

**pH**

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## **Study of pH gradient on liposomes trapping efficiency by spectrophotometry: using neutral red**

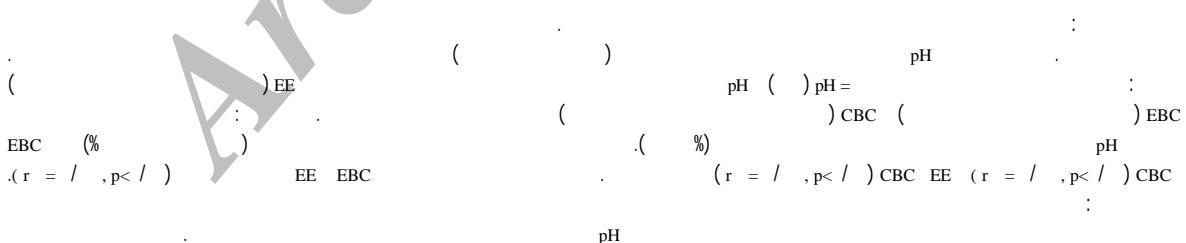
Mohammadi-Bardbori A.\*

Faculty of Pharmacy, Shiraz University of Medical Sciences

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**Objectives:** The purpose of this study was to establish a new experimental, rapid, simple and cost effective approach to study liposomes. NR (Neutral red) was selected as a model of lipophilic and hydrophilic drug. NR is an indicator of pH and a weak base with  $pK_a=6.7$  which is red while ionized in acidic media but yellow while non-ionized in alkaline media and is used in vital staining. **Methods:** In this study the effects of pH internal (5) and external media (5- 9) on liposomes trapping efficiencies in wavelength of 533nm were examined. **Results:** %EE is (the entrapment efficiency of liposome) increased between (%24-%87) in pH (external media) range 5-9. CBC (CBC, contrast between liposomal compartment for visualized by LM) and EBC (EBC, exchanged between liposomal compartment) and EE were calculated. Liposomes with the internal (pH = 5) and external media (pH = 9) is the most suitable option for staining with NR. No significant correlations were observed among the two index with a 95% coefficient interval (EBC with CBC,  $r^2 = 0.93$ ,  $p < 0.05$ , and EE with CBC  $r^2 = 0.93$ ,  $p < 0.05$ ). Significant correlations were also observed among the EBC index and EE index with a 95% coefficient interval (EBC with EE  $r^2 = 0.66$ ,  $p < 0.05$ ). **Conclusion:** NR accumulates in liposomes in ionized form therefore, giving sharp contrast with the surrounding media of liposomes. This makes them to be visualized by LM (light microscope).

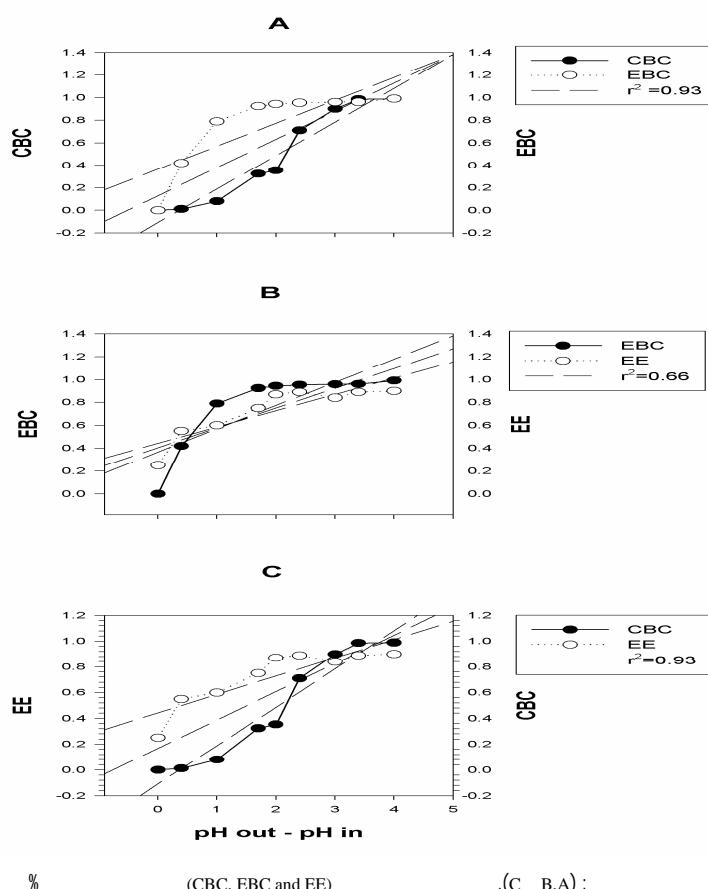
**Keywords:** Liposome; Molecular NR; Ionic NR; Trapping efficiency; spectrophotometry.



