high zinc concentration.

Effects of Zinc on Biomass of Seedlings

The biomass values of seedlings under different treatments are shown in (Table 1-2). The length or diameter of hypocotyls and epicotyls were significantly affected by zinc

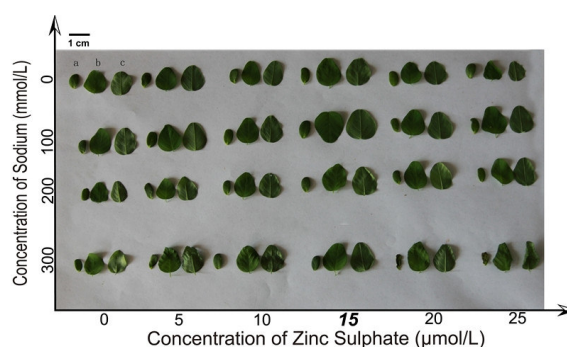


Figure 2. Leaves of *Glycine soja* seedlings sprayed with different zinc concentrations under various levels of NaCl stress. (a) Cotyledon; (b) Front surface of true leaf, (c) Back face of true leaf.

Table 1. Effects of zinc sprays on the seedling biomass under NaCl concentration of 0 and 100 mmol L⁻¹ for 7 days.^a

Plant part	Parameter	Zinc treatments (μmol L ⁻¹)					
		0	5	10	15	25	
NaCl concentration (0 mmol L ⁻¹)	Hypocotyl	88.01 ± 6.26 ab	87.26 ± 15.33 ab	95.20 ± 7.34 a	88.45 ± 4.95 ab	74.09 ± 9.32 b	87.33 ± 8.10 ab
	Diameter ^b	0.95 ± 0.18 a	1.03 ± 0.13 a	1.04 ± 0.11 a	1.05 ± 0.13 a	0.89 ± 0.18 a	0.84 ± 0.17 a
	Length	69.61 ± 6.45 a	57.22 ± 10.16 a	52.89 ± 17.16 a	55.56 ± 13.37 a	54.74 ± 5.73 a	50.17 ± 10.13 a
	Diameter	0.55 ± 0.04 bc	0.59 ± 0.09 abc	0.70 ± 0.09 a	0.67 ± 0.06 ab	0.49 ± 0.05 cd	0.37 ± 0.10 d
	FW ^c	27.67 ± 1.88 a	24.37 ± 13.44 a	27.17 ± 6.41 a	28.13 ± 9.27 a	21.63 ± 0.51 a	20.53 ± 5.51 a
	DW ^c	1.87 ± 0.25 c	2.53 ± 0.50 abc	2.60 ± 0.53 ab	2.80 ± 0.20 a	2.00 ± 0.36 bc	2.67 ± 0.15 ab
	FW	28.03 ± 7.08 ab	28.40 ± 6.67 ab	22.73 ± 10.90 b	35.90 ± 5.03 a	37.57 ± 4.14 a	21.83 ± 2.35 b
	DW	3.13 ± 0.96 bc	3.47 ± 0.61 abc	4.47 ± 0.99 ab	4.77 ± 0.87 a	4.90 ± 1.00 a	2.87 ± 0.32 c
	FW	35.33 ± 7.54 a	34.00 ± 6.41 a	36.07 ± 10.51 a	45.80 ± 22.06 a	25.07 ± 1.26 a	25.40 ± 3.35 a
	DW	2.47 ± 0.25 abc	2.67 ± 0.25 ab	2.13 ± 0.15 bc	2.80 ± 0.20 a	2.00 ± 0.61 c	2.00 ± 0.26 c
NaCl concentration (100 mmol L ⁻¹)	FW	143.93 ± 32.43 a	142.50 ± 32.03 a	150.50 ± 44.46 a	152.47 ± 26.36 a	134.90 ± 13.64 a	125.90 ± 4.47 a
	DW	11.47 ± 0.32 bc	15.07 ± 1.86 a	15.80 ± 0.10 a	13.27 ± 1.97 ab	13.87 ± 1.29 ab	10.33 ± 2.06 c
	Hypocotyl	88.01 ± 6.26 ab	87.26 ± 15.33 ab	95.20 ± 7.34 a	88.45 ± 4.95 ab	74.09 ± 9.32 b	87.33 ± 8.10 ab
	Diameter ^b	0.95 ± 0.18 a	1.03 ± 0.13 a	1.04 ± 0.11 a	1.05 ± 0.13 a	0.89 ± 0.18 a	0.84 ± 0.17 a
	Length	69.61 ± 6.45 a	57.22 ± 10.16 a	52.89 ± 17.16 a	55.56 ± 13.37 a	54.74 ± 5.73 a	50.17 ± 10.13 a
	Diameter	0.55 ± 0.04 bc	0.59 ± 0.09 abc	0.70 ± 0.09 a	0.67 ± 0.06 ab	0.49 ± 0.05 cd	0.37 ± 0.10 d
	FW ^c	27.67 ± 1.88 a	24.37 ± 13.44 a	27.17 ± 6.41 a	28.13 ± 9.27 a	21.63 ± 0.51 a	20.53 ± 5.51 a
	DW ^c	1.87 ± 0.25 c	2.53 ± 0.50 abc	2.60 ± 0.53 ab	2.80 ± 0.20 a	2.00 ± 0.36 bc	2.67 ± 0.15 ab
	FW	28.03 ± 7.08 ab	28.40 ± 6.67 ab	22.73 ± 10.90 b	35.90 ± 5.03 a	37.57 ± 4.14 a	21.83 ± 2.35 b
	DW	3.13 ± 0.96 bc	3.47 ± 0.61 abc	4.47 ± 0.99 ab	4.77 ± 0.87 a	4.90 ± 1.00 a	2.87 ± 0.32 c
FW	35.33 ± 7.54 a	34.00 ± 6.41 a	36.07 ± 10.51 a	45.80 ± 22.06 a	25.07 ± 1.26 a	25.40 ± 3.35 a	
DW	2.47 ± 0.25 abc	2.67 ± 0.25 ab	2.13 ± 0.15 bc	2.80 ± 0.20 a	2.00 ± 0.61 c	2.00 ± 0.26 c	
FW	143.93 ± 32.43 a	142.50 ± 32.03 a	150.50 ± 44.46 a	152.47 ± 26.36 a	134.90 ± 13.64 a	125.90 ± 4.47 a	
DW	11.47 ± 0.32 bc	15.07 ± 1.86 a	15.80 ± 0.10 a	13.27 ± 1.97 ab	13.87 ± 1.29 ab	10.33 ± 2.06 c	

^a Values represent mean±SD determined from three replicates. Means followed by the same letter within a row are not significantly different at $P=0.05$ according to Duncan's multiple-range test. The maximum calculated values that were significantly high are in **bold** font; ^b Unit of length and diameter is millimeter (mm); ^c FW and DW stand for fresh weight and dry weight, respectively, in milligram (mg); ^d TLP stands for true leaves part; ^e TS stands for total seedlings.

