Original Article

# Synthesis, Antiplatelet Activity and Cytotoxicity Assessment of Indole-Based Hydrazone Derivatives

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#### Abstract

A series of indole-based aryl(aroyl)hydrazone analogs of antiplatelet indole-3carboxaldehyde phenylhydrazone were synthesized by the Schiff base formation reaction and their antiplatelet activity was assessed using human platelet rich plasma. The platelet concentrate was obtained using a two-step centrifugation protocol and ADP, arachidonic acid and collagen were used as inducers of platelet aggregation. Based on the results, substituted phenylhydrazones showed promising activity. Among them, compound 1i was the most potent derivative with an IC<sub>50</sub> comparable to that of indomethacin as a standard drug. The hydrazone derivatives were also tested for their cytotoxicity using on platelet concentrates and fibroblast L929 cells. The majority of the derivatives showed an acceptable selectivity towards antiplatelet aggregation activity.

Based on the activity data, phenylhydrazone derivatives (1a-i) exhibited considerable antiplatelet activity and minimal toxic effect on platelet cells. The results of the present study could provide a better understanding of the structure activity relationship of antiplatelet indolehydrazones.

Keywords: Indole; Arylhydrazone; Antiplatelet aggregation; Arachidonic acid; Collagen.

#### Introduction

As a major cause of death and disability, thromboembolic disorders are a serious health problem in worldwide and even in industrial countries(1). It has been revealed that increased self-affinity and aggregation of platelets play animportant role in the pathogenesis of atherothrombosis(2). To prevent such a condition, antiplatelet agents including aspirin and clopidogrelare employed in the drug therapy regimens. However, there are serious limitations to these agents including increased risk of bleeding(3) and drug resistance(4). Therefore, discovery and development of new antiplatelet agents with improved efficacy and safety are an urgent need.

In the published works focused on the development of novel antiplatelet agents, a variety of hydrazone derivatives have been reported. The antiplatelet aryl(aroyl)

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Figure 1. The antiplatelet indole-3-carboxaldehyde phenylhydrazone.

hydrazones of aromatic aldehydes and methyl ketones reported by Coquelet *et al.*(5) and diverse groups of hydrazones including arylsulfonateacylhydrazones(6), phenothiazine-1-acylhydrazones(7)*N*-substituted-phenyl-1,2,3-triazole-4-acylhydrazones(8) and pyrazolylhydrazones(9) are some examples of hydrazone derivatives with potent antiplatelet aggregation activities.

Based on the above mentioned reports, we have investigated the antiplatelet activity of some novel indole and hydrazone derivatives(10,11, 12). Focused on the bioscreening of hydrazone derivatives of aromatic aldehydes, we found some arylhydrazones of electron rich heteroaromatic aldehydes as the most potent derivatives among the studied compounds. Of the most potent derivatives, indole-3-carboxaldehyde phenylhydrazone exhibited excellent inhibition of platelet aggregation induced by arachidonic acid (IC<sub>50</sub> = 10  $\mu$ M, Figure 1). This compound has a relatively small molecular weight (235.28) and an optimal choice to be considered as lead compound.

Therefore, in the present study, in order to better understand the structure-activity relationship of indole-based hydrazones and also to discover new derivatives with improved antiplatelet activity, we synthesized some new hydrazone derivatives of indole-3carboxaldehyde and assessed their inhibitory activity against arachidonic acid-, ADPand collagen-induced platelet aggregation. As shown in figure 2, the derivatives have been divided in 3 subgroups including arylhydrazones of indole-3-carboxaldehyde aroylhydrazones of (1a-i), indole-3carboxaldehyde (2a-g) and aroylhydrazones 5-chloroindole-3-carboxaldehyde (3a-g). Different substitutions have been made on the aromatic ring to evaluate the relationship between different physicochemical parameters and the observed bioactivities.

In addition to antiplatelet activity, the synthesized derivatives were evaluated for their potential toxicity on a normal cell line (L929) and on platelet itself (platelet membrane leakage assay).

