





<u>19<sup>th</sup> Iranian Seminar on Organic Chemistry</u> Vali-e-Asr University of Rafsanjan, 5 -7 Sep. 2012

## Nitration of Phenylacetylene; Synthesis of $\beta$ -nitrostyrene

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In this study, the nitration of phenylacetylene (H-C=C-Ph) was investigated. The investigation of the product has shown that the triple bond of the acetylenic part of phenylacetylene has participated in the nitration process.[1,2] It means that in the reaction condition the additional reaction on the -C=C- of phenylacetylene is much better than the substitution reaction on Ph- group.[1-3] The structures of the products were analyzed by the data of IR, 1H-NMR, 13C-NMR and MS spectroscopy. The B3LYP/6-31G\* shows that the trans- $\beta$ -nitrostyrene is more stable than its cisform. The HOMO level of the trans- and cis- isomers of  $\beta$ -nitrostyrene are -6.95 and -6.90eV, respectively. The experimental and theoretical aspects of this study will be presented.



## **References:**

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- [2] J. C. Hessler, Org. Synth.Coll, 1941, 1, 438.
- [3] G. Hilt, T. Vogler, W. Hess, F. Galbiati (2005). Chemical Communications, 2005, 11, 1474–1475.