Table 2. Effects of zinc sprays on the seedling biomass under NaCl concentration of 200 and 300 mmol L⁻¹ for 7 days.^a

Plant part	Parameter	Zinc treatments ($\mu\text{mol L}^{-1}$)						
		0	5	10	15	20	25	
NaCl concentration (200 mmol L ⁻¹)	Hypocotyl	Length ^b	63.71 ± 20.09 a	71.09 ± 13.81 a	65.35 ± 6.15 a	74.49 ± 13.03 a	63.66 ± 5.27 a	69.40 ± 8.34 a
		Diameter ^b	0.75 ± 0.05 a	0.72 ± 0.11 a	0.71 ± 0.07 a	0.81 ± 0.06 a	0.71 ± 0.12 a	0.73 ± 0.26 a
	Epicotyl	Length	50.10 ± 6.01 ab	50.84 ± 9.26 ab	44.18 ± 6.22 b	63.88 ± 11.87 a	52.05 ± 7.68 ab	57.65 ± 4.94 ab
		Diameter	0.50 ± 0.11 a	0.50 ± 0.11 a	0.53 ± 0.12 a	0.62 ± 0.21 a	0.45 ± 0.07 a	0.42 ± 0.09 a
	Cotyledon	FW ^c	18.40 ± 7.79 a	19.80 ± 6.67 a	27.20 ± 12.63 a	32.87 ± 4.58 a	24.47 ± 3.75 a	25.50 ± 7.02 a
		DW ^c	2.20 ± 0.10 a	1.97 ± 0.15 a	2.33 ± 0.49 a	2.50 ± 0.46 a	2.43 ± 0.45 a	2.40 ± 0.17 a
	TLP ^c	FW	31.03 ± 16.35 ab	38.03 ± 17.64 ab	47.83 ± 8.21 ab	52.03 ± 2.25 a	38.43 ± 2.85 ab	28.90 ± 10.91 b
		DW	2.70 ± 0.53 d	5.60 ± 1.82 abc	7.13 ± 1.65 ab	6.60 ± 0.20 a	4.40 ± 0.78 cd	4.53 ± 0.93 bcd
	Root	FW	331.50 ± 9.00 b	34.77 ± 9.99 ab	38.07 ± 18.47 ab	56.13 ± 6.57 ab	37.13 ± 6.57 ab	33.57 ± 8.76 b
		DW	2.97 ± 0.65 ab	2.10 ± 0.10 b	3.27 ± 0.64 a	2.30 ± 0.10 b	2.43 ± 0.47 ab	2.87 ± 0.42 ab
NaCl concentration (300 mmol L ⁻¹)	TS ^c	FW	138.60 ± 48.67 b	165.60 ± 24.25 ab	189.87 ± 69.10 ab	223.43 ± 39.35 a	163.80 ± 15.52 ab	148.97 ± 38.16 ab
		DW	12.63 ± 1.21 b	16.37 ± 3.01 ab	16.10 ± 3.35 ab	17.10 ± 0.10 a	13.03 ± 1.07 ab	14.23 ± 2.06 ab
	Hypocotyl	Length ^b	90.28 ± 14.04 a	74.50 ± 17.78 a	75.60 ± 6.97 a	76.13 ± 9.02 a	85.71 ± 28.14 a	61.68 ± 1.35 a
		Diameter ^b	0.73 ± 0.21 ab	0.54 ± 0.12 b	0.78 ± 0.00 ab	0.91 ± 0.22 a	0.71 ± 0.17 ab	0.61 ± 0.15 ab
	Epicotyl	Length	51.96 ± 15.52 a	53.12 ± 13.17 a	55.30 ± 7.84 a	56.22 ± 14.72 a	40.45 ± 11.09 a	56.79 ± 8.70 a
		Diameter	0.51 ± 0.10 a	0.51 ± 0.08 a	0.60 ± 0.18 a	0.61 ± 0.08 a	0.43 ± 0.08 ab	0.28 ± 0.10 b
	Cotyledon	FW ^c	20.70 ± 8.97 a	24.73 ± 8.46 a	17.33 ± 9.21 a	19.87 ± 4.50 a	9.63 ± 3.48 a	8.90 ± 9.01 a
		DW ^c	2.70 ± 0.72 a	2.33 ± 0.15 abc	1.80 ± 0.26 bc	1.67 ± 0.12 c	1.83 ± 0.15 bc	2.47 ± 0.31 ab
	TLP ^c	FW	9.53 ± 0.81 a	13.70 ± 7.82 a	17.07 ± 7.26 a	20.67 ± 5.37 a	12.97 ± 9.23 a	8.97 ± 8.94 a
		DW	1.00 ± 0.10 d	1.53 ± 0.38 cd	11.63 ± 2.21 a	4.20 ± 0.95 b	3.03 ± 0.15 bcd	3.70 ± 0.85 b
Root	FW	27.40 ± 1.65 c	30.53 ± 8.10 bc	38.87 ± 9.11 abc	45.53 ± 1.22 ab	51.90 ± 13.99 a	51.97 ± 6.79 a	
	DW	1.10 ± 0.20 a	1.20 ± 0.36 a	2.13 ± 0.15 a	1.77 ± 0.25 a	2.40 ± 0.36 a	3.33 ± 0.91 a	
TS ^c	FW	136.23 ± 12.79 ab	150.23 ± 45.25 ab	172.60 ± 29.41 a	132.70 ± 15.26 ab	133.63 ± 29.34 ab	114.80 ± 5.47 b	
	DW	7.87 ± 0.70 d	16.50 ± 3.40 b	20.99 ± 2.63 a	13.67 ± 1.21 bc	11.93 ± 1.21 cd	15.80 ± 3.47 c	

^a Values represent mean±SD determined from three replicates. Means followed by the same letter within a row are not significantly different at $P=0.05$ according to Duncan's multiple-range test. The maximum calculated values that were significantly high are in **bold** font; ^b Unit of length and diameter is millimeter (mm); ^c FW and DW stand for fresh weight and dry weight, respectively, in milligram (mg); ^d TLP stands for true leaves part, ^e TS stands for total seedlings.

concentration in the range of 10 to 15 $\mu\text{mol L}^{-1}$ at all NaCl treatments ($P < 0.05$). Significant impacts on dry weight (DW) values of root and total seedlings (TS) were also found. Dry weight values of root ($2.80 \text{ mg} \pm 0.20$) and cotyledon ($2.80 \text{ mg} \pm 0.20$) under 15 $\mu\text{mol L}^{-1}$ zinc concentration were significantly ($P < 0.05$) higher than other zinc treatments when seedlings grew in the absence of salt (0 mmol L^{-1}), while DW value of TS ($15.80 \text{ mg} \pm 0.10$) was significantly ($P < 0.05$) affected by zinc concentration of 10 $\mu\text{mol L}^{-1}$. Under 100 mmol L^{-1} NaCl concentration treatment, there was no difference in DW value of root in all zinc treatments. However, the DW value of TS ($19.73 \text{ mg} \pm 1.52$) showed significant differences ($P < 0.05$) in the 15 $\mu\text{mol L}^{-1}$ zinc treatment (Table 1). In addition, both DW value of root ($3.27 \text{ mg} \pm 0.64$) and TS ($17.10 \text{ mg} \pm 0.10$) at NaCl concentration of 200 mmol L^{-1} were significantly influenced by zinc concentration of 15 $\mu\text{mol L}^{-1}$, whereas both fresh weight (FW) value and DW value of cotyledon were not significantly ($P > 0.05$) affected by different zinc treatments (Table 2). Under NaCl concentration of 300 mmol L^{-1} treatment, DW value of TS ($20.99 \text{ mg} \pm 2.63$) was significantly high under zinc concentration of 10 $\mu\text{mol L}^{-1}$ (Table 2). Moreover, zinc concentration of 15 $\mu\text{mol L}^{-1}$ also significantly ($P < 0.05$) affected fresh weight (FW) and DW value of true leaves part (TLP) at 0, 100, and 200 mmol L^{-1} NaCl treatments (Tables 1-2). (Table 2) shows that 10 $\mu\text{mol L}^{-1}$ zinc concentration significantly ($P < 0.05$) affected DW value ($11.63 \text{ mg} \pm 2.21$) of TLP at 300 mmol L^{-1} NaCl stress.