#### **Experimental**

#### Materials and Methods

Melting points were measured by an Electrothermal 9200 apparatus and were uncorrected. The <sup>1</sup>H-NMR spectra were measured by a 400 MHz BrukerAvance DRX spectrometer in  $d_6$ -DMSO and TMS was used as an internal standard. The infrared spectra were obtained by a Bruker Tensor27ATR-FTIR spectrophotometer and samples were directly



Figure 2. General structure of the synthesized derivatives.

used without the addition of KBr. ESI-MS spectra were obtained by Agilent 6410 Triple Quad LC/MS. The derivatives were analyzed for C, H, N by a Costech model 4010 and the obtained values were in agreement with the proposed structures within  $\pm 0.4\%$  of the theoretical values. The synthesis and spectral characterization of compounds 1a(11), 1b, 1e, 1f(13), 3a(14)and 3c(15) are available in the literature. Calculation of the global physicochemical parameters including ClogP, polarizability (P), refractivity (R), surface area (SA) and molecular volume (V) were performed using Hyperchem 8.0 software.

# General procedure for the synthesis of hydrazone derivatives

(5-Chloro)Indole-3-carboxaldehyde and appropriate arylhydrazines/aroylhydrazides (1 mmol of each) were added to a mixture of ethanol (96%, 10 mL) and glacial acetic acid (8 drops). The resulting mixture was heated under reflux and the reaction progress was monitored by thin layer chromatography. The mixture was then cooled to room temperature. The precipitate was filtered, washed with *n*-hexane and dried in the open air to afford the final derivatives.

Indole-3-carboxaldehyde 2-chlorophenylhydrazone (1c)

Yield 69%,m.p. 150-153°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_{\delta}$ )  $\delta$ : 6.72 (dt, J = 8.0 Hz, J = 1.6 Hz, 1H, Ar H), 7.14-7.21 (m, 2H, indole C<sub>5,6</sub>-H), 7.29 (m, 2H, Ar H), 7.42 (dd, J = 6.8 Hz, J = 2.0 Hz, 1H, indole C<sub>7</sub>-H), 7.54 (d, J = 7.6 Hz, 1H, Ar H), 7.65 (d, J = 2.8 Hz, 1H, indole C<sub>2</sub>-H), 8.23 (d, J = 7.6 Hz, 1H, indole C<sub>4</sub>-H), 8.49 (s, 1H, imine H), 9.37 and 11.41 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3344, 1597, 1438, 1240, 736. ESI-MS *m/z*: 270, 272 (M + H<sup>+</sup>). Anal.Calcd for C<sub>15</sub>H<sub>12</sub>ClN<sub>3</sub>: C, 66.79; H, 4.48; N, 15.58. Found: C, 66.67; H, 4.47; N, 15.60.

*Indole-3-carboxaldehyde 3-chlorophenylhydrazone* (1d)

Yield 77%,m.p. 211-214°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_b$ ) $\delta$ : 6.67 (d, J = 8.00 Hz, 1H, Ar C<sub>6</sub>-H), 6.94 (d, J = 8.4 Hz, 1H, Ar C<sub>4</sub>-H), 7.01 (s, 1H, Ar C<sub>2</sub>-H), 7.14-7.23 (m, 3H, indole C<sub>5,6</sub>-H + Ar C<sub>5</sub>-H), 7.42 (d, J = 7.6 Hz, 1H, indole C<sub>7</sub>-H),

7.66 (d, J = 2.4 Hz, 1H, indole C<sub>2</sub>-H), 8.12 (s, 1H, imine H), 8.18 (d, J = 7.6 Hz, 1H, indole C<sub>4</sub>-H), 10.04 and 11.37 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3415, 1589, 1530, 1426, 1083, 750, 682. ESI-MS *m/z*: 270, 272 (M + H<sup>+</sup>). Anal.Calcd for C<sub>15</sub>H<sub>12</sub>ClN<sub>3</sub>: C, 66.79; H, 4.48; N, 15.58. Found: C, 66.68; H, 4.48; N, 15.57.

# *Indole-3-carboxaldehyde 2-methylphenylhydrazone* (1g)

Yield 83%,m.p. 176-178°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ : 2.20 (s, 3H, CH<sub>3</sub>), 6.64 (t, J = 7.2 Hz, 1H, Ar H), 7.02 (d, J = 7.2 Hz, 1H, Ar H), 7.12-7.19 (m, 3H, indole C<sub>5.6</sub>-H + Ar H), 7.42 (m, 2H, indole C<sub>2</sub>-H + Ar H), 7.62 (d, J = 2.4 Hz, 1H, indole C<sub>2</sub>-H), 8.24 (d, J = 6.8 Hz, 1H, indole C<sub>4</sub>-H), 8.37 (s, 1H, imine H), 8.99 and 11.33 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3242, 1601, 1531, 1242, 1118, 1074. ESI-MS *m/z*: 250 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>15</sub>N<sub>3</sub>. C, 77.08; H, 6.06; N, 16.85. Found: C, 77.17; H, 6.05; N, 16.87.

