

èèì è

:

pH

/ - ÿ/ÿ

/

/

$$LSI = pH - pH_s$$

$$pH_s = A + B - \log[Ca^{2+}] - \log[Alkalinity]$$

(°C) : A
: B

ISO 8044

()

pH : pH_s

CaCO₃ $\frac{mg}{lit}$: ALK

CaCO₃ $\frac{mg}{lit}$ = Ca²⁺

:
LSI < 0 -

(MgCO₃) (CaSO₄)
[] (MgCl₂)

LSI > 0 -

(DO) pH
[]

LSI < 0 -

[]

LSI = 0 -

(LSI)

$$RSI = 2(pH_s) - pH$$

(RSI)

: [] pH
pH: pH PH
pH : pH_s pHs pH

1-Langelier Saturation Index
2-Rizener Stability Index



$RSI \ll i$ -

() $RSI \gg$ -

$RSI \gg$ -

\hat{e}

\dot{y}/\dot{y}

\dot{y}

\hat{e}

l

m^3 / hr

\hat{e}

l

l

$\dot{y} \dot{y} m^3$

\hat{e}

\dot{y}

\dot{y}

$l \dot{y} l$

)

(

pH

RSI LSI

-

<i>RSI</i>		<i>LSI</i>			
/		-	-		
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/	ÿ	ÿ/			
/		/			
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% ÿÿ		% ÿÿ			

-

RSI LSI

<i>RSI</i>		<i>LSI</i>			
-	-	-	-		
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% ÿÿ	ÿ	% ÿÿ	ÿ		

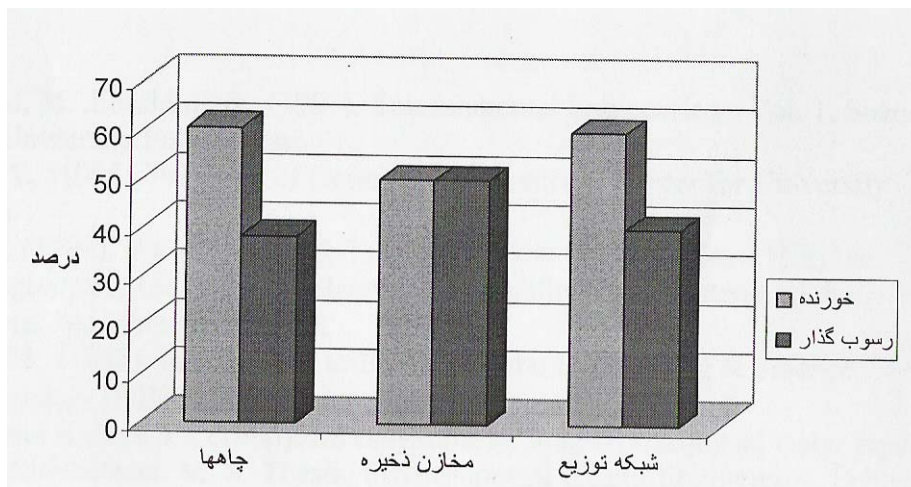
: RSI LSI

-

<i>RSI</i>		<i>LSI</i>			
/		-	-		
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% ÿÿ		% ÿÿ			

RSI LSI

RSI		LSI			
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-	-	-	-		
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% ȳȳ		% ȳȳ			



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