Maximal Efficiency of PSII Photochemistry (F_v/F_m) Characteristics

Under all salt treatments, a single peak was observed for F_v/F_m ratio at different zinc concentrations (Figure 3; Table 3). The peak was found nearby the 15 $\mu\text{mol L}^{-1}$ zinc concentration and the same trend was clearly

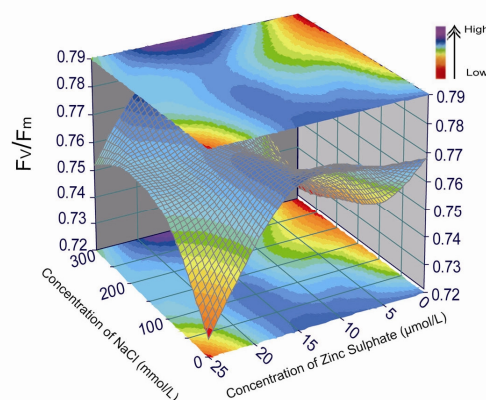


Figure 3. Three dimensional image of F_v/F_m response surface at different zinc and salt stress.

observed in 3D plot (Figure 3). High F_v/F_m ratio (purple) through the 15 $\mu\text{mol L}^{-1}$ area under all different NaCl concentrations and low F_v/F_m ratio (chrysoidine and yellow) was found at both ends of the zinc concentrations (0 and 25 $\mu\text{mol L}^{-1}$).

Other Chlorophyll Fluorescence Measurements

In absence of salt (0 mmol L^{-1} NaCl), the non-photochemical quenching (NPQ) for 5 $\mu\text{mol L}^{-1}$ zinc concentration was higher than all other treatments after the PAR increased to $600 \mu\text{mol m}^{-2} \text{ s}^{-1}$. However, in treatments less than 100 mmol L^{-1} NaCl concentrations, the NPQ for 15 $\mu\text{mol L}^{-1}$ zinc concentrations established the highest position as the PAR increased above $1,200 \mu\text{mol m}^{-2} \text{ s}^{-1}$. In addition, we observed the highest NPQ value under 5 $\mu\text{mol L}^{-1}$ zinc treatment in 200 mmol L^{-1} salt stress, and 20 $\mu\text{mol L}^{-1}$ zinc treatment had the highest NPQ values under 300 mmol L^{-1} salt stress. The coefficient of non-photochemical quenching (q_N) followed a similar pattern to NPQ (Figure 4).

Overall, there was a steady decline in the coefficient of photochemical quenching (q_P) with increasing irradiance (Figure 5). In

Table 3. Effects of zinc supplement (sprays) on the *Fv/Fm* ratio in the leaves of *Glycine soja* under different NaCl stress treatments.^a

Treatments	Fv/Fm ratio					
	Zinc concentration ($\mu\text{mol L}^{-1}$)					
	0	5	10	15	20	25
0 mmol L ⁻¹ NaCl	0.767 ± 0.008 a	0.768 ± 0.008 a	0.771 ± 0.005 a	0.771 ± 0.005 a	0.753 ± 0.025 ab	0.726 ± 0.030 a
100 mmol L ⁻¹ NaCl	0.768 ± 0.023 a	0.758 ± 0.018 a	0.760 ± 0.006 a	0.769 ± 0.010 a	0.769 ± 0.010 a	0.754 ± 0.015 a
200 mmol L ⁻¹ NaCl	0.731 ± 0.001 d	0.745 ± 0.001 c	0.756 ± 0.007 b	0.766 ± 0.004 a	0.761 ± 0.005 ab	0.760 ± 0.006 ab
300 mmol L ⁻¹ NaCl	0.725 ± 0.004 c	0.754 ± 0.011 b	0.775 ± 0.003 ab	0.782 ± 0.009 a	0.769 ± 0.016 ab	0.752 ± 0.025 b

^a Values represent means±SD determined from three replicates. Means followed by the same letter within a row are not significantly different at $P=0.05$ according to Duncan's multiple-range test. The maximum calculated values are in **bold** font.

absence of salt, the highest *qP* value was found for 10 $\mu\text{mol L}^{-1}$ zinc concentration. The zinc concentration required for the highest *qP* value increased with increasing salt stress. At 100 mmol L⁻¹ salt concentration, the *qP* value was the highest at 15 $\mu\text{mol L}^{-1}$ zinc treatment, whilst at 200 mmol L⁻¹ salt level, the *qP* value was the highest at 20 $\mu\text{mol L}^{-1}$ zinc concentration. However, zinc concentration required for the highest *qP* value decreased to 5 $\mu\text{mol L}^{-1}$ when exposed to 300 mmol L⁻¹ of salinity level (Figure 5). The changes of effective quantum yield of photochemical energy conversion in photosystem II (Φ_{PSII} or Yield, Y) was also determined for different zinc and salt concentrations. The Yield (Y) followed the same trend as *qP* (Figure 6).

The apparent electron transport rate (ETR) of photosystem II increased rapidly as the PAR increased for all different zinc and salt concentrations (Figure 7). Moreover, all ETR values peaked at the PAR of 600 $\mu\text{mol m}^{-2}\text{s}^{-1}$. The highest ETR values (42.03, 41.50, 37.53, 33.83) were found under 20, 15, 20, and 5 $\mu\text{mol L}^{-1}$ of zinc in 0, 100, 200, and 300 mmol L⁻¹ NaCl concentration salt treatments, respectively.

Characteristics of Chlorophyll Fluorescence Value under Different PAR

A maximum value of chlorophyll fluorescence parameter under all treatments could be identified under specific PAR. The highest *NPQ* and *qN* values were at the PAR of 1600 $\mu\text{mol m}^{-2}\text{s}^{-1}$, the highest ETR value was at 600 $\mu\text{mol m}^{-2}\text{s}^{-1}$, and the highest Yield value was at 0 $\mu\text{mol m}^{-2}\text{s}^{-1}$. In addition, the Pearson correlations between chlorophyll fluorescence parameters at these optimum PAR values and different salt treatments are shown in Table 4. There was a significant negative correlation (-0.594, $P < 0.01$) between *Fv/Fm* ratio and zinc concentration in the absence of salt, whereas the *Fv/Fm* ratio was positively correlated (0.805, $P < 0.01$) to increasing zinc concentration at 200 mmol L⁻¹ salt stress.