*Indole-3-carboxaldehyde* 4-*methylphenylhydrazone* (1*h*)

Yield 89%,m.p. 228-231°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ : 2.20 (s, 3H, CH<sub>3</sub>), 6.93 (d, J = 8.2 Hz, 2H, Ar C<sub>3,5</sub>-H), 7.02 (d, J = 8.2 Hz, 2H, Ar C<sub>2,6</sub>-H), 7.11-7.18 (m, 2H, indole C<sub>5,6</sub>-H), 7.39 (d, J = 7.2 Hz, 1H, indole C<sub>7</sub>-H), 7.59 (d, J = 2.4 Hz, 1H, indole C<sub>2</sub>-H), 8.07 (s, 1H, imine H), 8.22 (d, J = 7.2 Hz, 1H, indole C<sub>4</sub>-H), 9.65 and 11.27 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3296, 1608, 1242, 1079, 809, 751. ESI-MS *m/z*: 250 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>15</sub>N<sub>3</sub>: C, 77.08; H, 6.06; N, 16.85. Found: C, 77.15; H, 6.06; N, 16.82.

Indole-3-carboxaldehyde 4-methoxyphenylhydrazone (1i)

Yield 75%,m.p. 200-203°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ :3.68 (s, 3H, OCH<sub>3</sub>), 6.86 (d, J = 8.4 Hz, 2H, Ar C<sub>3.5</sub>-H), 6.96 (d, J = 8.4 Hz, 2H, Ar C<sub>2.6</sub>-H), 7.10-7.18 (m, 2H, indole C<sub>5.6</sub>-H), 7.39 (d, J = 6.8 Hz, 1H, indole C<sub>7</sub>-H), 7.57 (d, J = 2.8 Hz, 1H, indole C<sub>2</sub>-H), 8.06 (s, 1H, imine H), 8.23 (d, J = 7.2 Hz, 1H, indole C<sub>4</sub>-H), 9.53 and 11.26 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3285, 1610, 1511, 1441, 1240, 822. ESI-MS *m/z*: 266 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>15</sub>N<sub>3</sub>O: C, 72.43; H, 5.70; N, 15.84. Found: C, 72.31; H, 5.68; N, 15.89.

*Indole-3-carboxaldehyde benzoylhydrazone* (2*a*)

Yield 90%,m.p. 238-241°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ : 7.13-7.22 (m, 2H, indole C<sub>5,6</sub>-H), 7.43 (d, J = 7.6 Hz, 1H, indole C<sub>7</sub>-H), 7.49-7.58 (m, 3H, Ar C<sub>3,4,5</sub>-H), 7.81 (s, 1H, indole C<sub>2</sub>-H), 7.91 (d, J = 7.2 Hz, 2H, Ar C<sub>2,6</sub>-H), 8.29 (d, J = 7.6 Hz, 1H, indole C<sub>4</sub>-H), 8.61 (s, 1H, imine H), 11.49 and 11.56 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3209, 1603, 1575, 1443, 720. ESI-MS *m/z*: 264 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>13</sub>N<sub>3</sub>O: C, 72.99; H, 4.98; N, 15.96. Found: C, 73.06; H, 4.98; N, 15.92.

Indole-3-carboxaldehyde 3-chlorobenzoylhydrazone (2b)

Yield 78%,m.p. 215-218°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ :7.13-7.22 (m, 2H, indole C<sub>5,6</sub>-H), 7.43 (d, J = 8.0 Hz, 1H, indole C<sub>7</sub>-H), 7.56 (t, J = 8.0 Hz, 1H, Ar C<sub>5</sub>-H), 7.64 (d, J = 8.4 Hz, 1H, Ar H), 7.83 (s, 1H, indole C<sub>2</sub>-H), 7.88 (d, J = 8.4 Hz, 1H, Ar H), 7.96 (s, 1H, Ar C<sub>2</sub>-H), 8.28 (d, J = 7.6 Hz, 1H, indole C<sub>4</sub>-H), 8.60 (s, 1H, imine H), 11.52 and 11.59 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3259, 1603, 1357, 879, 680. ESI-MS *m/z*: 298, 300 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>12</sub>ClN<sub>3</sub>O: C, 64.54; H, 4.06; N, 14.11. Found: C, 64.45; H, 4.07; N, 14.13.