\*Corresponding Author: Dr. Afshin Mohammadi-Bardbori,  
Associate Professor, Faculty of Pharmacy, Shiraz University of  
Medical Sciences, Tel: 0711-2424127; Fax: 0711-2424128; E-mail:  
toxicology@sums.ac.ir

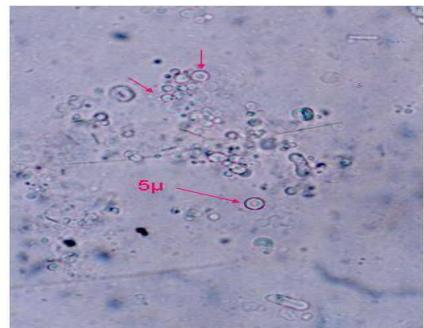
. ( )  
 mmol/L ( pH= / pH= / pH )  
 mmol °C mmol/L NaCl pH  
 pH ( ) pH ( ) pH ( ) pH  
 /M ( ) ( ) ( ) pH ( )  
 (pH = ) ( / mg/ml ) ( )  
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 pH ( ) ( ) ( ) pH ( )  
 Leitz ( )  
 × / × pH ( )  
 :  
 (EE) ( )  
 / ml pH ( )  
 / rpm pH ( )  
 :  
 Qt EE=Qt-Qs/Qt (Alabastar, AL) Avanti Polar Lipid (PC)  
 EE (Milwaukee) (CHOL) %  
 Qs HPLC  
 :  
 (SEM) ±  
 ANOVA Student - Newman Kules ( )  
 SPSS /  
 :

EBC = % up (%UP, percentage of unprotonated NR) out - % up in / % up total.  
EBC / pH  
pH /  
EE EBC CBC /  
% (CBC)  
CBC = % p (%p, percentage of protonated NR) in-%Pout /%P total  
CBC EBC (r = 1, p < 1) CBC EE (r = 1, p < 1)  
EE EBC (r = 1, p < 1)  
.C, B, A / pH  
pH



% (CBC, EBC and EE) .(C B,A) :

pH in	pH out	%p in	%p out	%up in	%up out	CBC	EBC	EE
5	5	98.04	98.04	1.95	1.95	0.000	0.000	0.25
5	5.4	98.04	95.23	1.95	4.77	.0114	0.419	0.55
5	6	98.04	83.36	1.95	16.6	0.080	0.789	0.6
5	6.7	98.04	50	1.95	50	0.324	0.924	0.75
5	7	98.04	33.44	1.95	66.5	0.354	0.943	0.87
5	7.4	98.04	16.6	1.95	83.36	0.710	0.954	0.75
5	8	98.04	4.78	1.95	95.2	0.900	0.959	0.84
5	8.4	98.04	1.56	1.95	98.43	0.986	0.961	0.89
5	9	98.04	.49	1.95	99.5	0.990	0.990	0.90



A

Leitz

.( ) × / ×

pH pH

pH

pH

Pka = /

% /

% /

% /

( )

pH

pH

( μ ) μ / μ

.( ) pH

μl

.( )

**6- References:**

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