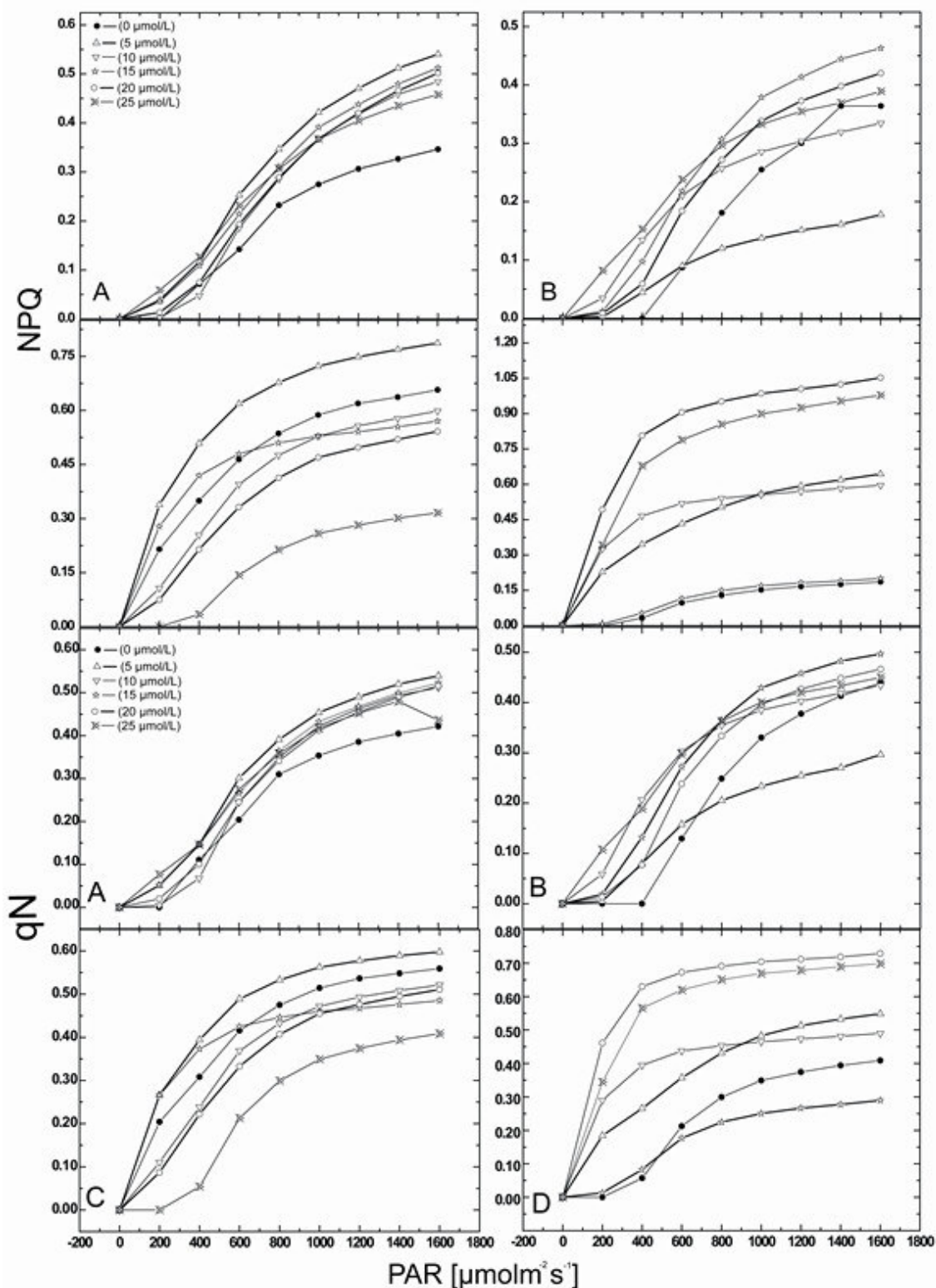


Figure 4. Changes in non-photochemical quenching (NPQ) and its coefficient (qN) in response to light irradiance at different levels of zinc treatments under the salt concentrations of 0 (A); 100 (B); (C), and 300 mmol L⁻¹ (D).

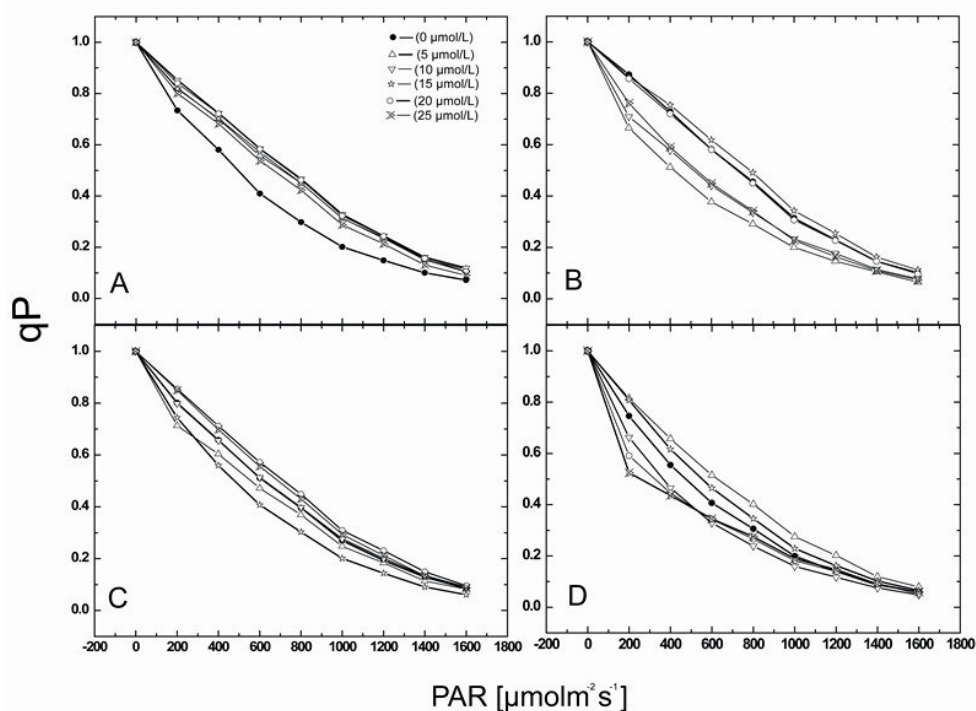


Figure 5. Changes in coefficient of photochemical quenching (qP) in response to light irradiance at different levels of zinc treatments under the salt concentrations of 0 (A); 100 (B); 200 (C), and 300 mmol L^{-1} (D).

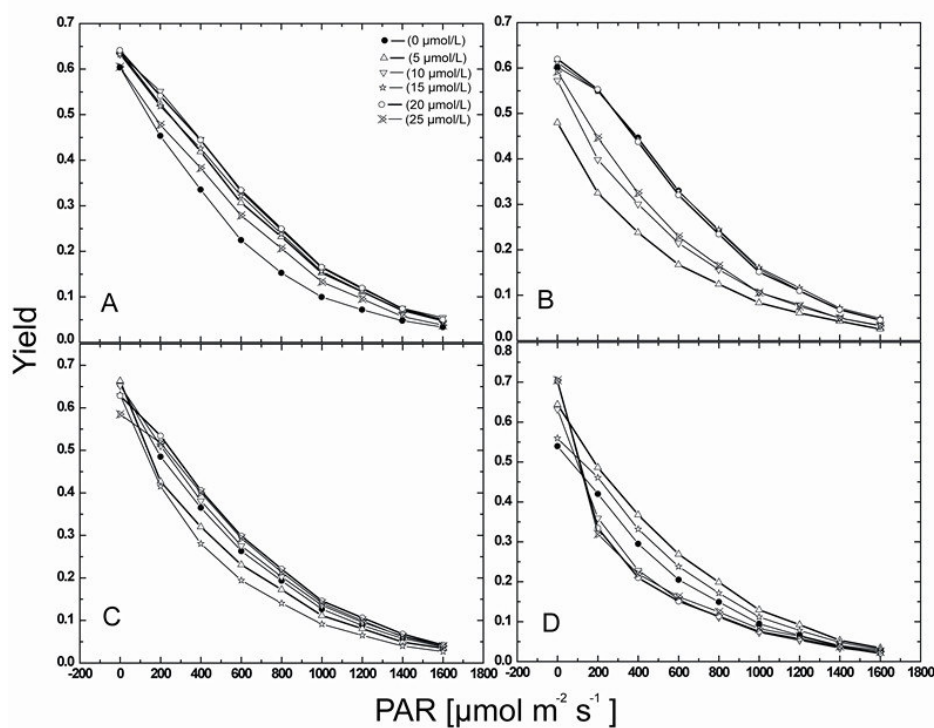


Figure 6. Changes in photochemical energy conversion (yield) in response to light irradiance at different levels of zinc treatments under the salt concentrations of 0 (A); 100 (B); 200 (C), and 300 mmol L^{-1} (D).

Table 4. The Pearson correlations between chlorophyll fluorescence parameters at optimum PAR values and different salt treatments.

Treatments	<i>Fv/Fm</i> ratio	<i>NPQ</i>	<i>qP</i>	<i>ETR</i>	Yield
		PAR of 1600 $\mu\text{mol m}^{-2}\text{s}^{-1}$	PAR of 200 $\mu\text{mol m}^{-2}\text{s}^{-1}$	PAR of 600 $\mu\text{mol m}^{-2}\text{s}^{-1}$	PAR of 0 $\mu\text{mol m}^{-2}\text{s}^{-1}$
0 mmol L ⁻¹ NaCl	-0.594 **	0.171	0.334	0.321	0.027
100 mmol L ⁻¹ NaCl	0.393	0.330	0.082	0.045	0.335
200 mmol L ⁻¹ NaCl	0.805 **	-0.347	0.312	0.230	-0.363
300 mmol L ⁻¹ NaCl	0.419	0.562 *	-0.577 *	-0.382	0.565 *

* $P < 0.05$, ** $P < 0.01$.