In  $d \circ l e - 3 - c a r b \circ x a l d e h y d e$ 4-chlorobenzoylhydrazone (2c)

Yield 86%, m.p. 230-233°C<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ :7.12-7.21 (m, 2H, indole C<sub>5.6</sub>-H), 7.43 (d, J = 8.0 Hz, 1H, indole C<sub>7</sub>-H), 7.59 (d, J = 8.4 Hz, 2H, Ar H), 7.82 (s, 1H, indole C<sub>2</sub>-H), 7.94 (d, J = 8.4 Hz, 2H, Ar H), 8.28 (d, J = 8.0 Hz, 1H, indole C<sub>4</sub>-H), 8.60 (s, 1H, imine H), 11.56 (s, 2H, NH); IR (cm<sup>-1</sup>): 3341, 1617, 1595, 1354, 1097, 843. ESI-MS *m/z*: 298, 300 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>12</sub>ClN<sub>3</sub>O: C, 64.54; H, 4.06; N, 14.11. Found: C, 64.40; H, 4.07; N, 14.16.

*Indole-3-carboxaldehyde* 4-nitrobenzoylhydrazone (2d)

Yield 71%,m.p. 250-253°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_b\delta$ :7.14-7.23 (m, 2H, indole C<sub>5,6</sub>-H), 7.44 (d, J = 7.6 Hz, 1H, indole C<sub>7</sub>-H), 7.86 (s, 1H, indole C<sub>2</sub>-H), 8.15 (d, J = 8.8 Hz, 2H, Ar C<sub>3,5</sub>-H), 8.29 (d, J = 7.6 Hz, 1H, indole

C<sub>4</sub>-H), 8.36 (d, J = 8.8 Hz, 2H, Ar C<sub>2.6</sub>-H), 8.63 (s, 1H, imine H), 11.62 and 11.79 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3336, 1639, 15595, 1350, 744. ESI-MS *m*/*z*: 309 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>12</sub>N<sub>4</sub>O<sub>3</sub>: C, 62.33; H, 3.92; N, 18.17. Found: C, 63.41; H, 3.92; N, 18.11.

*Indole-3-carboxaldehyde 3-methylbenzoylhydrazone* (2e)

Yield 74%,m.p. 212-214°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ : 2.39 (s, 3H, CH<sub>3</sub>), 7.14-7.20 (m, 2H, indole C<sub>5,6</sub>-H), 7.38-7.44 (m, 3H, Ar C<sub>4,5</sub>-H + indole C<sub>7</sub>-H), 7.70 (d, J = 6.8 Hz, 1H, Ar C<sub>6</sub>-H), 7.72 (s, 1H, Ar C<sub>2</sub>-H), 7.80 (s, 1H, indole C<sub>2</sub>-H), 8.29 (d, J = 7.6 Hz, 1H, indole C<sub>4</sub>-H), 8.60 (s, 1H, imine H), 11.44 and 11.56 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3205, 1600, 1575, 708. ESI-MS *m/z*: 278 (M + H<sup>+</sup>). Anal.Calcd for C<sub>17</sub>H<sub>15</sub>N<sub>3</sub>O: C, 73.63; H, 5.45; N, 15.15. Found: C, 73.52; H, 5.44; N, 15.14.

Indole-3-carboxaldehyde 4-methylbenzoylhydrazone (2f)

Yield 88%,m.p. > 270°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ :2.37 (s, 3H, CH<sub>3</sub>), 7.12-7.21 (m, 2H, indole C<sub>5,6</sub>-H), 7.31 (d, J = 8.0 Hz, 2H, Ar C<sub>2,6</sub>-H), 7.43 (d, J = 8.0 Hz, 1H, indole C<sub>7</sub>-H), 7.79 (s, 1H, indole C<sub>2</sub>-H), 7.83 (d, J = 8.0 Hz, 2H, Ar C<sub>3,5</sub>-H), 8.29 (d, J = 7.6 Hz, 1H, indole C<sub>4</sub>-H), 8.61 (s, 1H, imine H), 11.42 and 11.55 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3331, 1618, 1597, 1302, 837, 752. ESI-MS *m*/*z*: 278 (M + H<sup>+</sup>). Anal.Calcd for C<sub>17</sub>H<sub>15</sub>N<sub>3</sub>O: C, 73.63; H, 5.45; N, 15.15. Found: C, 73.70; H, 5.47; N, 15.12.

# 5-Chloroindole-3-carboxaldehyde 3-chlorobenzoylhydrazone (3b)

Yield 71%,m.p. 230-233°C<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ :7.22 (d, J = 8.4 Hz, J = 2.4 Hz, 1H, indole C<sub>6</sub>-H), 7.46 (d, J = 8.4 Hz, 1H, indole C<sub>7</sub>-H), 7.56 (t, J = 8.0 Hz, 1H, Ar C<sub>5</sub>-H), 7.64 (d, J = 8.0 Hz, 1H, Ar C<sub>4</sub>-H), 7.88 (d, J = 8.0 Hz, 1H, Ar C<sub>6</sub>-H), 7.92 (s, 1H, indole C<sub>2</sub>-H), 7.96 (s, 1H, Ar C<sub>2</sub>-H), 8.30 (d, J = 2.4 Hz, 1H, indole C<sub>4</sub>-H), 8.58 (s, 1H, imine H), 11.66 and 11.77 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3322, 1625, 1602, 1357, 785. ESI-MS *m/z*: 332, 334, 336 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>11</sub>Cl<sub>2</sub>N<sub>3</sub>O: C, 57.85; H, 3.34; N, 12.65. Found: C, 57.70; H, 3.35; N, 12.62.

### 5 - Chloroindole - 3 - carboxaldehyde 4-nitrobenzoylhydrazone (3d)

Yield 66%, m.p. > 270°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ : 7.22 (dd, J = 8.4 Hz, J = 2.0 Hz, 1H, indole C<sub>6</sub>-H), 7.47 (d, J = 8.4 Hz, 1H, indole C<sub>7</sub>-H), 7.95 (s, 1H, indole C<sub>2</sub>-H), 8.15 (d, J = 8.4 Hz, 2H, Ar C<sub>3.5</sub>-H), 8.31 (d, J = 2.0 Hz, 1H, indole C<sub>4</sub>-H), 8.37 (d, J = 8.4 Hz, 2H, Ar C<sub>2.6</sub>-H), 8.61 (s, 1H, imine H), 11.83 (s, 2H, NH); IR (cm<sup>-1</sup>): 3396, 1662, 1518, 1345, 1107. ESI-MS *m/z*: 343, 345 (M + H<sup>+</sup>). Anal.Calcd for C<sub>16</sub>H<sub>11</sub>ClN<sub>4</sub>O<sub>3</sub>: C, 56.07; H, 3.23; N, 16.35. Found: C, 56.13; H, 3.23; N, 16.34.

# 5 - Chloroindole - 3 - carboxaldehyde 3-methylbenzoylhydrazone (3e)

Yield 69%,m.p. 226-228°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ : 2.39 (s, 3H, CH<sub>3</sub>), 7.21 (dd, J = 8.8 Hz, J = 2.0 Hz, 1H, indole C<sub>6</sub>-H), 7.39 (m, 2H, Ar H), 7.46 (d, J = 8.8 Hz, 1H, indole C<sub>7</sub>-H), 7.71 (m, 2H, Ar H), 7.89 (s, 1H, indole C<sub>2</sub>-H), 8.31 (d, J = 2.0 Hz, 1H, indole C<sub>4</sub>-H), 8.59 (s, 1H, imine H), 11.52 and 11.75 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3321, 1622, 1601, 1657, 798. ESI-MS *m*/*z*: 312, 314 (M + H<sup>+</sup>). Anal.Calcd for C<sub>17</sub>H<sub>14</sub>ClN<sub>3</sub>O: C, 65.49; H, 4.53; N, 13.48. Found: C, 65.57; H, 4.54; N, 13.46.