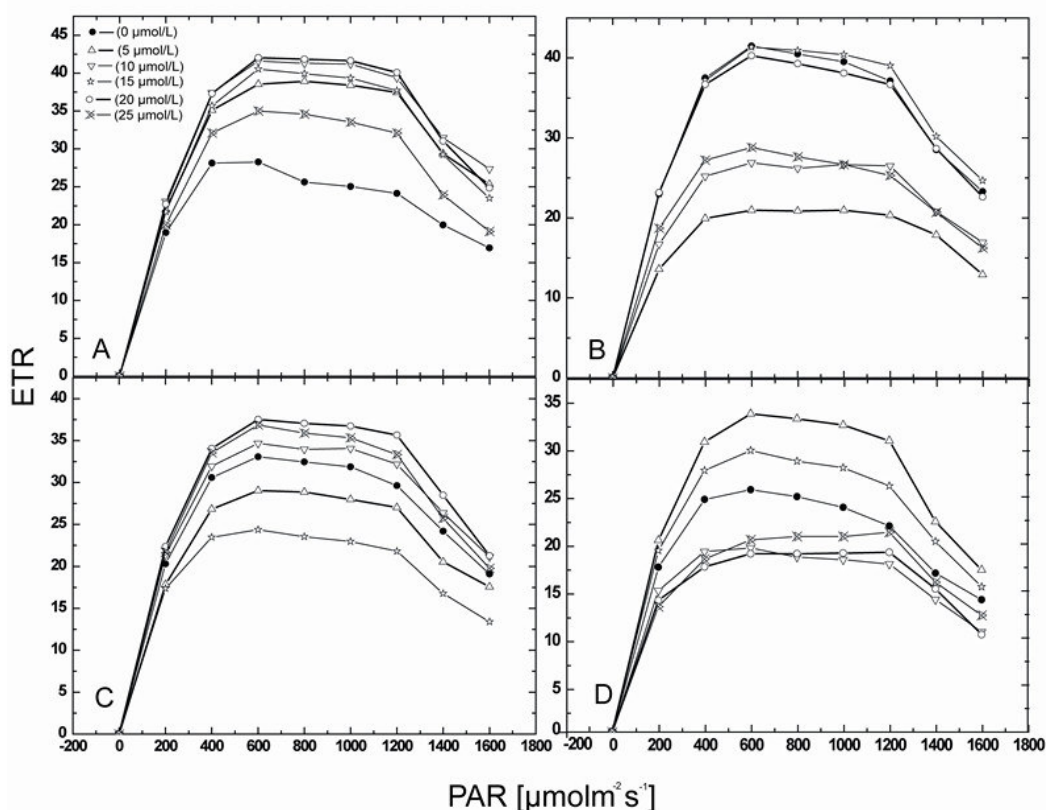


Figure 7. Changes in apparent electron transport rate (ETR) in response to light irradiance at different levels of zinc treatments under the salt concentrations of 0 (A); 100 (B); 200 (C), and 300 mmol L⁻¹ (D).

Further, *NPQ* and Yield were significantly positively correlated (0.562 and 0.565, $P < 0.05$) with zinc concentration at 300 mmol L⁻¹ salt stress, whilst the *qP* was negatively correlated (-0.577, $P < 0.05$) with zinc concentration at 300 mmol L⁻¹ of salt stress.

The comparison of all chlorophyll fluorescence parameters at optimum PAR and different zinc and salt concentration is shown in Table 5. Although there was no significant difference ($P > 0.05$) in *NPQ* and

qN values at different zinc concentrations under salt treatments of 0 and 200 mmol L⁻¹, a significant increase ($P < 0.05$) was observed for 15 $\mu\text{mol L}^{-1}$ zinc at 100 mmol L⁻¹ salt stress and 20 $\mu\text{mol L}^{-1}$ zinc at 300 mmol L⁻¹ salt stress. Similar results were obtained for Yield value at PAR of 0 $\mu\text{mol}^{-2}\text{s}^{-1}$. The Yield value of 5 $\mu\text{mol L}^{-1}$ zinc concentration was significantly ($P < 0.05$) lower than the other zinc treatments under 100 mmol L⁻¹ of salt stress. However, the

Table 5. Comparison of chlorophyll fluorescence parameters at optimum PAR and different concentrations of zinc and salt.^a

Zinc concentration ($\mu\text{mol/L}$)	Chlorophyll fluorescence parameters			
	Concentration of NaCl (mmol L^{-1})			
<i>(1) NPQ values at PAR of 1600 $\mu\text{mol m}^{-2}\text{s}^{-1}$</i>				
	0	100	200	300
0	0.3463 a	0.3640 ab	0.6570 a	0.1863 b
5	0.5403 a	0.1780 b	0.7880 a	0.6433 ab
10	0.4840 a	0.3343 ab	0.5983 a	0.5960 ab
15	0.5123 a	0.4630 a	0.5703 a	0.2007 b
20	0.5017 a	0.4203 ab	0.5417 a	1.0527 a
25	0.4577 a	0.3890 ab	0.3160 a	0.9783 a
<i>(2) qN values at PAR of 1600 $\mu\text{mol m}^{-2}\text{s}^{-1}$</i>				
	0	100	200	300
0	0.4217 a	0.4387 a	0.5590 a	0.2827 b
5	0.5393 a	0.2967 a	0.5977 a	0.5487 a
10	0.5120 a	0.4340 a	0.5213 a	0.4900 ab
15	0.5220 a	0.4967 a	0.4853 a	0.2900 b
20	0.5157 a	0.4663 a	0.5107 a	0.7287 a
25	0.4990 a	0.4500 a	0.4087 a	0.6983 a
<i>(3) qP values at PAR of 200 $\mu\text{mol m}^{-2}\text{s}^{-1}$</i>				
	0	100	200	300
0	0.7333 b	0.8730 a	0.8013 a	0.7463 ab
5	0.8213 a	0.6653 b	0.7133 a	0.8147 a
10	0.8520 a	0.7090 b	0.7987 a	0.6627 abc
15	0.8177 a	0.8637 a	0.7443 a	0.8083 a
20	0.8403 a	0.8563 a	0.8550 a	0.5910 bc
25	0.8007 ab	0.7613 ab	0.8493 a	0.5240 c
<i>(4) ETR values at PAR of 600 $\mu\text{mol m}^{-2}\text{s}^{-1}$</i>				
	0	100	200	300
0	28.2667 b	41.5000 a	33.0667 a	25.7667 ab
5	38.5333 a	20.9667 b	29.0333 a	33.8333 a
10	41.6000 a	26.9000 b	34.6667 a	19.5667 b
15	40.5333 a	41.3667 a	24.3667 a	29.9333 ab
20	42.0333 a	40.3000 a	37.5333 a	18.9667 b
25	35.0000 ab	28.8000 ab	36.8667 a	20.4333 b
<i>(5) Yield values at PAR of 0 $\mu\text{mol m}^{-2}\text{s}^{-1}$</i>				
	0	100	200	300
0	0.6037 a	0.6023 a	0.6543 a	0.5397 b
5	0.6373 a	0.4797 b	0.6633 a	0.6440 ab
10	0.6330 a	0.5727 a	0.6527 a	0.6320 ab
15	0.6323 a	0.6123 a	0.6307 a	0.5593 b
20	0.6413 a	0.6200 a	0.6287 a	0.7027 a
25	0.6050 a	0.5920 a	0.5850 a	0.7050 a

^a Values represent the mean of three variables. Values that are followed by different letters are significantly ($P < 0.05$) different from each other (Duncan's multiple-range test). Separate analysis was done for each column. The maximum calculated value is in **bold** font.

Yield value under zinc concentration of 25 $\mu\text{mol L}^{-1}$ was significantly ($P < 0.05$) higher at 300 mmol L^{-1} of salt. Also, both qP and

ETR values were not significantly ($P > 0.05$) affected at 200 mmol L^{-1} salt stress.



DISCUSSION

Salinity stress causes retardation in plant growth by decreasing length, fresh weight, and dry matter (Mantri *et al.*, 2012). Water deficit caused due to high salt concentration results in abnormal change in plant morphology (Ghassemi-Golezani *et al.*, 2009). In our experiments, 15 $\mu\text{mol L}^{-1}$ zinc concentration was found to be the most ideal for use as a foliar spray to alleviate salinity stress in *G. soja*. For most plants, zinc is an essential component of enzymes and takes part in the synthesis of chlorophyll and other proteins (Vallee and Auld, 1990). In this study, significant differences were observed in the leaf morphology and biomass variables of seedlings sprayed with 10–15 $\mu\text{mol L}^{-1}$ zinc under salt stress (Figure 2; Table 1). A significant increase in biomass, with respect to length or width of stem, true leaves part (TLP) and dry weight (DW) of total seedlings (TS) was observed by spraying the plants with 10–15 $\mu\text{mol L}^{-1}$ zinc, thus indicating that proper concentration of zinc is required for dry matter accumulation and seedling growth. The positive effects of appropriate zinc concentration (10–20 $\mu\text{mol L}^{-1}$) on DW of seedling and especially newly-developed TLP under NaCl stress could be explained by the decrease in uptake of excess Na^+ and Cl^- . Also, zinc may help nutrient translocations from the aged cells to newborn cells (Rockenfeller and Madeo, 2008). Zinc may, therefore, play an important role in membrane permeability, phospholipids (P) accumulation, and scavenging free oxygen radicals. These results correlate well with the findings of Alpaslan *et al.* (1999) and Qu *et al.* (2009) who demonstrated that, in the salt affected areas, zinc application could alleviate possible Na^+ and Cl^- injury in plants.

Chlorophyll fluorescence is an important tool that has been extensively used to monitor plant health (Belkhodja *et al.*, 1994; Jimenez *et al.*, 1997; Sayed, 2003; Dai *et al.*, 2009). The *Fv/Fm* ratio is frequently used as

a stress indicator and depicts the potential yield of the photochemical reaction (Bjorkman and Demmig, 1987). Our experiment showed that the *Fv/Fm* ratio of *G. soja* was significantly affected by NaCl concentration of 200, 300 mmol L^{-1} . We also found that 15 $\mu\text{mol L}^{-1}$ zinc spray was optimum for restoring the *Fv/Fm* ratio ($P < 0.05$). High *Fv/Fm* ratio represents maximal efficiency of PSII photochemistry (Lundmark *et al.*, 1998). Therefore, we conclude that salinity may reduce photosynthesis by indirect effect on the photosynthetic apparatus and the zinc concentration of 15 $\mu\text{mol L}^{-1}$ could be optimal for stress relief.

The chlorophyll coefficients NPQ and qP represent the energy that cannot be utilized to transport photosynthetic electrons down to being dissipated harmlessly as heat energy from PSII antennae (Veres *et al.*, 2006). Low NPQ value indicates that the plants efficiently utilized the energy absorbed by antenna pigments and effectively reduce the irradiance heat in PSII (Guo *et al.*, 2006). The *ETR* value represents the relative quantity of electrons passing through PSII in steady-state photosynthesis (Tezara *et al.*, 2003). Thus, *ETR* value reduction means loss of capture efficiency/excitation of chlorophyll. The *qP* value is an indication of the proportion of PSII reaction centers that are open; high qP is beneficial to electron transport and PSII yield, a decrease in qP demonstrates an increase in susceptibility to metal inhibition (Mao *et al.*, 2007).

Further, under salt stress, the plants sprayed with 0 and 5 $\mu\text{mol L}^{-1}$ zinc could not sustain growth. Zinc sprays of 10 to 20 $\mu\text{mol L}^{-1}$ could restore the chlorophyll fluorescence parameter values under different salt concentration. These results are consistent with previous study where supplementation of irrigation water with low concentration of zinc was found to have protective effects against the salinity stress, whilst, high levels of zinc (25 $\mu\text{mol L}^{-1}$) combined with NaCl stress caused metal-induced inhibition of chlorophyll biosynthesis (Vallee and Auld,

1990; Qu *et al.* 2009). Excessive zinc also disrupts the chloroplast envelope and thylakoid membrane system, thus affecting photosynthesis of the plant (Jin *et al.*, 2006).

In natural environment, plants are always subjected to a multitude of complex stresses among which salinity stress is frequently encountered (Mantri *et al.*, 2010; Rang *et al.*, 2011). Alleviating the salinity stress at the seedling stage is an ideal adaptation strategy for plants. Improvement of salt tolerance by addition of nutrients has been reported in rice (Matoh *et al.*, 1986), wheat (Ahmad *et al.*, 1992), tomato (Al-Aghabary *et al.*, 2005), and cucumber (Zhu *et al.*, 2004). Zinc is a vital component of some enzymes and is an important microelement required by plants. Our results reveal that salt toxicity in *Glycine soja* seedlings can be alleviated by foliar spray of zinc. The results are consistent with the previous findings of Qu *et al.* (2009), who added zinc to hydroponic media to alleviate salinity stress in soybean.

In conclusion, salinity stress significantly affected the growth of *Glycine soja* seedlings. However, foliar zinc sprays of 10–20 $\mu\text{mol L}^{-1}$ concentration were found to alleviate the stress. Zinc supplementation at 15 $\mu\text{mol L}^{-1}$ improved the biomass of the seedlings as revealed by the measurement of true leaves part (TLP), and dry weight (DW) of total seedlings (TS). The results correlate well with the plant chlorophyll fluorescence parameters. Thus, 15 $\mu\text{mol L}^{-1}$ zinc concentration is considered to be the optimum zinc concentration for use as foliar sprays for alleviating salinity stress of *G. soja*. Our results will help to obtain better yields of *G. soja* under salt stress. Future study will determine if these results can be extended to its close relative, soybean.

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REFERENCES

1. Keyser, H. H. and Li, F. 1992. Potential for Increasing Biological Nitrogen Fixation in Soybean. *Plant Soil.*, **141**: 119–135.
2. Concibido, V. C., Vallee, B. L., Mcclaird, P., Pineda, N., Meyer, J., Hummel, L., Yang, J., Wu, K. and Delannay, X. 2003. Introgression of a Quantitative Trait Locus for Yield from *Glycine soja* into Commercial Soybean Cultivars. *Tag Theoretical Applied Genetics*, **106**: 575–582.
3. Liu, Y. L., Li, Y. H., Zhou, G. A., Uzokwe, N., Chang, R. Z., Chen, S. Y. and Qiu, L. J. 2010. Development of Soybean EST-SSR Markers and Their Use to Assess Genetic Diversity in the Subgenus *Soja*. *Agr. Sci. China.*, **9**: 1423–1429.
4. Wang, K. J. and Takahata, Y. 2007. A Preliminary Comparative Evaluation of Genetic Diversity between Chinese and Japanese Wild Soybean (*Glycine soja*) Germplasm Pools Using SSR Markers. *Genet. Resour. Crop Ev.*, **54**: 157–165.
5. Kiang, Y. T., Chiang, Y. C. and Kaizuma, N. 1992. Genetic Diversity in Natural Populations of Wild Soybean in Iwate Prefecture, Japan. *J. Hered.*, **83**: 325–329.
6. Fujita, R., Ohara, M., Okazaki, O. and Shimamoto, Y. 1997. The Extent of Natural Cross-pollination in Wild Soybean (*Glycine soja*). *J. Hered.*, **88**: 124–128.
7. Wang, K. J., Li, X. H. and Li, F. S. 2008. Phenotypic Diversity of the Big Seed Type Sub-collection of Wild Soybean (*Glycine soja* Sieb. et Zucc.) in China. *Genet. Resour. Crop Ev.*, **55**: 1335–1346.
8. Mantri, N., Patade, V., Penna, S., Ford, R. and Pang, E. C. K. 2012. Abiotic Stress Responses in Plants: Present and Future. In: “*Abiotic Stress Responses in Plants: Metabolism to Productivity*”, (Eds.): Ahmad, P. and Prasad M. N. V. Springer, New York, PP. 1-19.
9. Munns, R. 2002. Comparative Physiology of Salt and Water Stress. *Plant Cell. Environ.*, **25**: 239–250.
10. Geilfus, C. M., Zorb, C. and Muhling, K. H. 2010. Salt Stress Differentially Affects