# 5 - Chloroindole - 3 - carboxaldehyde 4-methylbenzoylhydrazone (3f)

Yield 59%,m.p. 244°C (dec.).<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ :2.38 (s, 3H, CH<sub>3</sub>), 7.21 (d, J = 8.8 Hz, J = 1.6 Hz, 1H, indole C<sub>6</sub>-H), 7.32 (d, J = 8.0 Hz, 2H, Ar C<sub>3.5</sub>-H), 7.46 (d, J = 8.8 Hz, 1H, indole C<sub>7</sub>-H), 7.82 (d, J = 8.0 Hz, 2H, Ar C<sub>2.6</sub>-H), 7.89 (s, 1H, indole C<sub>2</sub>-H), 8.31 (d, J = 1.6 Hz, 1H, indole C<sub>4</sub>-H), 8.59 (s, 1H, imine H), 11.49 and 11.74 (each s, 1H, NH); IR (cm<sup>-1</sup>): 3177, 1630, 1610, 1384, 1103, 891. ESI-MS *m/z*: 312, 314 (M + H<sup>+</sup>). Anal.Calcd for C<sub>17</sub>H<sub>14</sub>ClN<sub>3</sub>O: C, 65.49; H, 4.53; N, 13.48. Found: C, 65.55; H, 4.54; N, 13.43.

### 5-Chloroindole-3-carboxaldehyde isonicotinoylhydrazone (3g)

Yield 73%,m.p. > 270°C.<sup>1</sup>HNMR (400 MHz, DMSO- $d_6$ ) $\delta$ :7.22 (dd, J = 8.8 Hz, J = 2.0 Hz, 1H, indole C<sub>6</sub>-H), 7.47 (d, J = 8.8 Hz, 1H, indole C<sub>7</sub>-H), 7.82 (d, J = 6.0 Hz, 2H, pyridine C<sub>2</sub>-H), 7.94 (s, 1H, indole C<sub>2</sub>-H), 8.30 (d, J =

2.0 Hz, 1H, indole C<sub>4</sub>-H), 8.60 (s, 1H, imine H), 8.77 (d, J = 6.0 Hz, 2H, pyridine C<sub>3.5</sub>-H), 11.79 (s, 2H, NH); IR (cm<sup>-1</sup>): 1661, 1412, 1127, 889, 782, 749. ESI-MS *m*/*z*: 299, 301 (M + H<sup>+</sup>). Anal.Calcd for C<sub>15</sub>H<sub>11</sub>ClN<sub>4</sub>O: C, 60.31; H, 3.71; N, 18.76. Found: C, 60.20; H, 3.70; N, 18.81.

# *In-vitro evaluation of antiplatelet aggregation activity*

The *In-vitro* antiplatelet activity of the derivatives was evaluated using human platelet rich plasma (PRP)on an APACT 4004 aggregometerby Born's reported turbidimetric method(16). ADP (5  $\mu$ M), arachidonic acid (1.25 mg/mL) and collagen (5  $\mu$ M) were used as inducers of platelet aggregation.  $IC_{50}$  was defined as the concentration of the test compound that inhibits the platelet aggregation by 50%. The detailed procedure has been described in our previous works(11,12,17).

### Cytotoxicity assay

The synthesized derivatives were assayed for their toxicity on Fibroblast L929 cell line by MTT [3-(4,5-dimethylthiazol-2-yl-2,5tetrazolium bromide)] method. The compounds were initially screened at 100  $\mu$ M concentration and IC<sub>50</sub> was calculated for those showing more than 50% toxicity. The detailed procedure has been described in our previous work(18).

#### Platelet lactate dehydrogenase (LDH) assay

The assay was performed according to the previously reported procedures(19). Briefly, PRP samples (200 µL) were incubated with test solutions (1 µL, final concentration of 100  $\mu$ M) for 30 min. The samples were then centrifugatedat 11000 rpm for 5 min and the supernatant was separated. The activity of lactate dehydrogenase was evaluated using the supernatant with an LDH assay kit (Pishtazteb, Iran) following the instructions provided by the manufacturer. DMSO and 1% Triton X-100 were used as vehicle and positive control respectively. The data was expressed as the proportion of released LDH following exposure to test compounds to the total LDH content measured by the addition of 1% Triton X-100.



Scheme 1. Synthetic route to the hydrazone derivatives; reagents and conditions: (*i*) arylhydrazines, ethanol, glacial acetic acid, reflux; (*ii*) aroylhydrazides, ethanol, glacial acetic acid, reflux.

#### **Results and Discussion**

### Chemistry

As disclosed in scheme 1, the final hydrazones were prepared by a single step procedure. To this end, (5-chloro)indole-3-carboxaldehyde was reacted to different arylhydrazines (compounds 1a-i) or aroylhydrazides (compounds 2a-g and 3a-g). The reactions were performed in ethanol using catalytic amounts of glacial acetic acid. Notably, the aroylhydrazides were purchased from the commercial suppliers or otherwise synthesized from their starting benzoic acids following the previously reported methods(20). Following the typical procedure (as described in the experimental section) the final hydrazone derivatives were prepared in good yields with excellent analytical purity as confirmed by thin layer chromatography and CHN elemental analysis.