- Growth-Mediating β -expansins in Resistant and Sensitive Maize (*Zea mays* L.). *Plant Physiol. Biochem.*, **48**: 993–998.
11. Ashraf, M. and Harris, P. J. C. 2004 Potential Biochemical Indicators of Salinity Tolerance in Plants. *Plant Sci.*, **166**: 3–16.
 12. Du, C. X., Fan, H. F., Guo, S. R., Tekafumi, T. and Li, J. 2010. Proteomic Analysis of Cucumber Seedling Roots Subjected to Salt Stress. *Phytochem.*, **71**: 1450–1459.
 13. Tavakkoli, E., Rengasamy, P. and McDonald, G. K. 2010. High Concentrations of Na^+ and Cl^- Ions in Soil Solution Have Simultaneous Detrimental Effects on Growth of Faba Bean under Salinity Stress. *J. Exp. Bot.*, **1**: 4449–4459.
 14. Alpaslan, M., Inal, A., Gunes, A., Cikili, Y. and Ozcan, H. 1999. Effect of Zinc Treatment on the Alleviation of Sodium and Chloride Injury in Tomato (*Lycopersicon esculentum* (L.) Mill. cv. Lale) Grown under Salinity. *Tr. J. Bot.*, **23**: 1–6.
 15. Tavallali, V., Rahemi, M., Eshghi, S., Kholdebarin, B. and Ramezani, A. 2010. Zinc Alleviates Salt Stress and Increases Antioxidant Enzyme Activity in the Leaves of Pistachio (*Pistacia vera* L. 'Badami') Seedlings. *Turk. J. Agri. For.*, **34**: 349–359.
 16. Aregheore, E. M. 2012. Nutritive Value and Inherent Anti-nutritive Factors in Four Indigenous Edible Leafy Vegetables in Human Nutrition in Nigeria: A Review. *J. Food Res. Sci.*, **1**: 1–14.
 17. Cherif, J., Derbel, N., Nakkach, M., Bergmann, H. V., Jemal, F. and Lakhdar, Z. B. 2010. Analysis of *In vivo* Chlorophyll Fluorescence Spectra to Monitor Physiological State of Tomato Plants Growing under Zinc Stress. *J. Photoch. Photobiol. B.*, **101**: 332–339.
 18. Cakmak, I. 2000. Possible Roles of Zinc in Protecting Plant Cells from Damage by Reactive Oxygen Species. *New Phytol.*, **146**: 185–205.
 19. Zago, M. P. and Oteiza, P. I. 2001. The Antioxidant Properties of Zinc: Interactions with Iron and Antioxidants. *Free Radical Bio. Med.*, **31**: 266–274.
 20. Bonnet, M., Camares, O. and Veisseire, P. 2000. Effects of Zinc and Influence of *Acremonium lolii* on Growth Parameters, Chlorophyll Fluorescence and Antioxidant Enzyme Activities of Ryegrass (*Lolium perenne* L. cv Apollo). *J. Exp. Bot.*, **51**: 945–953.
 21. Rout, G. R. and Das, P. 2003. Effect of Metal Toxicity on Plant Growth and Metabolism: I. *Zinc Agron.*, **23**: 3–11.
 22. Lingua, G., Franchin, C., Todeschini, V., Castiglione, S., Biondi, S., Burlando, B., Parravicini, V., Torrigiani, P. and Berta, G. 2008. Arbuscular Mycorrhizal Fungi Differentially Affect the Response to High Zinc Concentrations of Two Registered Poplar Clones. *Environ. Pollut.*, **153**: 137–147.
 23. Andrade, S. A. L., Gratao, P. L., Schiavinato, M. A., Silveira, A. P. D., Azevedo, R. A. and Mazzafer, P. 2009. Zn Uptake, Physiological Response and Stress Attenuation in Mycorrhizal Jack Bean Growing in Soil with Increasing Zn Concentrations. *Chemosphere*, **75**: 1363–1370.
 24. Wang, C., Zhang, S. H., Wang, P. F., Hou, J., Zhang, W. J., Li, W. and Lin, Z. P. 2009. The Effect of Excess Zn on Mineral Nutrition and Antioxidative Response in Rapeseed Seedlings. *Chemosphere*, **75**: 1468–1476.
 25. Chaloub, R. M., Magalhaes, C. C. P. D. and Dos Santos, C. P. 2005. Early Toxic Effects of Zinc on PSII of *Synechocystis aquatilis* F. *aquatilis* (cyanophyceae). *J. Phycol.*, **41**: 1162–1168.
 26. Dhir, B., Sharmila, P. and Pardha Sarad, P. 2008. Photosynthetic Performance of *Salvinia natans* Exposed to Chromium and Zinc Rich Wastewater. *Braz. J. Plant Physiol.*, **20**: 61–70.
 27. Verma, T. S. and Neue, H. U. 1984. Effect of Soil Salinity Level and Zinc Application on Growth, Yield, and Nutrient Composition of Rice. *Plant Soil.*, **82**: 3–14.
 28. Tzortzakis, N. G. 2010. Potassium and Calcium Enrichment Alleviate Salinity-Induced Stress in Hydroponically Grown Endives. *J. Hortic. Sci.*, **37**: 155–162.
 29. Yildirim, E., Turan, M. and Guvenc, I. 2008. Effect of Foliar Salicylic Acid Applications on Growth, Chlorophyll, and Mineral Content of Cucumber Grown under Salt Stress. *J. Plant Nutr.*, **31**: 593–612.
 30. Qu, Y. N., Zhou, Q. and Yu, B. J. 2009. Effects of Zn^{2+} and Niflumic Acid on Photosynthesis in *Glycine soja* and *Glycine max* Seedlings under NaCl Stress. *Environ. Exp. Bot.*, **65**: 304–309.
 31. Zheng, H., Lu, H. F., Zheng, Y. P., Lou, H. Q. and Chen, C. Q. 2010. Automatic Sorting of Chinese Jujube (*Zizyphus jujuba* Mill. cv. 'hongxing') Using Chlorophyll Fluorescence

- and Support Vector Machine. *J. Food Eng.*, **101**: 402–408.
32. Jiang, L., Shen, Z., Zheng, H., He, W., Deng, G. and Lu, H. 2013. Noninvasive Evaluation of Fructose, Glucose and Sucrose Contents in Fig Fruits during Development Using Chlorophyll Fluorescence and Chemometrics. *J. Agr. Sci. Tech.*, **15**: 333–342.
 33. Genty, B., Briantais, J. M. and Baker, N. R. 1989. The Relationship between Quantum Yield of Photosynthetic Electron Transport and Quenching of Chlorophyll Fluorescence. *BBA-Gen. Subjects*, **990**: 87–92.
 34. Seemann, J. R. and Critchley, C. 1985. Effects of Salts Stress on the Growth, Ion Content, Stomatal Behaviour and Photosynthetic Capacity of a Salt-sensitive Species, *Phaseolus vulgaris* L. *Planta*, **164**: 151–162.
 35. Ghassemi-Golezani, K., Taifeh-Noori, M., Oustan, S. and Moghaddam, M. 2009. Response of Soybean Cultivars to Salinity Stress. *J. Food Agric. Environ.*, **7**: 401–404.
 36. Vallee, B. L. and Auld, D. S. 1990. Zinc Coordination, Function, and Structure of Zinc Enzymes and Other Proteins. *Bioch.*, **29**: 5647–5659.
 37. Rockenfeller, P. and Madeo, F. 2008. Apoptotic Death of Ageing Yeast. *Exp. Gerontol.*, **43**: 876–881.
 38. Belkhdja, R., Morales, F., Abadia, A., Gomez-aparisi, J. and Abadia, J. 1994. Chlorophyll Fluorescence as a Possible Tool for Salinity Tolerance Screening in Barler (*Hordeum vulgare* L.). *Plant Physiol.*, **104**, 667–673.
 39. Jimenez, M. S., Gonzalez-rodriguez, A. M., Morales, D., Cid, M. C., Socorro, A. R. and Caballero, M. 1997. Evaluation of Chlorophyll Fluorescence as a Tool for Salt Stress Detection in Roses. *Photosynthetica.*, **33**: 291–301.
 40. Sayed, O. H. 2003. Chlorophyll Fluorescence as a Tool in Cereal Crop Research. *Photosynthetica*, **41**: 321–330.
 41. Dai, Y. J., Shen, Z. G., Liu, Y., Wang, L. L., Hannaway, D. and Lu, H. F. 2009. Effects of Shade Treatments on the Photosynthetic Capacity, Chlorophyll Fluorescence, and Chlorophyll Content of *Tetrastigma hemsleyanum* Diels et Gilg. *Environ. Exp. Bot.*, **65**: 177–182, 2009.
 42. Bjorkman, O. and Demmig, B. 1987. Proton Yield of O₂ Evolution and Chlorophyll Fluorescence Characteristics at 77K among Vascular Plants of Diverse Origins. *Planta*, **170**: 489–504.
 43. Lundmark, T., Bergh, J., Strand, M. and Kopel, A. 1998. Seasonal Variation of Maximum Photochemical Efficiency in Boreal Norway Spruce Stands. *Tree-Structure Function*, **13**: 63–67.
 44. Veres, S., Toth, V. R., Laposi, R., Olah, V., Lakatos, G. and Meszaros, I. 2006. Carotenoid Composition and Photochemical Activity of Four Sandy Grassland Species. *Photosynthetica*, **44**: 255–261.
 45. Guo, H. X., Liu, W. Q. and Shi, Y. C. 2006. Effects of Different Nitrogen Forms on Photosynthetic Rate and the Chlorophyll Fluorescence Induction Kinetics of Flue-cured Tobacco. *Photosynthetica*, **44**: 140–142.
 46. Tezara, W., Martinez, D., Rengifo, E. and Herrera, A. 2003. Photosynthetic Responses of the Tropical Spiny Shrub *Lycium nodosum* (Solanaceae) to Drought, Soil Salinity and Saline Spray. *Ann. Bot-London*, **92**: 757–765.
 47. Mao, L. Z., Lu, H. F., Wang, Q. and Cai, M. M. 2007. Comparative Photosynthesis Characteristics of *Calycanthus chinensis* and *Chimonanthus praecox*. *Photosynthetica*, **45**: 601–605.
 48. Jin, X. F., Yang, X. E., Islam, E., Liu, D., Mahmood, Q., Li, H. and Li, J. 2006. Ultrastructural Changes, Zinc Hyperaccumulation and Its Relation with Antioxidants in Two Ecotypes of *Sedum alfredii* Hance. *Plant Physiol. Bioch.*, **46**: 997–1006.
 49. Mantri, N., Pang, E. C. K. and Ford, R. 2010. Molecular Biology for Stress Management. In: “Climate Change and Management of Cool Season Grain Legume Crops”, (Eds.): Yadav, S. S., McNeil, D. N., Weeden, N. and Patil, S. S.. Springer, Heidelberg, PP. 377–408.
 50. Rang, Z. W., Jagadish, S. V. K., Zhou, Q. M., Craufurd, P. Q. and Heuer, S. 2011. Effect of High Temperature and Water Stress on Pollen Germination and Spikelet Fertility in Rice. *Environ. Exp. Bot.*, **70**: 58–65.
 51. Matoh, T., Kairusmee, P. and Takahashi, E. 1986. Salt-induced Damage to Rice Plants and Alleviation Effects of Silicate. *Soil Sci. Plant Nutr.*, **32**: 295–304.
 52. Ahmad, R., Zaheer, S. H. and Ismail, S. 1992. Role of Silicon in Salt Tolerance of Wheat (*Triticum aestivum* L.). *Plant Sci.*, **85**: 43–50.
 53. Al-Aghabary, K., Zhu, Z. J. and Shi, Q. H. 2005. Influence of Silicon Supply on Chlorophyll Content, Chlorophyll



- Fluorescence and Antioxidative Enzyme Activities in Tomato Plants under Salt Stress. *J. Plant Nutr.*, **27**: 2101–2115.
54. Zhu, Z. J., Wei, G. Q., Li, J., Qian, Q. Q. and Yu, J. Q. 2004. Silicon Alleviates Salt Stress and Increases Antioxidant Enzymes Activity in Leaves of Salt-stressed Cucumber (*Cucumis sativus* L.). *Plant Sci.*, **167**: 527–533.

غلظت بهینه سولفات روی در برگپاشی برای کاهش تنش شوری در *Glycine soja*

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چکیده

پژوهش‌های قبلی نشان داده است که عنصر روی تنش شوری را کم میکند. در پژوهش حاضر، به منظور تعیین غلظت بهینه سولفات روی در کودی که بر برگ‌ها پاشیده می‌شود تا اثر تنش شوری را کاهش دهد، پارامترهای مختلف مربوط به زیست توده و فلورسنس کلروفیل در گیاه سویای *Glycine soja* اندازه‌گیری شد. نخست، بوته‌های گیاه از طریق محلول غذایی تحت تنش غلظت‌های مختلف کلرید سدیم (NaCl) برابر $0, 100, 200, \text{ and } 300 \text{ mmol L}^{-1}$ قرار داده شدند و سپس غلظت‌های مختلف سولفات روی برابر $0, 5, 10, 15, 20, 25 \text{ } \mu\text{mol L}^{-1}$ بر برگ‌ها پاشیده شد. نتایج نشان داد که زیست توده و پارامترهای فلورسنس کلروفیل گیاهچه‌ها به طور معنی‌داری ($P < 0.05$) تحت تاثیر تنش شوری بودند. با این وجود، پاشیدن سولفات روی بر گیاه در سازگاری با تنش شوری کمک کرد. غلظت‌هایی از سولفات روی که در سازگاری گیاه با شوری‌های $100, 200, \text{ and } 300 \text{ mmol L}^{-1}$ کمک کردند، به ترتیب برابر بود با $15, 20, \text{ and } 25 \text{ } \mu\text{mol L}^{-1}$ و $10, 20, \text{ and } 25 \text{ } \mu\text{mol L}^{-1}$ میکرومول در لیتر. غلظت‌های کمتر سولفات روی در برابر تنش کارایی نداشتند و غلظت‌های بیشتر با ایجاد مسمومیت باعث توقف رشد می‌شدند. در تمام سطوح تنش شوری، غلظت $15 \text{ } \mu\text{mol L}^{-1}$ میکرومول در لیتر سولفات روی مناسب‌ترین تاثیر را داشت. اندازه‌گیری پارامترهای رشد مانند بخش برگ‌های واقعی و وزن خشک کل گیاهچه‌ها با پارامترهای فلورسنس کلروفیل هماهنگ بود که این امر نشانگر بهبود فعالیت فتوسنتزی برگ در غلظت‌های $10-20 \text{ } \mu\text{mol L}^{-1}$ میکرومول در لیتر بود.