The spectral data were in complete agreement with the structure of the derivatives. In the <sup>1</sup>H-NMR spectra, the hydrogens of the indole and 5-chloroindole nucleus resonated by a characteristic pattern. In compounds 1a-i and 2a-gindolehydrogensappeared as a set of multiplet in 7.10-7.25 ppm (assigned to indole H-5 and H-6), two sets of doublets at ~ 7.46 ppm and ~ 8.30 ppm (assigned to indole H-7 and H-4) and a singlet at 7.8-8.0 ppm (assigned to indole H-2). In compounds 3a-g, two singlets at7.8-8.0 ppm and ~ 8.30 ppm and ~ 8.40 ppm and ~

picks for 5-chloroindole nucleus. The aromatic hydrogens on the aryl (aroyl)hydrazone moiety were readily distinguishable in the <sup>1</sup>H-NMR spectra of the derivatives according to their multiplicity patterns. Analysis of the molecular mass of the derivatives was performed by ESI-MS and molecular ions of the compounds were observed as adducts of hydrogen and/or sodium.

#### Biological activity

The antiplatelet activity of the derivatives has been disclosed in table 1. All the compounds were evaluated for their In-vitro antiplatelet activity using arachidonic acid (AA), ADP and collagen as inducers of platelet aggregation. At the first view, it could be noted that the arylhydrazone derivatives 1a-i showed considerable activity against AA- and collagen-induced platelet aggregation and showed no activity when ADP was used as aggregation inducer. However, aroylhydrazones 2a-g and 3a-gwere completely inactive. It should be considered that Hydrazones contain two connected nitrogen atoms of different nature and a C-N double bond that is conjugated with a lone electron pair of the terminal nitrogen atom. These structural fragments are mainly responsible for the physical and chemical properties of hydrazones. Both nitrogen atoms of the hydrazine group are nucleophilic, although the amino type nitrogen is most reactive. Despite the high structural similarity between hydrazones and carbohydrazones, the electronic characteristics of these compounds Table 1. Antiplatelet activity and cytotoxicity of the derivatives.



Derivative	Х	Ar	Antiplatelet a	Cytotoxicity, IC <sub>50</sub>		
			AA	ADP	Collagen	(μM) <sup>b</sup>
1a	-	Phenyl	10 <sup>a</sup>	>100ª	38.75	>100
1b	-	4-fluorophenyl	28.8	>100	>100	>100
1c	-	2-chlorophenyl	14.6	>100	85.56	45.88
1d	-	3-chlorophenyl	28.12	>100	>100	33.89
1e	-	4-chlorophenyl	22.54	>100	62.54	66.66
1f	-	4-bromophenyl	21.2	100	>100	59.83
1g	-	2-methylphenyl	56.8	>100	75.21	85.07
1h	-	4-methylphenyl	57.19	>100	>100	>100
1i	-	4-methoxyphenyl	7.13	>100	>100	74.54
2a	Н	Phenyl	>100	>100	>100	>100
2b	Н	3-chlorophenyl	>100	>100	>100	>100
2c	Н	4-chlorophenyl	>100	>100	>100	>100
2d	Н	4-nitrophenyl	>100	>100	>100	>100
2e	Н	3-methylphenyl	>100	>100	>100	>100
2f	Н	4-methylphenyl	>100	>100	>100	>100
2g	Н	4-pyridyl	>100ª	>100ª	>100	>100
3a	Cl	Phenyl	>100	>100	>100	>100
3b	Cl	3-chlorophenyl	>100	>100	>100	55.45
3c	Cl	4-chlorophenyl	>100	>100	>100	71.32
3d	Cl	4-nitrophenyl	>100	>100	>100	>100
3e	Cl	3-methylphenyl	>100	>100	>100	49.59
3f	Cl	4-methylphenyl	>100	>100	>100	62.73
3g	Cl	4-pyridyl	>100	>100	>100	>100
Indometacin			3.0	>100	1.5	
Aspirin			30.3	>100	9.7	

<sup>a</sup> Obtained in our previous study (11)

<sup>b</sup> Assayed against fibroblast L929 cell line after 24 h of exposure.

are quite different: the presence of carbonyl, conjugated with the lone pair of electrons will change the delocalization pattern of electrons on carbohydrazone derivatives.

In carbohydrazones, the amide-type resonance between the nitrogen and carbonyl alters the electron density so that the complexationbehavior of these compounds differ substantially from that of hydrazones. Therefore the striking difference between the biological activities of hydrazine derivatives and their carbohydrazone congeners could be on account of the different electronic interaction of these two group of compounds with their receptors. It is also worth mentioning that the companionship of carbonyl moiety in carbohydrazone derivatives will also change the shape and size of the molecules.

In the arylhydrazone series, halogen and methoxy substituted derivatives exhibited acceptable activity against AA-induced platelet

Derivative		R	Р	V	SA	
	CLogP				Approx	Grid
1a	4.13	74.01	29.04	728.65	350.18	450.79
1b	4.27	74.22	28.94	732.23	356.62	452.41
1c	4.65	78.81	30.96	768.53	376.46	474.03
1d	4.65	78.81	30.96	781.96	392.35	486.85
1e	4.65	78.81	30.96	772.99	385.48	473.83
1f	4.92	81.63	31.66	792.01	394.56	485.00
1g	4.60	79.05	30.87	782.24	382.59	485.04
1h	4.60	79.05	30.87	781.82	393.27	478.63
1i	3.88	80.47	31.51	806.46	408.80	495.65

Table 2. Global physicochemical parameters calculated for compounds 1a-i.

aggregation. Compound 1i was the most potent derivative in the current study with IC<sub>50</sub> value of 7.13  $\mu$ M which was lower that its parent compound (1a). The activity of compound 1c (R = 2-chlorophenyl) and 1f (4-bromophenyl) was still acceptable.Bioevaluation of the synthesized derivatives against collagen-induced platelet aggregation revealed that some arylhydrazones including 1a, 1e and 1g showed moderate activity. Among them, compound 1a was the most active one with IC<sub>50</sub> value of 38.75  $\mu$ M.

The global physicochemical parameters for the active compounds 1a-i including ClogP, refractivity, polarizability, molecular volume and surface area were calculated and the results were presented in table 2. Analysis of the activity data and calculated physicochemical parameters shows a weak correlation between  $IC_{50}$ values for AA-induced platelet aggregationand lipophilicity(CLogP). According to table 2, the most active compounds including 1a and 1i have the lowest calculated LogP values in the arylhydrazone series; and as a general trend, the increase of lipophilicity is accompanied by reduction of bioactivity. As mentioned above, this implies that lipophilicity may play a role to some extents in the antiplatelet activity of compounds 1a-i.

Cytotoxicity evaluation of a drug candidate is indeed crucial to its fate in lead identification and the following phases of drug discovery process. Therefore, we examined the cytotoxicity of the hydrazone derivatives on fibroblast L929 cells as a normal cell line and also on platelet concentrates. As shown in table 1, compounds

1a, 1b and 1h showed antiplatelet activity at concentrations which were not toxic to fibroblast cells. In addition, compound 1i as the most potent derivative, showed an acceptable selectivity. Besides the fibroblast cells, the safety of the active arylhydrazone derivatives 1a-i was evaluated on platelet cells. It is known that platelet membrane damage leads to the leakage of lactate dehydrogenase (LDH) from the platelet cytosol to the surrounding medium and the amount of released LDH is proportional to the damaged platelets (19). Therefore, platelet LDH assay is widely used to evaluate the effect of drugs and xenobiotics on platelets(19,22-24). Our experiments showed that incubation of platelet with test compounds did not lead to a significant LDH release as compared to vehicle. This shows the safety of antiplatelet arylhydrazonesto the platelet cells.

#### Conclusion

The present work describes the synthesis and antiplatelet aggregation activity of a series of aryl(aroyl)hydrazone derivatives which were designed by considering indole-3carboxaldehyde laas lead compound. The activity data showed the complete loss of bioactivity when arylhydrazone fragment was replaced by aroylhydrazone. Similar to compound 1a, the other arylhydrazones showed inhibitory activity against AA- and collagen-induced platelet aggregation. Among them, compound 1i proved to be the most potent antiplatelet derivative with an IC<sub>s0</sub> value comparable to that of indomethacin as a standard inhibitor of AA metabolism. In addition, toxicity assessment of the hydrazone derivatives revealed that compounds 1a-i possess an acceptable selectivity towards the antiplatelet activity.

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